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INSTRUMENTATION FOR MULTI-ELECTRODE VOLTAMMETRY

by

I-PIN HO

A dissertation submitted in partial fulfillment of the requirements for the degree of

DOCTOR OF PHILOSOPHY in ENVIRONMENTAL SCIENCES AND RESOURCES: CHEMISTRY

Portland State University C 1990

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TO THE OFFICE OF GRADUATE STUDIES:

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C. William Savery, Interim Vice Provost for Graduate Studies and Research developed to detect metal pollutants at very low concentration levels. However, one important factor is often ignored, i.e., rarely is species determination required. Electrochemical (EC) methods have the particular advantage of being, in principle, a species-sensitive method rather than an element-sensitive method for the study of metal speciation in natural waters.

The goal of this research was to develop an instrument using the EC technique for speciation and general voltammetric studies. It was accomplished by designing a flow-through EC cell containing multi-electrodes to which various fixed potentials over a selected range were applied. A special potentiostat was designed to supply a selected potential to each electrode in the cell. Potential control was provided by placing a combined counter-reference electrode at circuit ground and connecting each working electrode to the inverting input of a current follower which had a potential applied to its non-inverting input from two digital-to-analog converters and a resistor network. Integrating current followers were used for measuring signal currents generated by the electrolytes samples on each electrode. A multiplexing circuit, including an analog-todigital converter, was used to fulfill data acquisition. These circuits were interfaced to a computer and the readout was a pseudo-voltammogram which is a plot of amperometric currents versus various applied fixed potentials on each

electrode. Details of the instrumentation, software, and some initial results are described.

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CHAPTER I

INTRODUCTION

Life begins in water. Life forms in water exposed to numerous metals. Since the late 1960s and early 1970s metal ions and their complexes have gained great attention in environmental studies because of the wide spread occurrence and toxicological importance of metal pollutants in the aquatic environment (1).

Many metals in the aquatic environment have more than one oxidation state and species. The toxicity of this metal toward the aquatic environment is far more dependent on the activity of certain metal species than the activity of the total concentration of that metal (2). The determination of the total concentration in a sample does not clearly describe the potential risks associated with its toxicity and the resulting hazard from exposure. For example, chromium can exist in the +2, +3, +4, +5, and +6 oxidation states in aqueous solution. Among these oxidation states, the most common states are Cr(III) and Cr(VI). Whereas Cr(III) is non-toxic and may exist as anionic or cationic species, Cr(VI) complexes are highly toxic (3) and exist as anionic species (4).

Numerous sophisticated techniques and instruments have

been developed to detect metal pollutants at low concentration levels. However, one important factor is often ignored. That is, rarely is species determination done. Some of the most sensitive analytical techniques, such as atomic absorption spectrophotometry, atomic emission spectrophotometry, and X-ray fluorescence, are not suitable for species studies because they determine only the total concentration of a metal. These techniques may be accompanied with some preliminary treatment and separation methods, e.g. complex formation (5), or high-performance liquid chromatography (HPLC) (6) for the determination of metal species. However, they have serious drawbacks, such as introduction of contaminants, and incomplete separation, by these preliminary procedures.

During the past decade, there has been a rapid increase in the use of the various forms of electrochemical (EC) methods for speciation. Osteryoung (7) reviewed voltammetric techniques for determining trace metals. Florence (2) reviewed the EC approaches to metal speciation in waters. There is a growing interest in EC detection with the simultaneous use of two or more working electrodes. For example, Lunte et al. (8) used dual-electrode detectors to improve selectivity, detection limits and qualitative information content relative to single-electrode detectors.

Several characteristics make EC methods attractive for metal species analysis and studies of environmentally

significant metals in all types of waters. First, EC methods have the particular advantage of being, in principle, a species-sensitive method rather than an element-sensitive method since current is proportional to concentration and potential is proportional to chemical potential (free energy) of each species. Second, they also have sensitivities for simultaneous determination of various metals. Finally, EC methods have important advantages in that the sample requires much less handling and is in contact with fewer sources of contamination than when other speciation methods, such as solvent extraction, dialysis, or ultrafiltration, are used.

There has also been increasing emphasis in recent years on the need for continuous or automated monitoring in areas such as environmental monitoring and pollution control. This trend leads to the use of EC techniques in conjunction with a flow injection system for the analysis of small samples. Several factors have motivated this recent effort. The primary reason for this trend is its high sensitivity. This is due to an improved mass transfer rate that is caused by pumping the solution toward the electrode surface since most of the analyte in an unstirred cell is not located at the electrode surface. Secondly, the hydrodynamic boundary theory (9) of the diffusion layer has been well developed. Third, EC methods include a tendency to reject timeconsuming methodologies of conventional voltammetric

electrolysis applied to static solutions in macrocells which are usually associated with manual transfer of solutions and lengthy deaeration periods. Finally, it is relatively simple, reproducible, and economic. Furthermore, one particular advantage of the use of flow injection for EC detection is the ease of mixing reagent streams, including supporting electrolyte, with the sample stream to promote the detection process. The other intrinsic qualities of this system which are favorable include small sample volumes, minimal reagent consumption, and a high sample throughput.

EC methods have some drawbacks. One of the main limitations of EC methods is generation of charging current, which causes sensitivity decrease, due to potential scans. EC methods are dynamic systems that draw current through the solution and disturb ionic equilibria. Davison (10) found that, during the metal deposition step, the equilibrium among metal species was disturbed as a result of the electrode potential across the electrode-solution interface. The gradual fouling of the electrode surface is always a problem with EC methods. Wang (11) and DeCastro (12) found the gradual loss of electrode activity by various organic surfactants during cyclic scans. The interference due to intermetallic compounds and adsorption-desorption processes on the electrode surface is another problem (2).

The use of series and parallel combinations of working

electrodes in a flow stream can avoid the need for potential scans which are often impractical due to generation of charging currents during potential scans and short duration of solute peaks. The combination of coulometric and amperometric electrode designs is a potentially powerful technique to remove interferences and improve detection limits. Extension of these ideas leads to multi-electrode voltammetry (MEV) in which an array of electrodes in a flowthrough cell at different potentials provides a pseudovoltammogram for speciation studies (13).

These ideas can be accomplished by designing a flowthrough EC cell containing multi-electrodes to which various fixed potentials in a range are applied. The number of electrodes required by this system depends upon the accessible potential range of electrode material and the potential difference between the neighboring electrodes. Commercial instrumentation for electrochemical measurements of this type costs several thousand dollars and one instrument would be required for each electrode. Hence, a special inexpensive potentiostat will be needed to supply the selected potential range to this multi-electrode cell.

Potential control can be provided by placing a combined counter-reference electrode at circuit ground and connecting each working electrode to the inverting input of a current follower which has a control potential applied to its noninverting input from a resistor network. Current followers

are used for integrating signal currents generated by the analytes on each electrode, respectively. The dynamic range of the current follower can be extended by placing a capacitor and a parallel switch instead of a resistor in the feedback loop. The output of each integrating current follower, corresponding to total faradaic charge, can be converted to digital values for easy handling, processing, and storing. Hence a custom designed and built circuitry will be needed to fulfill data acquisition since no commercial instruments of this type are available.

Finally, all of the circuits can be controllable by a personal computer through a computer interface and software programs. With a large, but reasonable number of electrodes at tens of millivolt intervals, a current-potential curve of adequate resolution can be constructed in a fraction of a second with no degradation of signal-to-noise ratio compared to single electrode detection. The qualitative and quantitative results of speciation can be obtained through a pseudo-voltammogram which is a plot of currents (charge per time interval) versus selected, applied potentials on each electrode. This pseudo-voltammogram may have the classical voltammogram shape.

The main advantages of this system will be that (i) the double layer charging current will not enter the measurement and therefore the sensitivity and accuracy will increased; (ii) the rate of mass transfer at the electrode surface will

be larger than that of diffusion alone and therefore the signal will be increased and the detection limit will be lowered; (iii) minimum sample handling will be required to avoid contamination; (iv) the instrument can be used for routine analysis like other EC techniques and reliably differentiate metal species; and (v) the system can also be used as a detector for HPLC and ion chromatography. Furthermore, the EC cell in a flow system linked with this data acquisition system will make this technique even more attractive.

CHAPTER II

FLOW INJECTION ANALYSIS SYSTEM AND INSTRUMENTATION

The entire system consists of a flow injection analysis (FIA) system, a potential source, an integrating current follower unit, and a data acquisition system, all interfaced with a Commodore 64 personal computer (C-64) system. The block diagram is shown in Figure 1. The C-64 sends out the signals via the control of the software to the digital-toanalog converters which then generate and deliver the required potentials to the electrodes of the FIA system via a voltage divider. The analog currents produced by electrolysis at each electrode are sent to the current follower unit to be integrated. The data acquisition system then converts these integrated analog signals, one by one, to the digital values through the selection of the multiplexers. These values are then temporarily saved in the RAM of the C-64. Finally, these values are combined according to the "resolution time", the time interval as a basic unit for current integration, in order to reduce the requirement of the RAM of the C-64. These combined values are saved on a floppy disk for further data manipulation. Details of each unit, including interface, are discussed in





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the following sections.

FLOW INJECTION ANALYSIS SYSTEM

The Flow Injection Analysis (FIA) system used for this research consists of a supporting electrolyte reservoir, a pump, a pulse damper and ground current isolator, a sample injection valve, and a flow-through electrochemical (EC) cell.

Electrochemical Cell Configuration

The heart of an FIA system is a flow-through EC cell. The concerns of the EC cell design for quantitative applications are the linearity of the response, the precision, and the long term stability. The concerns for qualitative applications are accurate control of electrode potential and an adequate response time for electrochemical tasks. In either case, the sensitivity and the detection limit are of concern. In order to obtain the optimum performance of this cell, certain design conditions have to be kept in mind. These include (i) high sensitivity, (ii) well-defined hydrodynamic characteristics, (iii) low effective volume, (iv) low ohmic drop, (v) ease of construction and maintenance, and (vi) a constant effective electrode area. The performance of this cell depends on the electrode material, the electrode configuration, and the cell configuration.

A variety of cell geometries, differing mainly in the

flow type and the position of the electrode, has been explored over the course of two decades. Stulik and Pacáková (14) have discussed the different flow-through EC cell geometries, including the tubular, planar, and wall-jet type. The characteristic common to most of these configurations is that the rate of transport of electrochemically active species to the electrode surface is determined by the flow rate of solution.

However, according to theoretical comparison of the performance of these EC cells by Hanekamp and de Jong (15), the best design of flow-through EC cell is the wall-jet type. The term "wall-jet", first defined by Glauert (16), describes the phenomenon when a jet of fluid strikes a wall perpendicularly and then spreads rapidly over the surface of the wall. The electrode is mounted in the wall. The hydrodynamic theory and applications of the wall-jet electrode are fully developed. Yamada and Matsuda (17) have derived the theoretical expression of the limiting diffusion current and the experimental verification of the theoretical predictions for this type electrode. Gunasingham and Fleet (9) have derived the equation for the hydrodynamic boundary layer thickness in terms of the hydrodynamic parameters characteristic of the wall-jet type electrode. The hydrodynamic properties and mass transfer characteristics of this type of EC cell have also been investigated by Dalhuijsen et al.(18). The wall-jet type EC cell offers

high sensitivity, low solution hold-up, and is thus suited for EC measurements in flowing solution. Therefore, a walljet type EC cell design is chosen for this research.

Since the behavior of the wall-jet type EC cell depends on whether or not the cell allows for unrestricted development of the hydrodynamic boundary layer, the cell fabricated for this research is somewhat different from that of the original wall-jet concept and is, in effect, a constricted wall-jet with the working electrodes forming a smaller cell volume compared to the earlier design of Yamada and Matsuda (17) and the original design by Fleet and Little (19).

In a multi-electrode (ME) cell, the effluent from the up-stream unit will be analyzed at the down-stream working electrode. Therefore, neither Ag/AgCl nor saturated calomel electrode (SCE) can be used as a reference electrode because the chloride ion in the reference electrode compartment may leak into the flow stream and be oxidized at the down-stream working electrode. On the other hand, Zn/Zn^{2+} in 1 M sodium acetate electrode is well-known and the half-cell potential of this electrode is very stable at low current densities. Hence, the zinc/zinc acetate electrode was chosen as the reference electrode for this research.

Fabrication of Wall-Jet Type Electrochemical Cell Body

The fabrication of a single unit of the EC cell body, containing a reference electrode compartment, is described as follows. Kel-F (poly(chlorotrifluoroethylene)) (3M, USA) is used as the material to fabricate the cell body because of its wide range of desirable properties, including its being electrochemically inert and its ease of fabrication in any desired shape. Figure 2 shows a single unit of the EC cell body. In each unit a short piece, ca. 2 cm length, of zinc strip housed in an annulus compartment surrounding the inlet nozzle is soldered to a copper wire which is sealed by epoxy in a hole through the side of the cell. The inlet nozzle diameter is 0.34 mm with a funnel-shaped opening on the upstream side.

Working Electrode Material and Configuration

The heart of EC cells is the working electrode. The working electrodes in the EC cell must: (i) ideally be stable for an infinite period of time, (ii) be simple to prepare; and (iii) most importantly, have an excellent signal-to-noise (S/N) ratio. Employing electrode materials and configurations, which have a minimum amount of residual current and yet yield maximum current in response to electrochemically active species, can maximize the S/N ratio.



Top View



Side View

Figure 2. Schematic illustration of a single unit of the electrochemical cell, containing a reference electrode compartment. (a) Flow inlet with 0.34 mm diameter; (b) zinc strip; (c) lead; (d) reference electrode compartment; (e) flow direction. The use of solid electrodes, including carbon and metal electrodes, in EC cells has gained popularity. One of the primary reasons for this is their applicability to anodic oxidations. Such electrodes thus permit important analytical applications not amenable to the mercury electrode. Secondarily, they are easy to construct and use in a flow system. The sensitivity and detection limit are thus increased by placing an electrode in the flow system. However, metal electrodes such as platinum and gold have not been as widely used as carbon electrodes because they tend to exhibit higher detection limits when used at positive potentials. This is because higher anodic residual currents arise from the oxygen evolution at lower positive potentials (20).

A variety of carbon electrodes has been used in EC cells for a long time mainly because they produce better S/N ratio in response to electrochemically-active species in contrast to other electrode materials. Furthermore, several different forms of the carbon used as electrode materials are also readily available. However, most of the carbon used as electrode material has drawbacks. Pure graphite has had limited use as an electrode material because it exhibits high porosity which leads to high residual or background currents (21). Glassy carbon (22) has been popular because of its relatively reproducible performance. It can also be used at both positive and negative potentials in aqueous

solution with a range of ca. 2 V. However, glassy carbon electrodes have residual currents that are higher and tend not to give such a favorable S/N ratio compared with carbon paste or composite electrodes (23). Carbon paste electrodes (20) have been extensively used for anodic oxidation reactions due to their low residual current and wide positive potential range. However, these electrodes also have several drawbacks. First, from preparation to preparation, carbon paste electrodes have variations. Second, the dissolution of binder in organic solution restricts the use of this kind of electrode. Third, the binder may restrict the potential range of the electrode.

Electrochemists noted that composite electrodes such as Kel-Graf, which consist of Kel-F and graphite, showed improved S/N ratio (24). These electrodes also exhibit less dependence on flow-dependent noise. These improvements are attributed to the existence of small graphite particles. When the surface of the electrode is examined under the microscope, the small graphite particles existing on the surface of the electrodes are isolated from each other and act as a multiple array of microelectrodes. Anderson et al. (25) have found that small diameter electrodes are easier to use at high potential, have less noise at all potentials, and have higher sensitivity than ordinary homogeneous electrodes.

Two theories have been derived to explain these

results. First is the edge effect (26). At large electrodes, the majority of the steady current arises mainly from the diffusion of the analyte in a region of solution perpendicular to the electrode surface. The amount of current due to the analyte arriving from the lateral part of this volume is negligible. If the electrode surface has a large perimeter-to-area (P/A) ratio such as graphite particles, lateral diffusion will become more important. Therefore, an increase of the P/A ratio of an electrode can increase the sensitivity and lower the detection limit.

The second effect is the noise dependency on surface area. There is an identified noise source, i.e., incomplete wetting of the electrode surface (27). It causes variations in the path of solution flow and fluctuations in the applied potential. Since these two variations are proportional to the electrode surface areas, a reduction in the electrode surface area will reduce this noise level. Therefore, using microelectrodes can capture the advantages of both the edge effect and the less noise dependency on electrode diameter. The S/N ratio can be increased with these microelectrodes.

At present, carbon fibers have been shown to be the most promising electrode materials and are used as microelectrodes due to their possession of characteristics of small electrodes that overcome many of the imperfections of carbon paste electrodes and other carbon electrodes. For example, although the carbon fiber microelectrode has a

sensitivity similar to the glassy carbon electrode, the detection limits improve relative to glassy carbon electrode at more positive potentials because the background signal of the carbon fiber microelectrodes is less dependent on fluctuations in the velocity of the mobile phase (27). Schulze and Frenzel (28) have shown that carbon fiber microelectrodes give a very low background current, good peak resolution, and good reproducibility.

Several electrochemists have demonstrated that the carbon fiber array electrode (CFAE) (24), i.e., multiple carbon fiber microelectrodes connected in parallel, yield much higher currents which have the same desirable features of single carbon fiber microelectrodes. In addition to the current amplification effect, a number of other advantages have recently been reported (27). These are an improved S/N ratio, flow independence, and higher sensitivity in flowthrough EC cells. Furthermore, the EC response of this kind of electrode has also been shown to be relatively constant with time. Hence, carbon fibers were chosen as the electrode material to fabricate CFAEs.

Carbon fibers are microscopic carbon materials formed from the pyrolysis of pitch or polyacrylonitrile. These fibers bonded with polymer can be formed in a variety of shapes and can also be machined to optimum dimensions. The documented morphology (29) and manufacture of carbon fibers suggests that the surface of the fiber perpendicular to the

fiber axis is ideal for electrochemical applications. Therefore, the electrodes are fabricated with the electrode surface perpendicular to the fiber axis.

Fabrication of the Carbon Fiber Array Electrode

The carbon fiber array electrode (CFAE) is fabricated by inserting a bundle of 10-cm long carbon fibers of 8 μ m nominal diameter (Panex 30, Stackpole Fibers Co., Mass), ca. 0.0015 g, into a short piece of thin-walled Teflon tube of 0.30 mm inner diameter so that ca. 8-9 cm of the fiber bundle protrudes. This protruded portion of the fiber bundle is thoroughly coated with an excess of an epoxy mixture consisting of 5 parts ISOCHEMREZ 1312 LV resin mixed with 1 part 15AX hardener (Isochem Resins Co., Lincoln, R.I.). Then, the fiber bundle is pulled into the Teflon tube. The fiber bundle is then cured overnight at 60°C to form a solid rod. The rod containing epoxy-embedded carbon fibers is cut into 0.8 cm lengths to make several CFAEs. Then, electrical contact to the electrode is made by using electrically conducting silver-loaded paint (Silver Print, GC Electronics, Rockford, IL) to connect a copper lead to one end of a CFAE.

Fabrication of the Working Electrode Disk Containing 4 CFAEs

Four CFAEs and four 2-cm long pieces of Teflon tube of 0.76 mm outer diameter are carefully arranged to ensure that all electrical contacts of CFAEs are completely buried in

the epoxy mixture and all electrodes and Teflon tubes are also centered in the mold, shown in Figure 3. The same composition epoxy mixture is injected into the mold by means of a plastic syringe, then cured for 24 hours at 60°C to form a Working Electrode Disk (WED). After the WED is released from the mold, four Teflon tubes are removed. Both sides of the disk are polished with 600 grit emery paper perpendicular to the carbon fiber axis until the surface is The surface with exposed CFAEs is then successively flat. polished on a felt polishing cloth (Buehler Ltd., Evanston, IL) with 5.0, 1.0, 0.3, and 0.05 μ m diameter alumina mixed with distilled water until a mirror finish is obtained. Residual polishing material is removed from the surface by sonication in a water bath for five minutes after each polishing process. In all polishing steps, a figure-8 pattern is employed to prevent uneven polishing of the WED surface. CFAEs contained in a WED and prepared by these procedures require electrochemical activation before use.

The fabricated WED was designed to be used in the walljet type EC cell. Examination of the disk surface under a low-power microscope (of power 35X) revealed an array of carbon fibers whose ends are exposed in the plane of the WED.



Figure 3. Schematic illustrations of the working electrode disk fabrication mold and a working electrode disk. (a,e) Mold holder; (b) positioner of CFAE and solution exit hole; (c) stainless plug; (d) mold body.

Wall-Jet Type Electrochemical Cell Assembly

A single unit assembly of the EC cell is schematically illustrated in Figure 4. A zinc/zinc acetate reference electrode is made by completely filling the annulus compartment surrounding the inlet nozzle with 1 M sodium acetate solution. The potential of this zinc/zinc acetate reference electrode measured versus a SCE is -1.055 V. This reference electrode makes contact with the cell cavity via a piece of annulus-shaped Nafion-12 film (E.I. duPont de Nemours & Co.(Inc.), DE). The Nafion-12 film contains a centered hole, which is large enough to just uncover the nozzle but completely covers the reference electrode compartment. A WED sandwiched by two gaskets is incased into the upper compartment of the cell body. Gaskets which contain a centered hole large enough to expose the central area of the WED are made from Kalrez sheet (#K05011, E.I. duPont de Nemours & Co.(Inc.), DE). The solution introduced through the inlet nozzle hits the center of the WED, spreads radially, exits through the four holes in that disk, and then enters into the nozzle of the next unit of the EC cell.

Each single unit is connected on-line by three screws, which compress both Kalrez gaskets when tightened to prevent solution leaks. The inlet-electrode separation can be varied by tightening the three bolts on the cell. Four single units are then stacked together to form a wall-jet type ME cell. However, any number of single units may be



Figure 4. Schematic illustration of the wall-jet type multi-electrode cell. (a) Top piece; (b) Nafion film; (c) Kalrez spacer; (d) working electrode disk; (e) reference electrode holder; (f) bottom piece. stacked together to form an ME cell. The cell is simple to construct, and the cell components are demountable for ease of maintenance.

Flow Injection Analysis System Assembly

The schematic diagram of the assembled FIA system is shown in Figure 5. Teflon tubing with 1.59 mm outer diameter and polyethylene fittings are used for all flow lines and connections between units except where specified. A graduated cylinder serves as a reservoir (a) to store the supporting electrolyte solution. A 0.45 μ m filter (b), connected to the suction side of a minipump (d) (LDC/Milton Roy, Model 396-57, Reviera Beach, FL) through a piece of 3.18 mm outer diameter Teflon tube, is submerged in the supporting electrolyte solution. At the discharge side of the pump, a pressure gauge (e) is connected to the outlet of the pump and a plastic tube (g), ca. 70 ml capacity. The plastic tube acts as a flow pulse damper and ground-current insulator. At the down stream of the plastic tube, a sample injection valve (Hamilton, Model HV LX6-6, Reno, NV) with a sample loop (h) of 1 ml is connected on-line just ahead of the multi-electrode (ME) cell containing sixteen CFAEs and four zinc/zinc acetate reference electrodes. The connection tube between the sample injection valve and the ME cell is made as short as possible. The outlet of this ME cell is put into a waste solution container.


Figure 5. Flow injection analysis system setup. (a) Supporting electrolyte reservoir; (b) filter; (c) supporting electrolyte; (d) minipump; (e) pressure gauge; (f) Teflon tube; (g) pulse damper and ground current insulator; (h) sample injection valve with a sample loop; (i) multi-electrode cell; (j) waste solution container.

INSTRUMENTATION

The circuits, including two digital-to-analog converters (DACs), an integrating current follower unit, a data acquisition system, and an interface, are contained on two circuit boards. These two circuit boards are connected together by two ribbon cables. An external power supply of +/-15 volt and +5 volt is also required. A block diagram of the entire instrumentation is shown in Figure 6. The details of these units, including the interface, are discussed in the following sections. A complete schematic diagram of the circuitry with the component identification is shown in Appendix A.

Computer System

A Commodore 64 personal computer (C-64) based on a 6510 microprocessor is used for data control. It is connected to (i) a Commodore 1541 floppy disk drive; (ii) a Philco TV monitor; and (iii) a Fujitsu DX 2100 printer to handle command executions for the control of the potentiostat, data acquisition, data storage, and data management. The block diagram of the computer system is shown in Figure 7. The signal lines, including the address bus and the data bus, of the expansion port of the C-64 are buffered by two tristate buffers (LS 245 & LS 244) because they have limited drive capacity.



Figure 6. Block diagram of the electronic instrumentation.



Figure 7. Block diagram of the computer system.

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Potential Source

The block diagram of the potential source, which supplies the required potentials to the working electrodes relative to the reference electrode (i.e., circuit ground), is shown in Figure 8. A short BASIC program as shown in Appendix B and a digital voltmeter are used to set up the voltage outputs.

The input signals are obtained from two 12-bit DACs. The two constant current outputs from these two DACs are set, respectively, via the control of the software. These two current outputs are then converted into two voltages by two operational amplifiers. Both voltage outputs are connected to a voltage divider consisting of fifteen 1-K ohm resistors (R2) in series. These voltage outputs can be changed via the control of the same software.

Integrating Current Follower Unit

A block diagram of the integrating current follower unit is shown in Figure 9. It consists of sixteen integrating current followers with sixteen reset switches, respectively, and switch-controlling circuitry. The inverting input of each current follower is connected to one electrode in the multi-electrode cell. The noninverting inputs are connected to the voltage divider, described in the previous section. The analog signals generated by the electrodes are fed to the integrating current followers and



Figure 8. Block diagram of the potential source.

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Figure 9. Block diagram of the integrating current follower unit.

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continuously integrated. All integrators can be reset, one by one, via the control of an assembly language subroutine as described in Appendix D.

Data Acquisition System

Figure 10 shows a block diagram of the microprocessorbased data acquisition system. It is comprised of the following components: (i) A sample and hold (S/H) amplifier, including an S/H chip (LF 298, National Semiconductor), to perform the "hold" state for the conversion of analog signals which are generated by the electrolytes to the digital values; (ii) A peripheral interface adapter (PIA) circuitry, including a PIA chip (MC 68B21P, Motorola) which contains two 8-bit bi-directional data buses and four interrupt/control lines for interfacing the C-64 with the peripheral devices; (iii) A 12-bit analog-to-digital converter (ADC) (AD574A, Analog Devices) for converting electrical signals from the analog domain to the digital domain. This 12-bit ADC resolves the full-scale voltage range into 4096 units. (iv) A multiplexer circuitry, including three multiplexers (DG 508 & LF 13508), forming a two-level structure for sequentially allowing only one output of the current followers to reach the S/H chip (LF 298) at a time. (v) A decoder circuitry, including a decoder chip (LS 138) that is used to address and control several chips.





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Under the control of the software, only one output of the current followers is selected and reaches the S/H chip via the multiplexers. Then the A/D conversion circuitry converts this analog signal from the output of the current follower to a 12-bit parallel digital signal. The output of the 12-bit A/D conversion is read in as two bytes and is addressed in the memory space to be stored to accommodate the 6510 microprocessor's 8-bit data bus. After the signal is stored, the output of next current follower is sent to the S/H chip via a different combination of the opened channels of the multiplexers. Then the cycle is repeated again and again until the preset time period is over.

The A/D conversion cycle is timed under the control of a cycle-counting subroutine in an assembly language subroutine. The C-64 itself provides two presettable 16-bit down counters (30) which can be either independently programed as two separated timers or jointly programed to The data acquisition frequency is controlled form a clock. by these two presettable 16-bit down-counter timers. Timer #1 is used to set a time interval for each integrating current collected. It is set as a multiple of 1/60 sec (60 Hz) in order to get rid of the noises generated by 60 Hz power lines. Timer #2 is set so that the time of a cycle is one second longer than that of Timer #1 in order to acquire the integrating time of current signals. The parameters to set up these two timers are interrelated. They also

depended on the sample size, the sample concentration, and the RAM capacity of the C-64. The combination of Timer #2 and a cycle counter is used to record the real time at each data acquisition and the total covered time period for each on-line analysis.

Software

Several software programs are used in conjunction with this multi-electrode voltammetric instrumentation. All of them are written in BASIC language except one, named "DATACQ", which is written in assembly language. The main program, named "COLLECT.DATA", is used to control the entire instrumentation, to obtain coulometric data, and to save data onto a disk. "DATACQ" software is a subroutine called up by this main program to convert analog signals generated at the electrodes to digital values and to store them in the RAM of the C-64. Two more software programs are used for further data management. The first one, named "NORMALIZE", is used for the surface area normalization factors of 16 CFAEs in the ME cell. The second one, named "RESULT", is used to get amperometric data for plotting pseudovoltammograms after the data have been normalized. A11 software programs are discussed and attached in Appendices C, D, E, and F.

CHAPTER III

MATERIALS AND APPARATUS

MATERIALS

All the materials used are analytical grade and are used without any further purification. The water used to prepare all solutions is triply distilled in a quartz still (Q-water) and stored in 1-liter polyethylene bottles.

1 M Acetate Buffer Stock Solution

82.03 g of anhydrous sodium acetate and 17.2 ml of glacial acetic acid are dissolved and diluted with Q-water to equal 1 liter. The solution is then filtered through 0.45 micrometer Millipore filters (Millipore, Type HA). This solution is prepared weekly.

<u>Argon Gas</u>

Argon gas (AIRCO Inc., Montvale, N.J.) without any further purification is used.

0.1 M Acetate Buffer Solution

0.1 M acetate buffer solution is prepared by dilution of 1 M acetate stock solution with Q-water.

Millimolar Hydroquinone Sample Solution

Hydroquinone solutions are prepared daily by dissolving the appropriate amount of hydroquinone in 0.1 M acetate buffer solution, followed by deoxygenation with argon gas for 1 hour prior to use.

1 M Sodium Acetate Solution

8.20 g of anhydrous sodium acetate is dissolved and diluted by Q-water to 100 ml.

APPARATUS

Apparatus for Cyclic Voltammetry

A 25-mL beaker is employed as an electrochemical (EC) cell body. A carbon fiber array electrode (CFAE) contained in a working electrode disk (WED) is used as a working electrode, a saturated calomel electrode (SCE) as a reference electrode, and a platinum wire as an auxiliary electrode. All three electrodes are joined to the beaker through holes in its plastic cover to form a conventional EC cell. This cell in conjunction with a potentiostat (HEKA, Type PG28, West Germany) and an X-Y recorder (Linear Instruments Corporation, Model 8036, Irvine, CA) is used to perform cyclic voltammetry. A square wave function generator (EXACT Electronics, Inc., Model 120, Hillsboro, OR) is used in conjunction with the potentiostat to reactivate the CFAEs. Potentials are monitored with a digital multimeter (John Fluke Co. Inc., Model FLUKE 75, Everett, WA). All potentials measured and reported for this 3-electrode cell are referred to the SCE.

Apparatus for Pseudo-Voltammetry

The previously discussed multi-electrode (ME) cell and instrumentation are used to perform the measurements. In the ME cell, 16 CFAEs contained in 4 WEDs are deployed as working electrodes. For each 4 CFAEs on the same WED, a zinc strip in 1 M sodium acetate solution is used as a reference electrode. The potential of these reference electrodes measured versus a SCE is -1.055 V. The potentials applied on the CFAEs are generated by the potentiostat via the control of the software. The whole flow injection analysis system is as described in the previous chapter.

CHAPTER IV

EXPERIMENTAL PROCEDURES

EXPERIMENTS WITH A CONVENTIONAL ELECTROCHEMICAL CELL

Since the electrochemical (EC) behavior of electrodes can be evaluated by cyclic voltammetry, cyclic voltammograms were used to examine the performance of carbon fiber array electrodes (CFAEs) on the working electrode disk (WED). All preliminary experiments were performed in an unstirred solution with a conventional EC cell. This conventional EC cell is described in the previous chapter. All potentials were measured and reported versus a saturated calomel electrode (SCE) for this conventional cell.

All cyclic scans, which were performed with the same CFAE on a WED in a solution containing approximately 1 mM HQ in 0.1 M acetate buffer (AB) solution at a scan rate of 20 mV/sec, started at +0.0 V with automatic scan reversal at +0.9 V and then scanned back to +0.0 V except where specified. The cyclic voltammograms obtained were recorded on the X-Y recorder.

Square Wave Parameter Optimization

To evaluate the results of CFAE reactivation, the same conventional EC cell was used to perform cyclic voltammetry. The electrode reactivation process involves cycling a square wave, generated by a function generator, on a CFAE dipped in a 20 ml 0.1 M AB solution for a period of time. The different combinations of the square wave parameters, including the upper and lower potential limits, frequencies, and duration, were studied in order to find the optimized parameters of a square wave for CFAE reactivation.

After potential cycling, the electrode was moved to a hydroquinone (HQ) solution which had been purged by argon gas and held at the initial potential (+0.0 V) of a cyclic scan for a time delay period until the current generated by the electrode reactivation process on the CFAE subsided. Different time delay periods were also investigated. Finally, a cyclic scan with a scan rate of 20 mV/s was performed as described in the previous section.

EXPERIMENTS WITH THE MULTI-ELECTRODE CELL

Before WEDs were deployed in the flow cell, every CFAE on WEDs was individually examined for noise level and current leakage by cyclic voltammetry with a conventional cell. After the assembly of the ME cell, CFAEs were never polished again. All the potentials measured and reported for this ME cell referred to the zinc/zinc acetate reference electrode.

The experimental procedures are the same for both CFAE normalization factors and pseudo-voltammetry. A complete

experimental cycle involves three steps: connection of the ME cell on-line, reactivation of CFAEs in the cell (including time delay for subsidence of the currents generated by the reactivation process), and coulometric data collection.

Multi-Electrode Cell On-Line

The 0.1 M AB solution was first pumped through the flow system at a constant flow rate produced by a minipump without connection of the ME cell to the circuits. This avoids supplying an unpredictable voltage to CFAEs in the cell, which may cause damage to the CFAEs. The flow rate was measured with a graduated cylinder and a stop watch. The ME cell was then pumped several times by using a plastic syringe (which was temporarily connected with the outlet of the cell via a connector) in order to avoid air bubbles being trapped in the cell.

Prior to performing the experiments, the potentials required at all CFAEs in the cell for either CFAE normalization factors or pseudo-voltammetry were determined by the cyclic voltammograms obtained from the preliminary experiments. For normalization factors of CFAE surface areas in the ME cell, a constant potential of +1.95 V was applied to all CFAEs. In contrast, the potentials in the range between +1.05 V and +1.95 V, equally divided, were applied to CFAEs, respectively, in ME cell for pseudovoltammetry. This potential range started before the

oxidation of HQ occurs and ended at where maximum steadystate current is reached, i.e., the concentration of HQ at the electrode surface approaches zero.

The software program "COLLECT.DATA", loaded in the RAM of the C-64, was activated first. Then, the experimental conditions were input through the keyboard. After the instrument warmed up, the required potentials for CFAEs, generated by the digital-to-analog converter (DAC) units on the circuit board, were ready. The zinc/zinc acetate reference electrodes in the ME cell were connected to the circuit ground. The CFAEs in the cell were connected, in order, to the inverting input of the current follower, respectively.

Reactivation of CFAEs in the ME Cell

During working with CFAEs, inconsistent performance and high background current with noise were always observed after they were in contact with the supporting electrolyte solution for a long period of time. The sensitivity of CFAEs was also diminished during the successive scans without the electrode reactivation. Therefore, reactivation of CFAEs was performed just prior to data collection.

The reactivation process of the CFAEs in the ME cell was performed by using a square wave which was generated by the DACs via the control of the same program. The parameters of this square wave were as follows: frequency, ca. 5 Hz; upper potential limit, +2.05 V; lower potential

limit, +1.05 V; and duration, 5 seconds. After the CFAEs were reactivated, each CFAE was held at the required potential 4 minutes to allow the residual current generated by this electrode surface reactivation process to subside.

Collection of Coulometric Data

During the wait for the subsidence of the residual currents, a sample was loaded into the sample injection loop. After the time delay period was over, data acquisition was performed as described in Appendix D. During the first 20 seconds the data collected represented the integrated background currents. After that, the sample injection loop was manually switched into the flow stream. The data collected after this point corresponded to the composition of the sample solution.

When either the setup of the sampling time was over or the RAM of the C-64 was filled up with the data, data collection was stopped. Then the data were combined according to the resolution time, followed by saving combined data onto a floppy disk along with all experimental information for either the normalization factors of CFAEs or the construction of pseudo-voltammograms.

CHAPTER V

RESULTS AND DISCUSSION

For voltammetric studies in water, the supporting electrolyte selected must be very inert and also have high overvoltages associated with the formation of hydrogen and oxygen. That was why 0.1 M acetate buffer (AB) was chosen as the supporting electrolyte for this study.

The accessible potential range of 0.1 M AB solution was determined by cyclic voltammetry in a conventional electrochemical (EC) cell with a carbon fiber array electrode (CFAE) contained in a working electrode disk (WED). As shown in Figure 11, the cyclic voltammograms (CVs) were obtained in the AB solution either without or with deoxygenation, respectively. Compared with each other, a reduction wave, caused by the reduction of O_2 to hydrogen peroxide (31), is observed at potential ca. -0.6 V in Figure However, in both CVs the currents increased notably 11a. with potentials either more positive than +1.5 V or more negative than -1.4 V, which was due to the evolution of oxygen and hydrogen from water, respectively. Hence, the accessible potential range of 0.1 M AB solution at CFAEs is between -1.4 V and +1.5 V with deoxygenation; otherwise, between -0.3 V and +1.5 V.



<u>Figure 11</u>. Cyclic voltammograms of 0.1 M acetate buffer at a carbon fiber array electrode. Scan rate, 50 mV/sec; (a) without deoxygenation; (b) with deoxygenation.

With consideration of the accessible range of the AB solution without deoxygenation, hydroquinone (HQ) was chosen as the probe to evaluate the results obtained from all experiments because its EC behavior is well-defined and it is also very sensitive to the surface state of the electrode. Furthermore, its oxidation wave falls in the potential range of 0.1 M AB solution where no other redox wave is produced.

Surface preparation is crucial for optimal performance of the CFAEs. Figure 12 shows the results of cyclic scans obtained after different polishing methods were performed. A freshly emery-polished CFAE without any other treatment showed good sensitivity, as shown in Figure 12a. However, the observed background current, noise levels, and half-wave potentials $(E_{1/2}$'s) of HQ obtained after each polishing of a CFAE were different; even though identical polishing procedures were employed. Kamau (22) has found that a variety of surface functional groups created by polishing causes the variation of the background current and noise The $E_{1/2}$ shift may be caused by anodization of the level. electrode surface. The background current and noise level were decreased dramatically by using a freshly aluminapolished CFAE, as shown in Figure 12b. The $E_{1/2}$, compared with Figure 12a, shifts to a more anodic value. If the alumina-polished CFAE was cleaned either by sonic waves or a cyclic scan, the noise level and background current



Figure 12. Effects of carbon fiber array electrode surface preparation upon the cyclic voltammogram. Scan rate, 20 mV/sec; (1) 0.1 M AB; (2) 1 mM HQ; (a) freshly emery-polished CFAE; (b) freshly alumina-polished CFAE; (c) freshly sonic wave-cleaned CFAE; (d) freshly cyclic scan-cleaned CFAE.

decreased a small amount as shown in Figure 12c and 12d. The $E_{1/2}$ shifts slightly in the anodic direction.

REACTIVATION OF CARBON FIBER ARRAY ELECTRODES

Electrochemists have observed that the performance of electrodes is always hampered by a gradually fouling of the surface during exposure of an electrode to solutions (31, 32). This leads to a gradual decrease in detection limit and sensitivity of the electrode. Such loss of electrode surface activity is usually attributed to the adsorption of analyte itself, reaction products, or impurities on the electrode surface (32).

The occurrence of analyte adsorption on the CFAE surface is demonstrated in Figure 13. Figure 13b shows a small bump at ca. +0.3 V (vs. SCE) after leaving a CFAE in HQ solution with 0.0 V applied on it for 1 hour. However, Figure 13a does not have any bump after leaving the same electrode in AB solution for the same time. This indicates that slow impurities adsorption on the CFAE surface was taking place during CFAE exposure to HQ solution. It also indicates that the adsorbent on the CFAE surface is HQ. Furthermore, the amount of HQ adsorbed on the electrode surface is proportional to time, as shown in Figure 13c, d, and e. However, this small bump disappeared and a reproducible CV was obtained after several successive scans, as shown in Figure 13c. It indicates that the adsorbent on



Figure 13. Time effects upon the cyclic voltammogram. Scan rate, 20 mV/sec; (a) after leaving a CFAE in AB for 1 hour with 0 volt applied at it; (b,d) after leaving a CFAE in HQ solution for 1 hour; (e) after leaving it in HQ solution for 3 hours; (c) after successive cyclic scans. Voltammograms are obtained with (a,b) 0.1 M AB; (c,d,e) 1 mM HQ. the electrode surface can be removed by cyclic scans. It also indicates that CFAEs can reach a reproducible condition.

After leaving this activated CFAE in AB solution overnight and without potential applied at it, the $E_{1/2}$ shift in the cathodic direction by up to 200 mV and the decreased separation $E_{p}-E_{p/2}$ were observed, as shown in Figure 14a, compared with the values obtained in Figure 13c. It indicates that the electrode reverted to a typical condition after leaving it in the AB solution overnight without a potential applied on it. Fortunately, during successive cyclic scans between +0.0 V and +0.9 V in HQ solution, the $E_{1/2}$ gradually shifted back. And a reproducible CV was obtained eventually, as shown in Figure 14b. It indicates that the electrode surface is in a reproducible condition. This process can also be hastened by applying a positive potential at the CFAE, depending on how positive the potential is and how long it is applied. Therefore, CFAEs must be reactivated in order to get a reproducible result. A simple and rapid electrode reactivation method for the CFAEs in a multi-electrode (ME) cell needs to be found to produce a reproducible surface condition.



Figure 14. Effects of an applied potential at a carbon fiber array electrode upon the cyclic voltammogram. Solution, 0.9 mM HQ; scan rate, 20 mV/sec; (a) after leaving a CFAE in 0.1 M AB overnight and without potential applied at it; (b) after the reproducible result obtained with cyclic scans in 0.1 M AB.

In Situ Carbon Fiber Array Electrode Reactivation

Many methods used to reactivate electrode surfaces have been reported (33), such as a surface polished with an abrasive material, treated with radio frequency plasmas, heated at low pressures, irradiated with a laser, and treated with potential cycling. With the consideration of the configuration and instrumentation of multi-electrode voltammetry (MEV), the electrode surface reactivation with a square wave appears to be the most convenient means of achieving in situ surface activation to improve the performance of graphite electrodes in the cell. The primary reason is because a square wave is very easy to generate with software. The CFAEs placed in the ME cell are intended to be used for a considerable time without cell disassembly. When CFAEs become fouled, they can be reactivated by this method without the need of cell disassembly.

The work discussed herein is to determine the parameters of the square wave used for CFAE reactivation and to improve the sensitivity and reproducibility of CFAEs. The parameters include the upper and lower potential limits, frequency, and duration. A CFAE known to exhibit good performance was used to test the different combinations of experimental conditions and to determine all the parameters of the square wave. The apparatus and experimental procedures were described in Chapter III and IV. All potentials measured and reported in this section are versus

the saturated calomel reference electrode (SCE) except where specified. The effect of CFAE surface reactivation using a square wave is evaluated by potential shifts and limiting current variations of cyclic voltammograms generated by the HQ solution.

Applying a very positive potential on the CFAE is not advisable because this may cause the destruction of the electrode (20). According to the preliminary experiments, a dramatic loss of sensitivity was observed if the upper potential limit of the square wave supplied at a CFAE was over ± 2.0 V and lasted a period of several minutes. The sensitivity lost under this condition can not be recovered by EC reactivation except by repolishing the electrode. Also, the accessible range of AB solution is between ± 1.4 V and ± 1.5 V (vs. SCE). Beyond that range, gas bubbles due to the formation of H₂ and O₂ by the decomposition of H₂O may be trapped on the electrode surface and cause a change of electrode activity. Therefore, the potential limits for the square wave need to be between ± 1.4 V and ± 1.5 V.

Figure 15 shows the CVs of HQ obtained after a CFAE reactivated by a square wave at frequency of 0.5 HZ, duration of 30 sec, upper potential limit of +0.9 V, but with different lower potential limits. Comparison of all CVs in Figure 15, $E_{1/2}$ as well as the limiting currents on the plateau shows that they are the same. The results obtained are independent of the lower potential limit of the



Figure 15. Effects of lower potential limit of a square wave upon the cyclic voltammogram. Solution, 1 mM HQ; scan rate, 50 mV/sec; (a) a cyclic scan-reactivated CFAE; a square wave-reactivated CFAE with parameters: upper potential limit, +0.9 V; lower potential limit, (b) -0.2 V, (c) -0.6 V, (d) -1.6 V; frequency, 0.5 Hz; duration, 30 sec; delay, 4 min. square wave.

The correlation between the upper potential limit of the square wave and the EC behavior of the reactivated CFAE is shown in Figure 16. The experimental conditions were the same as in Figure 15 except the lower potential limit is 0 V and with different upper potential limits, respectively. Comparison of all CVs in Figure 16 shows that the $E_{1/2}$'s are constant when the upper potential limits of the square wave are not greater than +1.4 V. The shape of curves and the limiting currents on the plateau are the same. However, as the upper potential limit of the square wave is made more positive than +1.4 V, remarkable effects on the curve shape are observed, i.e., a poorly-defined wave and $E_{1/2}$ shift toward more anodic direction, as shown in Figure 16d. Therefore, the upper potential limit of the square wave chosen must be not greater than +1.4 V.

The correlation between the frequency of the square wave and the EC behavior of the reactivated CFAE is shown in Figure 17. The experimental conditions were the same as in Figure 16 except the upper and lower potential limits of the square wave were ± 1.0 V and ± 0.1 V, respectively, and with different frequencies. Comparison of these curves shows that there is no significant difference when a variety of frequencies are applied. Figure 18 shows the correlation between the duration of the square wave and the EC behavior of the reactivated CFAE. The experimental conditions were



Figure 16. Effects of upper potential limit of a square wave upon the cyclic voltammogram. Solution, 1 mM HQ; scan rate, 50 mV/sec; (a) a cyclic scan-reactivated CFAE; a square wavereactivated CFAE with parameters: upper potential limit, (b) +1.1 V, (c) +1.3 V, (d) +1.5 V; lower potential limit, 0 V; frequency, 0.5 Hz; duration, 30 sec; delay, 4 min.



Figure 17. Effects of frequency of a square wave upon the cyclic voltammogram. Solution, 1 mM HQ; scan rate, 50 mV/sec; (a) a cyclic scanreactivated CFAE; a square wave-reactivated CFAE with parameters: upper potential limit, +1.0 V; lower potential limit, -0.1 V; frequency, (b) 5 Hz, (c) 50 Hz, (d) 500 Hz; duration, 30 sec; delay, 4 min.



Figure 18. Effects of duration of a square wave upon the cyclic voltammogram. Solution, 1 mM HQ; scan rate, 50 mV/sec; (a) a cyclic scanreactivated CFAE; a square wave-reactivated CFAE with parameters: upper potential limit, +1.0 V; lower potential limit, -0.1 V; frequency, 0.5 Hz; duration, (b) 5 sec, (c) 15 sec, (d) 30 sec; delay, 4 min. the same as in Figure 17 except the frequency was 0.5 Hz and with different durations. Comparison of all CVs also shows that the EC behaviors of the CFAE shows no significant changes when different duration square waves are applied. Therefore, EC behavior of the reactivated CFAE are independent of the frequency and the duration of the square wave.

Subsidence of the Charging Current

After the CFAE reactivation process stopped, the switch of the potential at CFAEs to the required potentials for data collection generates a current spike which is due to double-layer charging. Figure 19 shows the decay of this current versus time immediately after a CFAE was reactivated with a square wave, which had the upper potential limit, +1.0 V; the lower potential limit, -0.1 V; duration, 30 sec; and frequency, 0.5 HZ. This current decayed very rapidly during the first two minutes, then gradually decreased at a constant rate. Prior to performing a cyclic scan, a fourminute delay was used to allow decay of the current generated by the electrode reactivation; this procedure was adopted for all the following experiments. During this time of delay, the potential at a CFAE was held at the required value for data collection. Therefore, a square wave having an upper potential limit, +1.0 V; a lower potential limit, -0.1 V; a frequency, 1.1 HZ; and duration, 30 sec; with delay



Figure 19. Decay of a charging current after the applied potential at a carbon fiber array electrode was switched to 0 V.
time, 4 min; was chosen for the CFAE reactivation. And the CVs obtained were reproducible if the CFAEs were reactivated with these optimized parameters.

Surface Morphology of the Reactivated CFAE

After a CFAE was reactivated by a square wave, the electrode surface was examined by scanning electron microscopy (SEM) and X-ray photoelectron spectrum (XPS) with the assistance of Dr. John Dash (34). The SEM of the CFAE surface showed, as in Figure 20, that carbon fibers are parallel-aligned and the ends of carbon fibers extended out of the epoxy surface. Most carbon fibers are partially isolated from each other. This arrangement of carbon fibers in a CFAE will give good sensitivity and low detection limit due to the edge effect. The XPS of the CAFE surface is shown in Figure 21. The surface is not completely cleaned and still have trace amounts of different metals, such as Al and Si. The dispersed alumina and emery particles are present on the electrode surface even after cleaning by sonic waves and square waves.

CYCLIC VOLTAMMETRY IN A FLOW CELL

In order to compare the performance of CFAEs in the designed ME cell with that in a conventional cell, a ME cell having a single electrode disk was assembled, as described in Chapter II. The experimental procedures were the same as



Figure 20. Scanning electron micrograph of a carbon fiber array electrode. Magnification is 500 X.



Figure 21. X-ray photoelectron spectrum of a carbon fiber array electrode.

described in Chapter IV except the pump was stopped after the HQ solution was injected into the EC cell. The potential held at the CFAE was +1.05 V (vs. the zinc/zinc acetate electrode) during the subsidence of the charging current. After that, cyclic scans were performed at a scan rate of 50 mV/s. The shape of voltammograms, as shown in Figure 22, is strongly dependent upon the tightness of the The residual currents increased as the screws of screws. the flow cell were tightened up, as shown in Figure 22a. This change in residual current was believed to be caused by the severe distortion of the CFAE disk in the flow cell due to the unevenly tightened screws. Also, the shape of the cyclic voltammogram of HQ became not well-defined, as shown in Figure 22b. A small bump was also shown at ca. +1.3 V, which might be caused by the trapped HQ in the crack adsorbing on the electrode surface. If the screws were not completely tightened, air bubbles trapped in the flow cell were not able to be removed by applying vacuum. The zinc acetate solution encapsulated in the reference electrode compartment could leak out. This problem was not correctable without changing the design of the cell body and the fabrication method of WED.

CHARACTERISTICS OF FLOW INJECTION ANALYSIS SYSTEM

Minimization of the Potential Drop in the ME Cell

One of the primary concerns of designing an EC cell is



E(V vs. Zn/Zn²⁺ in 1 M NaCH₃COO)



Figure 22. Effects of tightness of the cell screws upon the cyclic voltammogram. (a) 0.1 M AB; (b) 1 mM HQ; tightness of the screws: (1) very loose, (2) loose, (3) tight, (4) very tight.

to minimize the potential drop across a solution. This concern is especially important for designing a ME cell. Α flow cell generally has an advantage of high mass-transport rate of analytes, which increases sensitivity due to current increase. However, this current increase may also lead to an increased potential drop and consequent impairment of efficient potential control at the working electrode. In a ME cell, the current response at one electrode which causes a potential drop across the solution may also affect the potentials across all electrode-solution interfaces. This resulting "cross-talk" can render a ME cell useless if the potential drop is too large. Both can be minimized by decreasing either the cell impedance or the current.

The minimization of current response at CFAEs was accomplished by making the electrode surface area small. The minimization of the cell impedance was achieved by using the supporting electrolyte combined with positioning the reference electrode as close as possible to the working electrode, i.e., by placing a zinc/zinc acetate reference electrode in a compartment directly across a thin layer from the CFAEs with the separation of a Nafion film, as shown in Figure 4.

Nozzle-Electrode Surface Separation

For the flow cell, although the equation for the limiting current derived by Yammada and Matsuda (17) does not contain any term relating to the inlet-electrode surface

separation, it has been well-known that, according to the theory (35), the jet nozzle must be located outside the hydrodynamic boundary layer for the maximum efficiency obtained from the electrode; otherwise, interference of the flow in the boundary layer would result in a loss of current efficiency. Also, the jet nozzle-electrode surface separation must not be too far; otherwise the jet may spread out and cause loss of resolution as the distance from the nozzle increased.

With these considerations in mind, the jet nozzle must be designed to remain outside the hydrodynamic boundary layer even at high flow rate. The hydrodynamic boundary layer thickness (δ_{bl}) has been calculated with the following equation by Gunasingham and Fleet (9).

> $$\begin{split} \delta_{bl} &= 5.8 \ \pi^{3/4} \ a^{1/2} \ v^{3/4} \ x^{5/4} \ v^{-3/4} \end{split}$$
> where a : diameter of the nozzle (cm). V : a volume flow rate (cm³/s). v : the kinematic viscosity (10⁻² cm²/s for aqueous system). x : the distance from the electrode surface

Substituting representative values for this flow system into the above equation, where $v = 10^{-2} \text{ cm}^2/\text{s}$, inlet diameter of 0.343 mm, an electrode with a radius of ca. 1 mm at flow rate between 1.1 - 3.3 cm³/min, the boundary layer thickness

(cm).

is between 0.4 and 0.8 mm. The jet nozzle-electrode separation of this ME cell was ca. 0.8 mm, i.e., a thickness of a Nafion film and a Kalrez film, which is just adequate to keep the nozzle of the inlet outside the hydrodynamic boundary. This separation can be slightly adjusted by tightening the screws. The diameters of the centered hole on the Nafion film and the Kalrez spacer are large enough to avoid disturbing the hydrodynamic boundary layer of the jet flow. The inlet diameter of the jet was 0.343 mm because the cross section of an ideal wall-jet need to be as small as possible, compared to the electrode surface area.

Geometric and Effective ME Cell Volumes

One of the characteristic features of the wall-jet cell, not present in other designs, is that the effective cell volume is only the order of the hydrodynamic boundary layer and not the geometric cell volume. Since the effective cell volume at each CFAE is very small, as seen from the previous section, the detection process does not produce appreciable change of sample concentration down stream. Although the larger geometric cell volume does not effect the hydrodynamic boundary layer, it may cause broadening of the sample band. The cell volume needs to be as small as possible.

The geometric cell volume is calculated as follows:

In each single unit, Volume of the inlet = $\pi r^2 l$ = 5.28 $\times 10^4$ cm³ where the inlet diameter (r) is 0.343 mm; the inlet length (l) is 5.72 mm Volume on the electrode disk = ($\pi r^2 l$) x 4 = 9.69 x 10⁻³ cm³ where the tube diameter in hole is 3.18 mm; the inlet length is 0.305 mm Volume of the center hole in the Nafion film = $\pi r^2 l = 9.88 \times 10^4$ cm³ where the center hole diameter is 1.78 mm; the thickness of Nafion film is 0.397 mm Volume of the center hole in two Kalrez films = ($\pi r^2 l$) x 2 = 1.30 x 10⁻² cm³ where the center hole diameter is 4.56 mm the thickness of Kalrez film is 0.397 mm

There are four units in the ME cell. The total geometric volume is ca. 9.7×10^{-2} cm³. Since the geometric cell volume is small, compared with the sample in the sample injection loop (ca. 0.8 ml), and the distance between the sample injection valve and the cell was made very short, the sample band broadening by diffusion was insignificant through the whole cell.

The Pulse Damper and Ground Current Insulator

The minipump used to pump solutions through the flow system is a synchronous reciprocating pump which has advantages of having reproducible flow and a small internal volume. The relationship between pump scale and flow rate is shown in Table I. Figure 23 shows that the flow rate is proportional to the pump scale, but not linear. This pump also produces a pulsating flow in the flow system and conducts a small ground current through the system from the AC power line. Since the mass-transport rate of analytes or impurities in the solution is proportional to the flow rate, this flow fluctuation caused by the minipump was a noise source for MEV. The unwanted ground current conducted by the pump was imposed on the measured current and caused the sensitivity to be decreased. A simple device was made to overcome both drawbacks. It is made from two pieces of plastic syringe tube connected with a piece of glass tube, then placed between the minipump and the sample injection valve. This tube was only partially filled with solution. The empty part of the capacity served as both the pulse damper and the ground current isolator. With the use of this device flow fluctuation was decreased dramatically and so was noise. The ground current was also insulated from the ME cell.

THE TRUE CAPACITANCE AND LEAKAGE CURRENT OF THE INTEGRATOR

Although the nominal values of the capacitance of the integrators are known from the specifications, the true values can be different. The leakage current may be generated by the offset voltage between positive input and negative input of the amplifier. A set of dummy resistors

TABLE I

RELATIONSHIP OF PUMP SCALE AND FLOW RATE

Pump Scale	Time for 10 mL (sec)
20.0	540
40.0	265
50.0 60.0	210 180



Figure 23. Correlation between the flow rate and the pump scale.

and a voltage source were used to simulate the current generation by the electroactive species at each capacitor and the current leakage generated by the offset voltage of the amplifier. One side of the resistor was connected with the negative input of the amplifier. The other side was connected with a potential source. Two potentials which were symmetrical to the circuit ground were used to obtain the currents which was generated by the voltage drop across the resistor in order to solve the combined equations for two unknown values of every integrator. The equation used is as follows:

$$E_{out} = \frac{1}{C} \int \frac{V}{R} dt + \frac{1}{C} \int i_{leak} dt$$

where E_{out}: output voltage of an amplifier; C: capacitance in Coulomb; V: potential applied at electrode; R: measured resistance in ohm; t: time in sec; i_{leak}: leakage current of current follower;

10 MΩ resistors were chosen in order to lessen the chance of too large a signal entering the current integrating circuits, whose maximum limit is +/-12 V. The true resistance of each resistor was measured by a digital multimeter. The calculated capacitance of each capacitor and

leakage current of each integrator are shown in Table II. The calculated leakage currents were all very small, compared with the resolution ability of the analog-todigital converter. They were insignificant and were ignored during the data handling for MEV.

TABLE II

RESISTANCE, LEAKAGE CURRENT, AND CAPACITANCE OF THE INTEGRATOR

Integrator	Resistance	Capacitance	Leakage Current
-	(MΩ)	(nF)	(nA)
# 1	10.10	0.901	0.225
# 2	9.97	0.863	0.276
# 3	10.07	0.898	0.340
# 4	10.13	0.841	0.262
# 5	9.93	0.839	0.369
# 6	10.09	0.902	0.315
# 7	9.98	0.895	0.240
# 8	9.95	0.867	0.474
# 9	10.13	0.924	-0.040
# 10	10.24	0.892	-0.083
# 11	10.14	0.897	-0.306
# 12	10.12	0.897	-0.016
# 13	9.98	0.913	0.258
# 14	9.95	0.892	0.020
# 15	10.08	0.824	0.099
# 16	9.87	0.856	0.047

* Measured by a multi-digital meter. ** Based on 4 sets of data.

NORMALIZATION FACTORS OF CARBON FIBER ARRAY ELECTRODES IN THE MULTI-ELECTRODE CELL

In the ME cell, 16 CFAEs are employed. Careful procedures have been taken to assure that reproducible results were obtained at each electrode, such as electrode reactivation procedures. The pseudo-voltammograms constructed are based on the data collected at all 16 CFAES. The signals generated at each CFAE under the same experimental condition may differ because the electrode surface areas are not always the same, which might be caused by some carbon fibers being broken during the fabricating process, even though all CFAEs were fabricated the same way. Depending on the way the ME cell was assembled, the flow patterns at each CFAE in the cell may also differ. It has also been shown CFAEs foul gradually due to the adsorption of impurities on the surface. Hence, the surface areas of CFAEs in the ME cell need to be calibrated just prior to the collection of data for MEV in order to obtain the active electrode surface.

The experimental procedures of electrode surface area normalization were described in Chapter IV. The manipulation of amperometric data for CFAE normalization factors were described in Appendix E. All potentials measured and reported in this section are relative to the zinc/zinc acetate reference electrode. Prior to data collection for normalization factors of the CFAEs in the ME cell, a potential sufficient to spontaneously oxidize HQ and also to provide the maximum sensitivity at CFAEs needs to be determined. The potential meeting this requirement is in the limiting-current plateau of the CV. However, when the applied potential at CFAEs is at the positive end of the CV plateau, the increases of the background current generated

by the oxidation of H_2O and of the noise level caused by the evolution of O_2 become limiting factors. Hence, the best sensitivity can be obtained by choosing a potential which is near the middle of the CV plateau. A potential of +1.95 V was chosen as a compromise between maximizing the analyte current response and minimizing the background current for CFAE surface area normalization. 0.5 mM HQ solution was used for optimum S/N ratio and system performance.

Flow Rate Dependency

A disadvantage of amperometric electrodes in flow cells is the noticeable effect of flow rate on the analytical signal since mass transfer in the flow system is proportional to flow rate. The flow rate dependence of the current is substantially reduced when microelectrodes are used in flow cells (27, 36). The dependency of the flow rates for normalization factors of CFAEs in the ME cell was therefore studied.

During preliminary experiments, sudden disappearance of signals at some electrodes and the expulsion of small gas bubbles from the ME cell into its outlet occurred more often when the flow was slower. It indicates that small gas bubbles generated by the fluctuation of pressure in the cell are adsorbed on the electrode surface more often when the flow is slow. Inhomogeneity of the sample solution caused by dispersion of sample was observed. Low flow rates were excluded from the study. When the flow rate is too high,

the pulse damper becomes ineffective because cyclicly fluctuating data due to pulsating flow have been observed. Hence, a range of flow rate from 1.67 to 3.33 ml/min was studied, i.e., the pump scale set between 30.0 and 60.0.

To evaluate the performance of the ME cell, the residual currents and the net currents obtained at an applied potential of ± 1.95 V (vs. Zinc/Zinc Acetate electrode) in 5.45 x 10^4 HQ solution with different flow rates are compared. Table III shows the residual currents and the net currents obtained at different flow rates. Both the residual and the net currents show only a small dependence on flow rate as is expected for ultramicroelectrodes.

The currents are different at each electrode because the effective areas are different. A normalizing factor can be obtained from the data of Table III by calculation the ratio of the current at each electrode to the current at one electrode. Table IV shows the normalized factors of CFAEs relative to electrode #2 at different flow rates. The relative standard deviations are between 0.5 % and 3.6 %.

Repeatability and Reproducibility

Insufficient stability and reproducibility are often drawbacks of EC flow cells. Since air bubbles were much easier to be formed at the low flow rate, both stability and

TABLE III

INFLUENCE OF THE FLOW RATE ON THE CARBON FIBER ARRAY ELECTRODE RESPONSE CURRENT

		Currer	nt (Amp)	
		at Flow	v Rate of	
	1.67 m	L/Min	2.22 ml	L/min
Electrode	Residual	Net	Residual	Net
# 1	5.26E-10	3.11E-8	4.96E-10	3.19E-8
# 2	4.81E-10	3.61E-8	4.77E-10	3.88E-8
# 3	5.39E-10	4.82E-8	4.90E-10	5.11E-8
# 4	2.56E-10	4.02E-8	2.44E-10	4.19E-8
# 5	1.41E-09	5.74E-8	1.17E-09	5.82E-8
# 6	1.61E-09	6.08E-8	1.35E-09	6.31E-8
# 7	1.93E-09	3.99E-8	1.69E-09	4.03E-8
# 8	9.87E-10	5.36E-8	8.95E-10	5.80E-8
# 9	1.60E-09	4.96E-8	1.31E-09	5.02E-8
# 10	1.88E-09	4.82E-8	1.62E-09	5.06E-8
# 11	2.10E-09	4.88E-8	1.71E-09	5.04E-8
# 12	1.75E-09	4.04E-8	1.43E-09	4.16E-8
# 13	7.79E-10	5.07E-8	6.90E-10	5.40E-8
# 14	1.07E-09	5.48E-8	9.98E-10	5.822-8
# 15	2.22E-09	6.03E-8	2.10E-09	6.36E-8
# 16	4.80E-09	5.60E-8	4.07E-09	5.89E-8

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Current (Amp)

		at FIOW	v Race or	
	2.86 ml	L/Min	3.33 ml	L/min
Electrode	Residual	Net	Residual	Net
# 1	6.12E-10	3.16E-8	6.12E-10	3.27E-8
# 2	5.89E-10	3.94E-8	5.85E-10	4.11E-8
# 3	6.81E-10	5.21E-8	6.34E-10	5.43E-8
#4	3.06E-10	4.22E-8	3.10E-10	4.27E-8
# 5	1.85E-09	6.06E-8	1.74E-09	6.14E-8
#6	2.20E-09	6.65E-8	2.07E-09	6.91E-8
# 7	2.37E-09	4.22E-8	2.28E-09	4.48E-8
# 8	1.33E-09	5.91E-8	1.23E-09	6.11E-8
# 9	2.23E-09	5.29E-8	2.13E-09	5.48E-8
# 10	2.65E-09	5.21E-8	2.49E-09	5.19E-8
# 11	3.12E-09	5.34E-8	2.97E-09	5.55E-8
# 12	2.73E-09	4.36E-8	2.38E-09	4.61E-8
# 13	9.96E-10	5.74E-8	9.05E-10	6.06E-8
# 14	1.42E-09	6.21E-8	1.35E-09	6.50E-8
# 15	2.63E-09	6.51E-8	2.58E-09	6.68E-8
# 16	6.49E-09	6.24E-8	6.27E-09	6.23E-8
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TABLE IV

CORRELATION	BETWEEN	FLOW RATE AND	NORMALIZATION	FACTORS
Electrode	Norma	lization Facto	r at Flow Rate	(mL/Min)
	1.67	2.22	2.86	3.33
# 1	0.86	0.82	0.80	0.80
# 2	1.00	1.00	1.00	1.00
# 3	1.33	1.31	1.32	1.32
# 4	1.11	1.08	1.07	1.04
# 5	1.59	1.50	1.54	1.50
# 6	1.68	1.62	1.69	1.68
# 7	1.11	1.04	1.07	1.09
# 8	1.48	1.49	1.50	1.49
# 9	1.37	1.29	1.34	1.33
# 10	1.34	1.30	1.32	1.26
# 11	1.35	1.30	1.36	1.35
# 12	1.12	1.07	1.11	1.12
# 13	1.40	1.39	1.46	1.48
 # 14	1.52	1.50	1.58	1.58
# 15	1.67	1.64	1.65	1.63
# 16	1.55	1.52	1.59	1.52

		Normalization H	Tactor	
Electrode	Mean	Std. Dev.	Rel. Std.	Dev.(%)
# 1	0.82	0.03	3.6	
# 2	1.00	0.00	0.0	
# 3	1.32	0.01	0.6	
# 4	1.08	0.03	2.7	
# 5	1.53	0.04	2.9	
# 6	1.67	0.03	1.8	
# 7	1.08	0.03	2.7	
# 8	1.49	0.01	0.5	
# 9	1.34	0.03	2.4	
# 10	1.30	0.03	2.4	
# 11	1.34	0.03	2.1	
# 12	1.10	0.02	2.1	
# 13	1.43	0.04	2.9	
# 14	1.54	0.04	2.7	
# 15	1.65	0.02	1.1	
# 16	1.54	0.03	2.1	

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reproducibility of normalization was studied at the flow rate of 3.33 mL/min. Table V shows the repeatability of a series of three injections of 5.45 x 10⁻⁴ M HQ in 0.1 M AB solution. The relative standard deviations of the net limiting current for 16 CFAEs in the ME cell scatters between 0.5 % and 5.4 %. Table VI shows the reproducibility of three injections of the same HQ solution in a 6 hour period. For these injections, the relative standard deviations ranges between 2.2 % and 6.9 %.

Normalization Factor

The average of the net currents obtained at the flow rate of 3.33 mL/min for repeatability studies, as shown in Table V, were used to calculate the normalization factors of CFAEs in the ME cell. The procedures are described in Appendix E. The normalization factors, as shown in Table VII, were used to normalize all amperometric data for construction of pseudo-voltammograms.

MULTI-ELECTRODE VOLTAMMETRY

The analytical responses of MEV were examined with HQ solutions at a flow rate of 3.33 mL/min. The experimental procedures were described in Chapter IV. The pseudovoltammograms are constructed by plotting the normalized amperometric data vs. the applied potentials at each CFAE. All potentials reported in this section are relative to the

TABLE V

REPEATABILITY OF CARBON FIBER ARRAY ELECTRODE RESPONSE CURRENTS AT APPLIED POTENTIAL OF +1.95 V

		Net Current (Amp)	
Electrode	Set 1	Set 2	Set 3
# 1	3.27E-8	3.25E-8	3.11E-8
# 2	4.10E-8	4.00E-8	3.93E-8
# 3	5.43E-8	5.12E-8	5.09E-8
# 4	4.27E-8	4.10E-8	4.00E-8
# 5	6.14E-8	5.98E-8	5.53E-8
# 6	6.91E-8	6.76E-8	6.52E-8
# 7	4.48E-8	4.34E-8	4.35E-8
# 8	6.10E-8	5.95E-8	5.74E-8
# 9	5.48E-8	5.29E-8	5.07E-8
# 10	5.19E-8	5.14E-8	5.04E-8
# 11	5.55E-8	5.44E-8	5.29E-8
# 12	4.61E-8	4.32E-8	4.37E-8
# 13	6.06E-8	5.82E-8	5.78E-8
# 14	6.50E-8	6.37E-8	6.24E-8
# 15	6.68E-8	6.65E-8	6.60E-8
# 16	6.25E-8	6.01E-8	5-82E-8

		Net Current	(Amp)	
Electrode	Mean	Std. Dev.	Rel. STd. Dev.	(%)
# 1	3.21E-8	8.36E-10	2.6	
# 2	4.01E-8	8.64E-10	2.1	
# 3	5.21E-8	1.88E-09	3.6	
# 4	4.12E-8	1.36E-09	3.2	
# 5	5.88E-8	3.20E-09	5.4	
# 6	6.73E-8	1.95E-09	2.9	
# 7	4.39E-8	7.72E-10	1.7	
# 8	5.93E-8	1.81E-09	3.0	
# 9	5.28E-8	2.02E-09	3.8	
# 10	5.12E-8	7.57E-10	1.4	
# 11	5.43E-8	1.33E-09	2.4	
# 12	4.43E-8	1.56E-09	3.5	
# 13	5.89E-8	1.50E-09	2.5	
# 14	6.37E-8	1.30E-09	2.0	
# 15	6.64E-8	3.79E-10	0.5	
# 16	6.03E-8	2.18E-09	3.6	

TABLE VI

REPRODUCIBILITY OF CARBON FIBER ARRAY ELECTRODE RESPONSE CURRENTS AT THE APPLIED POTENTIAL OF +1.95 V

		Net Current (Amp)	
Electrode	Set 1	Set 2	Set 3
# 1	3.27E-8	3.42E-8	3.16E-8
# 2	4.11E-8	4.17E-8	3.99E-8
# 3	5.43E-8	5.36E-8	5.16E-8
# 4	4.27E-8	4.77E-8	4.21E-8
<u></u> # 5	6.14E-8	6.61E-8	6.15E-8
# 6	6.91E-8	7.19E-8	6.56E-8
# 7	4.48E-8	4.63E-8	4.42E-8
# 8	6.11E-8	6.21E-8	5.90E-8
# 9	5.48E-8	5.52E-8	5.22E-8
# 10	5.19E-8	5.28E-8	4.93E-8
# 11	5.55E-8	5.75E-8	5.33E-8
# 12	4.61E-8	4.63E-8	4.44E-8
# 13	6.06E-8	6.56E-8	6.25E-8
# 14	6.50E-8	6.92E-8	6.30E-8
# 15	6.68E-8	7.16E-8	6.97E-8
# 16	6.23E-8	6.42E-8	6.88E-8

		Net Current	(Amp)
Electrode	Mean	Std. Dev.	Rel. STd. Dev. (%)
# 1	3.28E-8	1.28E-09	3.9
# 2	4.09E-8	9.09E-10	2.2
# 3	5.32E-8	1.40E-09	2.6
# 4	4.42E-8	3.07E-09	6.9
# 5	6.30E-8	2.69E-09	4.2
# 6	6.88E-8	3.18E-09	4.6
# 7	4.51E-8	1.05E-09	2.3
# 8	6.07E-8	1.60E-09	2.6
# 9	5.40E-8	1.66E-09	3.0
# 10	5.13E-8	1.84E-09	3.5
# 11	5.54E-8	2.13E-09	3.8
# 12	4.56E-8	1.02E-09	2.2
# 13	6.29E-8	2.52E-09	4.0
# 14	6.57E-8	3.14E-09	4.7
# 15	6.93E-8	2.42E-09	3.4
# 16	6.51E-8	3.368-09	5.1

TABLE VII

CARBON FIBER ARRAY ELECTRODE NORMALIZATION FACTORS FOR PSEUDO-VOLTAMMOGRAMS

Elect	rode	Ave. Net Current (Amp)	Normalization Factor
#	1	3.21E-8	0.80
#	2	4.01E-8	1.00
#	3	5.21E-8	1.30
#	4	4.12E-8	1.03
#	5	5.88E-8	1.46
#	6	6.73E-8	1.68
#	7	4.39E-8	1.09
#	8	5.93E-8	1.48
#	9	5.28E-8	1.32
#	10	5.12E-8	1.28
#	11	5.43E-8	1.35
#	12	4.43E-8	1.10
#	13	5.89E-8	1.47
#	14	6.37E-8	1.59
#	15	6.64E-8	1.66
#	16	6.03E-8	1.50

zinc/zinc acetate reference electrode.

HQ concentrations ranging from 7.26 x 10^{-5} M up to 5.45 x 10^{-4} M were used to study the analytical response of this instrumentation. The potential range between +1.05 V and +1.95 V was applied at the CFAEs in the multi-electrode cell.

Reproducibility of Multi-Electrode Voltammetry

The reproducibility of MEV was studied by a series of three injections of a 5.45×10^{-4} M HQ solution at the flow rate of 3.33 mL/min in a 6-hour period. The normalized net currents at each CFAE are shown in Table VIII. The relative standard deviations scatter between 167% in the residual current region and 0.6 % in the limiting current region.

TABLE VIII

REPRODUCIBILITY OF MULTI-ELECTRODE VOLTAMMETRY

	Potential (V)	Ne	t Current (Amp)
Electrode	Applied	Set 1	Set 2	Set 3
# 1	1.05	-3.23E-11	4.45E-12	-5.67E-11
# 2	1.11	7.87E-12	3.46E-11	6.07E-11
# 3	1.17	-3.02E-11	-1.86E-12	1.08E-12
# 4	1.23	-1.22E-11	-8.87E-12	-1.44E-11
# 5	1.29	-2.72E-11	-3.88E-11	-2.33E-11
# 6	1.35	1.18E-11	1.42E-11	4.20E-11
# 7	1.41	5.23E-10	5.13E-10	5.25E-10
# 8	1.47	3.26E-09	3.20E-09	3.20E-09
# 9	1.53	2.51E-09	2.54E-09	2.64E-09
# 10	1.59	7.68E-09	7.72E-09	7.99E-09
# 11	1.65	1.13E-08	1.18E-08	1.23E-08
# 12	1.71	1.53E-08	1.74E-08	1.72E-08
# 13	1.77	2.25E-08	2.34E-08	2.58E-08
# 14	1.83	2.78E-08	2.88E-08	2.90E-08
# 15	1.89	3.27E-08	3.23E-08	3.27E-08
# 16	1.95	3.92E-08	3.75E-08	3.72E-08

	Net Current (Amp)							
Electrode	Mean	Std. Dev.	Rel. Std. Dev. (%)					
# 1	-2.82E-11	3.08E-11	-109.2					
# 2	3.44E-11	2.64E-11	76.7					
# 3	-1.03E-11	1.73E-11	-167.4					
# 4	-1.18E-11	2.78E-12	-23.6					
# 5	-2.98E-11	8.09E-12	-27.2					
# 6	2.27E-11	1.68E-11	74.1					
# 7	5.20E-10	6.25E-12	1.2					
# 8	3.22E-09	3.33E-11	1.0					
# 9	2.56E-09	7.11E-11	2.7					
# 10	7.80E-09	1.66E-10	2.1					
# 11	1.18E-08	4.74E-10	4.0					
# 12	1.66E-08	1.16E-09	6.9					
# 13	2.39E-08	1.71E-09	7.1					
# 14	2.85E-08	5.95E-10	2.0					
# 15	3.26E-08	2.15E-10	0.6					
# 16	3.80E-08	1.10E-09	2.8					

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The very large relative standard deviation at low potential end of the pseudo-voltammogram is due to the small currents and the resolution of the ADC. The low relative standard deviation at high potential range indicates that MEV has very good reproducibility over this time period.

Shape of Pseudo-Voltammograms

Figure 24 shows a set of sigmoidal-shaped pseudovoltammograms constructed from the data, shown in Table IX, obtained using different concentrations of HQ solution at the flow rate of 3.33 mL/min. The shapes of these voltammograms are similar to the cyclic voltammograms obtained at a CFAE in a stationary solution, shown in Figure 22. A small bump was also shown at ca. +1.5 V (vs. the zinc/zinc acetate electrode), which might be caused by the trapped HQ in the crack adsorbing on the electrode surface. The whole instrument package worked as designed.

Calibration Curve and Detection Limit of MEV

The normalized net currents obtained at an applied potential of +1.95 V were chosen for plotting a calibration curve because the currents generated at this potential represent the limiting current. The graph plotted with these limiting currents vs. the concentrations, respectively, is shown in Figure 25. The results of the linear regression analysis of this calibration curve had a correlation coefficient of 0.9999. Obviously, the dynamic



<u>Figure 24</u>. Pseudo-voltammograms of hydroquinone at different concentrations. Flow rate, 3.33 mL/min; (a) 5.45×10^4 M; (b) 3.63×10^4 M; (c) 1.82×10^4 M; (d) 1.09×10^4 M; (e) 7.26×10^5 M.

TABLE IX

DATA OBTAINED FOR PSEUDO-VOLTAMMOGRAMS AT A FLOW RATE OF 3.33 ML/MIN

Electrode	Applied Potential	(V)
# 1	1.05	
# 2	1.11	
# 3	1.17	
# 4	1.23	
# 5	1.29	
# 6	1.35	
# 7	1.41	
# 8	1.47	
# 9	1.53	
# 10	1.59	
# 11	1.65	
# 12	1.71	
# 13	1.77	
# 14	1.83	
# 15	1.89	
# 16	1.95	

	Net Current (Amp)					
Electrode	at Concentration (M) of					
	7.26E-5	1.09E-4	1.82E-4	3.63E-4	5.45E-4	
# 1	-3.56E-11	3.77E-11	1.02E-11	6.28E-12	1.72E-11	
# 2	-2.96E-11	-7.62E-12	-4.98E-12	3.06E-12	-1.59E-11	
# 3	-4.42E-12	-1.34E-12	-1.72E-12	-1.57E-11	-3.13E-12	
# 4	-5.96E-12	-1.70E-11	-2.93E-11	7.50E-12	-1.70E-11	
# 5	-2.52E-11	-3.56E-11	-1.36E-11	-2.98E-11	-3.85E-11	
# б	-1.44E-11	-2.10E-11	-8.95E-12	1.28E-11	3.82E-11	
# 7	3.74E-11	7.98E-11	1.59E-10	3.59E-10	5.07E-10	
# 8	4.65E-10	7.00E-10	1.16E-09	2.24E-09	3.22E-09	
# 9	4.45E-10	6.76E-10	1.09E-09	2.04E-09	2.88E-09	
# 10	1.31E-09	1.80E-09	3.08E-09	5.96E-09	8.45E-09	
# 11	2.16E-09	3.25E-09	5.34E-09	1.02E-08	1.48E-08	
# 12	2.69E-09	4.08E-09	6.69E-09	1.36E-08	2.08E-08	
# 13	4.20E-09	6.14E-09	1.09E-08	2.09E-08	2.85E-08	
# 14	5.48E-09	7.94E-09	1.24E-08	2.30E-08	3.22E-08	
# 15	5.99E-09	9.00E-09	1.40E-08	2.58E-08	3.57E-08	
# 16	6.07E-09	8.66E-09	1.42E-08	2.72E-08	3.84E-08	

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Figure 25. Calibration curve of multi-electrode voltammetry. The experimental conditions are the same as in Figure 24.

range of linear response of MEV extends over two orders of magnitude of the concentration. The detection limit based on these conditions is estimated to be 5 x 10^{-5} M HQ at 95% confidence level, ie., on the basis of a 2 times the standard deviation of the calibration curve.

CHAPTER VI

CONCLUSION

Carbon electrodes have been used in electrochemical (EC) cells for a long time. There are several reasons for using them. One of the primary reasons is the applicability to anodic oxidations of many organic and inorganic species, which are not amenable to the mercury electrode. The applicable potential range on carbon electrodes is much wider compared with electrodes made of other materials. Many forms of carbon have been used to fabricate electrodes for EC detection in the oxidative mode. Among them, carbon fiber provides an inexpensive and readily fabricated alternative to materials such as glassy carbon or pyrolytic graphite, which are expensive and difficult to fabricate. Carbon fiber overcomes many drawbacks of other types of carbon electrodes (28), such as very low background current, good peak resolution, and good reproducibility.

Based on pseudo-voltammograms obtained, multi-electrode voltammetry (MEV) is an effective method for the determination of electroactive substances. The reproducibility and stability of MEV are dependent upon the careful assembly of the flow cell, including no gas bubbles in the cell and the proper tightness of the screws. The sensitivity of MEV matches most of the EC methods, but with less expensive instrumentation and more convenient operation. This good sensitivity is obtained due to two reasons. First, capacitative charging current, one of the limiting factors in all EC techniques, is reduced to insignificant proportions because of the application of constant voltages at CFAEs. Second, the rate of mass transport to CFAEs increases due to the edge effect of CFAE. As a consequence of reduced capacitative charging currents and increased mass transport rates, CFAEs exhibit good sensitivity. Electrode surface fouling by electrochemically active species and impurities is diminished in this flow system because a very small sample solution is injected into the system so that the exposure of the electrode to the sample is relatively brief. Also, electrodes are always reactivated using a square wave prior to data collection in order to obtain a fresh surface.

Two improvements for this instrumentation need to be made in the future. First, building an interface with a computer which has larger RAM capacity is necessary. This is an important point because a lot of time was wasted waiting to record and transfer data. This diminished the speed advantage of using flow injection analysis. An unique feature of this system is the option to change the time interval according to the sample size and sample concentration in order to expand the dynamic range of the

instrument. This is a very important advantage if it is used as a detector for HPLC and ion chromatography. Second, a new type of pump is needed. Gradual diminution of signal current has been occasionally observed during data collection, due to the adsorption of micro gas bubbles at CFAE surfaces. The generation of these bubbles may be caused by the combination of two factors: (1) the saturation of gas in the flow stream while the solution drips into the pulse damper in order to insulate the ground current from the AC power line and dampening the pulsating current generated by the pump; (2) the periodic pressure fluctuation in the flow cell due to the reciprocal pump.

From the constructed pseudo-voltammograms, it was shown that MEV can be used like other electrochemical techniques with minimal manipulation. It is possible that MEV can be used to determine the complexation ability between a metal ion and a ligand by the measurement of a half-wave potential shift, i.e. the stronger the binding of the complex, the larger is the shift from $E_{1/2}$ of that metal ion.

EC cells have been used as detectors for HPLC for a long time. Chemists have found that it is very unlikely that a single peak of HPLC effluent represents one and only one component of the mixture (37). The results of this study make EC detectors more important since they have the ability to determine the purity of the effluent peak. At present, EC detectors use only one or two working electrodes

and can not distinguish between eluted species on the basis of redox properties. Its resolving ability can be enhanced by the use of more than one working electrode. Series and parallel combinations in a flow stream avoid the need for potential scans which is impractical due to short duration of solute peaks and slow decay of background currents of some electrode materials. Therefore, the possibility of incorporating MEV as a detector for HPLC should be investigated. With an MEV detector, additional resolving power is available and it can help in the identification of unresolved column effluent.

Anodic stripping techniques provide concentration measurements, not speciation unless additional chemical separation steps are included. There is a source of error that has been largely ignored even though it was identified many years ago (2). This error is due to intermetallic compound formation between metals during deposition. This may decrease anodic peak height or even eliminate the peak. Further, the peaks due to oxidation of the intermetallic compounds are displaced on the potential axis so that identification may be in error. MEV can easily alleviate this problem by deploying a coulometric electrode, such as porous carbon, upstream to completely remove selected metal ions in the flow so that other ions may be deposited free of interference on the electrodes used for anodic stripping. A coulometric electrode can also remove impurities in

supporting electrolytes and/or certain contaminants of the sample matrix, which may interfere with species of interest. The combination of this coulometric electrode and MEV is a potentially powerful technique to improve detection limits.

At present, only 16 CFAEs have been installed in MEV. This system is ready to expand to as many as 64 electrodes. Choosing different electrode materials and expanding the number of electrodes in the flow cell may be studied to extend the domain of application.

REFERENCES

- (1) A. Mazzucotelli, R. Frache, A. Viarengo, and G. Martino, <u>Talanta</u>, <u>35</u>, 693 (1988).
- (2) T.M. Florence, <u>Analyst</u>, <u>111</u>, 489 (1986).
- (3) G.F. Nordburg, Sci. of Total Environ., 71, 243 (1988).
- (4) I.T. Urasa and S.H. Nam, <u>J. Chromatogr. Sci.</u>, <u>27</u>, 30 (1989).
- (5) K.S. Subramanian, <u>Anal. Chem.</u>, <u>60</u>, 11 (1988).
- (6) B.D. Karcher and I.S. Krull, <u>J. Chromatogr. Sci.</u>, <u>25</u>, 472 (1987).
- (7) J. Osteryoung, <u>Methods Enzymol.</u>, <u>158</u>, 243 (1988).
- (8) C.E. Lunte, T.H. Ridgway, and W.R. Heineman, <u>Anal.</u> <u>Chem.</u>, <u>59</u>, 761 (1987).
- (9) H. Gunasingham and B. Fleet, <u>Anal. Chem.</u>, <u>55</u>, 1409 (1983).
- (10) W. Davison, <u>J. Electroanal. Chem.</u>, <u>87</u>, 395 (1978).
- (11) J. Wang and L.D. Hutchins, <u>Anal. Chem.</u>, <u>57</u>, 1536 (1985).
- (12) E.S. Decastro, E.W. Huber, D. Villarroel, C. Galiatsatos, J.E. Mark, and W.R. Heineman, <u>Anal. Chem.</u>, <u>59</u>, 134 (1987).
- (13) D.K. Roe and I.P. Ho, Pittsburgh Conference 1984, Abst. No. 532. (1984).
- (14) K. Stulik and V. Pacáková, <u>CRC Crit. Rev. Anal. Chem.</u>, <u>14</u>, 297 (1984).
- (15) H.B. Hanekamp and H.G. de Jong, <u>Anal. Chim. Acta</u>, <u>135</u>, 351 (1982).
- (16) M.B. Glauert, <u>J. Fluid Mech.</u>, <u>1</u>, 625 (1956).
- (17) J. Yamada and H. Matsuda, <u>J. Electroanal. Chem.</u>, <u>44</u>, 189 (1973).

- (18) A.J. Dalhuijsen, Th.H. van der Meer, and C.J. Hoogendoorn, <u>J. Electroanal. Chem.</u>, <u>182</u>, 295 (1985).
- (19) B. Fleet and C.J. Little, <u>J. Chromatogr. Sci.</u>, <u>12</u>, 747 (1974).
- (20) D.C. Johnson, S.G. Weber, A.M. Bond, R.M. Wightman, R.E. Shoup, and I.S. Krull, <u>Anal. Chim. Acta</u>, <u>180</u>, 187 (1986).
- (21) W.J. Blaedel and G.A. Mabbott, <u>Anal. Chem.</u>, <u>50</u>, 933 (1978).
- (22) G.N. Kamau, Anal. Chim. Acta, 207, 1 (1988).
- (23) J.W. Dieker, W.E. van der Linden, and H. Poppe, <u>Talanta</u>, <u>25</u>, 151 (1978).
- (24) S.G. Weber, Anal. Chem., 61, 295 (1989).
- (25) J.L. Anderson, K.K. Whiten, J.D. Brewster, T.Y. Ou, and W.K. Nonidez, <u>Anal. Chem.</u>, <u>57</u>, 1366 (1985).
- (26) R.M. Wightman, Anal.Chem., 53, 1125A (1981).
- (27) W.L. Caudill, J.O. Howell, and R.M. Wightman, <u>Anal.</u> <u>Chem.</u>, <u>54</u>, 2532 (1982).
- (28) G. Schulze and W. Frenzel, <u>Anal. Chim. Acta</u>, <u>159</u>, 95 (1984).
- (29) M.A. Dayton, J.C. Brown, K.J. Stutts, and R.M. Wightman, <u>Anal. Chem.</u>, <u>52</u>, 946 (1980).
- (30) <u>Commodore 64 Programmer's Reference Guide</u>, p. 419, Commodore Business Machines, Inc. (1982).
- (31) I.F. Hu, D.H. Karweik, and T. Kuwana, <u>J. Electroanal.</u> <u>Chem.</u>, <u>188</u>, 59 (1985).
- (32) J. Wang and M.S. Lin, <u>Anal. Chem.</u>, <u>60</u>, 499 (1988).
- (33) L.J. Kepley and A.J. Bard, <u>Anal. Chem.</u>, <u>60</u>, 1459 (1988).
- (34) J. Dash, Professor, Physics Dept., Portland State Univ., P.O. Box 751, Portland, OR 97201.
- (35) H. Gunasingham, K.P.Ang, and C.C. Ngo, <u>Anal. Chem.</u>, <u>57</u>, 505 (1985).

- (36) S.B. Khoo, H. Gunasingham, K.P. Ang, and B.T. Tay, J. Electroanal. Chem., 216, 115 (1987).
- (37) T. Demana, D. Peterman, J. Shaffer, and P.R. Melaragno, <u>Anal. Chem.</u>, <u>61</u>, 1216 (1989).

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APPENDIX A

SCHEMATIC DIAGRAM OF THE CIRCUITRY WITH COMPONENT IDENTIFICATIONS

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SCHEMATIC DIAGRAM OF THE CIRCUITRY WITH COMPONENT IDENT

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E CIRCUITRY WITH COMPONENT IDENTIFICATIONS

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APPENDIX B

SOFTWARE "SET.7548.OUTPUT"

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120 REM *
               2/6/1989
                                    ٠
140 REM *
            "SET.7548.OUTPUT"
                                    ±
160 REM *
180 REM * PROGRAM TO FIND EQUATIONS
                                    *
200 REM * FOR DAC 7548 VOLT. OUTPUTS *
220 REM *
500 FL=56896:FH=56897:REM ADDRESS OF DAC 7548 #1
520 SL=56864:SH=56865:REM ADDRESS OF DAC 7548 #2
600 PRINT
620 PRINT
640 INPUT"HAVE FORMULA FOUND(Y/N)";DM$
660 IF DM$="Y" THEN GOTO 2000
1000 PRINT
1020 PRINT"** IF VH%<0 THEN END OF THIS PART !!
1040 PRINT
1200 INPUT"ENTER VH% & VL% FOR BOTH DACS"; VH%, VL%
1240 INPUT"ARE BOTH OK(Y/N)";DM$
1260 PRINT
1280 IF DM$<>"Y" THEN 1000
1300 IF VH%<0 THEN 1700
1400 POKE FL, VL%: POKE FH, VH%
1420 POKE SL, VL%: POKE SH, VH%
1440 PRINT
1460 PRINT" MEASURE THE VOLAGE OUTPUTS ON BOTH PIN 1 AND PIN 8."
1480 PRINT" AND WRITE THEM DOWN !!"
1500 PRINT
1600 GOTO 1000
1700 PRINT*LIST 2300-2700 * :STOP
2000 PRINT
2020 PRINT"** IF VOLT LESS THAN -10 VOLT"
2040 PRINT"
            THEN END OF THIS TEST !!
2060 PRINT
2100 INPUT" ENTER VOLT FOR BOTH DAC";V
2140 INPUT"
            IS IT OK(Y/N)";DM$
2160 PRINT
2180 IF DM$<>"Y" THEN 2100
2200 IF V=<-10 THEN 3000
2300 REM FORMULA FOR PIN 8(DAC#1)
2320 B1%=INT(V*4095/6.011+2037)
2340 LH%=INT(B1%/256)
2360 LL%=INT(B1%-LH%*256)
2380 POKE FL,LL%:POKE FH,LH%
2500 REM FORMULA FOR PIN 1(DAC#2)
2520 B2%=INT(V*4095/6.014+2047)
2540 VH%=INT(B2%/256)
2560 VL%=INT(B2%-VH%*256)
2580 POKE SL, VL%: POKE SH, VH%
2700 PRINT
2720 PRINT V; "VOLT APPLIED ON BOTH DAC"
2760 PRINT
2800 PRINT" MEASURE VOLAGE OUTPUTS ON PIN 1 & PIN 8"
2820 PRINT" OF LF444 AND COMPARE WITH INPUT VALUES !"
2840 PRINT
2900 GOTO 2000
3000 END
```

APPENDIX C

SOFTWARE "COLLECT.DATA"

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Written in BASIC language, the instruction-driven program "COLLECT.DATA" is used to instruct the experimental procedures and to perform the acquisition of coulometric data which represented the integrated analyte currents. The flow chart of this program is shown in Figure 26. The entire program is included at the end of this section.

It is first used to load the assembly language subroutine "DATACQ" into the designated area in the RAM of the C-64, followed by the input of the conditions to perform the experiment. After that, the reactivation of CFAEs in the ME cell with a square wave is performed under the direction of several parameters. These parameters of electrode reactivation are the upper and lower potential limits of the square wave, the frequency of the square wave, and the duration of the reactivation process.

The experimental conditions are applied to the CFAEs after their reactivation. These conditions included the potential applied on each electrode, the setup of timer #1 and #2, and the delay time to allow the charging currents caused by the reactivation process to subside. After the currents subside, the assembly language subroutine is activated to perform data acquisition, as described in the next section. The collected coulometric data are temporarily stored in the RAM of the C-64 until the preset time period is over.

Since the RAM capacity of the C-64 is so limited,



Figure 26. Flow chart of software "COLLECT.DATA".

several sets of collected data for each CFAE are then combined together in order to reduce the requirement of the RAM according to the so-called "Resolution Time". This procedure also eliminates the random errors caused by the timing of integration. Then, these combined values are stored onto a floppy disk for further data management.

If any overflow is detected during the combination of digital values, a prompt statement is printed on the monitor screen to alert the operator that the time resolution is not set appropriately. After the time resolution is changed, the data are reloaded into the matrix from the previously saved data file. Then, the combining of the digital values according to the new resolution time is performed again.

The internal clock of the C-64 is used for timing the duration of the electrode reactivation and the delay between the end of the electrode reactivation and the beginning of the data acquisition. The two internal timers of the C-64 are used for timing the frequency of the data acquisition. The parameters of these two timers were interrelated (53). Their settings are dependent on the sample size, the sample concentration, and the RAM capacity of the C-64. The use of these two timers is a unique feature of this system. Parameters for the electrode surface activation, the setup of timer #1 and #2, and experimental conditions are all input through the keyboard.

2 IF AA=Ø THEN AA=1:LOAD"DATACQ.BIN",8,1 2/9/1988 11 REM * "COLLECT.DATA" 20 REM * 21 REM * 30 REM * DATA ARE COLLECTED BY 31 REM * AN ASSEMBLY SUBROUTINE 32 REM * & ALSO SAVE INTO A FILE * 100 DIM DD%(16,700) 120 POKE 56928,128: REM KEEP ALL SWITCHES CLOSED 200 PRINT 220 PRINT"** DO NOT CONNECT CELL !!" 24Ø PRINT 300 INPUT" TODAY IS(2/9) ": DA\$ 32Ø INPUT" IS DATE OK(Y/N)"; DM\$ 340 PRINT 360 IF DM\$="N" THEN 300 400 REM DD% (M,N) = DIGITAL DATA OBTAINED 410 REM TR=TIME OF RESOLUTION 420 REM U2=LEAST U2 BYTE 430 REM TT=TOTAL COVERED TIME 440 REM UL=UPPER POTENTIAL LIMIT OF SQUARE WAVE 450 REM LL=LOWER POTENTIAL LIMIT OF SQUARE WAVE 460 REM HZ=FRQUENCY OF SQUARE WAVE 470 REM DU=DURATION OF SQUARE WAVE 480 REM TD=TIME OF DELAY TO STABLE RESIDUAL CURRENT 490 REM DA\$= DATE OF EXPT. RUN 500 AD=56832: AC=56833: REM DDR-A & CR-A OF PIA 510 BD=56834: BC=56835: REM DDR-B & CR-B OF PIA 520 CD=56960: REM ADDRESS OF CD4514 530 HC=56928: REM ADDRESS OF 74HC273 540 L1=56896:H1=56897:REM ADDRESS OF DAC7548 #1 550 L2=56864:H2=56865:REM ADDRESS OF DAC7548 #2 560 REM \$DC06 IS THE TB-LO ADDRESS OF 6526 U1 570 REM \$DD04 IS THE TA-LO ADDRESS OF 6526 U2 600 REM POKE AC,0 & POKE BC,0 610 REM POKE AD,0 & POKE BD,240 620 REM POKE AC, 52 & POKE BC, 60 630 REM 640 IS EQUILVANT TO 600,610,& 620 640 SYS 49152: REM \$C000 TO SET UP PIA 66Ø ID\$="'NONE'" 700 PRINT"** MAKE SURE FLOW PUMP IS ON" 72Ø PRINT" & INJECTOR IS AT STANDBY" 74Ø PRINT" THEN PUSH 'G' TO GO ON" 76Ø PRINT 78Ø GET A\$: IF A\$<>"G" THEN 78Ø 800 PRINT"** IF BEEP IS NOT HEARD," 820 PRINT" TURN THE VOLUME UP !!" 84Ø PRINT 860 GOSUB 8000: REM TEST BEEP 1000 GOSUB 13000:REM SET STANDBY POTENTIAL 1100 GOSUB 13500: REM APPLY STANDBY POTENTIALS TO ELECTRODES 1200 PRINT ** EVACUATE FLOW SYSTEM &" 1220 PRINT" THEN CONNECT CABLE TO IT" 124Ø PRINT" PUSH 'G' TO GO ON ! " 126Ø PRINT 1280 GET A\$: IF A\$<>"G" THEN 1280

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1 IF AA=1 THEN 100

```
1300 PRINT"-- DO NOT INJECT SAMPLE & DO NOT"
1320 PRINT" PRECONDITION ELECTRODE BECAUSE"
1340 PRINT"
              THIS IS THE FIRST RUN TO CHECK"
1360 FRINT"
             IF ELECTRODE LEAKS"
1380 PRINT
1400 INPUT TIME OF DELAY (>=180 SEC) "; TD
1420 INPUT"IS IT OK (Y/N) "; DM$
1440 PRINT
1460 IF DM$<>"Y" THEN 1400
1500 TI$="000000"
2000 INPUT"WANT TO COLLECT DATA(Y/N)";DM$
2020 PRINT
2040 IF DM$="N" THEN 5000
2100 REM SET PARAMETERS FOR DATA COLLECTION
2120 GOSUB 15000
2200 PRINT"-- NOW, TIME IS ";TI$
2220 PRINT"
             TIME MUST BE LESS THAN"; TD; "SEC"
224Ø PRINT
2260 PRINT"
              WAIT FOR BEEP !"
2280 PRINT
2400 BT=TD:GOSUB 9200:REM LET OVERSHOOT CURRENT SUBSIDE DOWN
2500 PRINT"-- NOW, COLLECTING DATA !!"
2520 PRINT
2600 POKE 56928,0:REM PIN 1 & 19 OF 74HC244 LOW
2620 SYS 49209: REM $C039 OBTAIN DATA
2640 PRINT"-- DATA'VE BEEN COLLECTED !"
2660 PRINT
2700 POKE 56928,128:REM KEEP ALL SWITCHES CLOSED
2720 GOSUB 13500: REM APPLY STANDBY POTENTIAL
2800 PRINT -- HANG ON, CHECKING DATA !"
2820 FRINT
2840 GOSUB 10500:REM FIND OUT HOW MANY SETS OBTAINED
2860 GOSUB 8000:REM BEEP
2880 PRINT"** MOVE FLOW OUTLET BACK TO CYLINDER !"
2900 PRINT
3000 GOSUB 9600: REM CHECK A DATA SET
3100 INPUT"PRINT OUT UNTREATED DATA(Y/N)";DM$
3120 PRINT
3140 IF DM$<>"Y" THEN 3200
316Ø GOSUB 20000
3200 INPUT"SATISFY WITH DATA(Y/N)"; DM$
3220 PRINT
3240 IF DM$="N" THEN 5000
3500 REM COMBINE DATA TOGETHER ACCORDING TO RESOLUTION TIME
3520 PRINT"WANT TO COMBINE DATA"
3540 INPUT" ACCORDING TO TR(Y/N) "; DM$
3560 PRINT
3580 IF DM$<>"Y" THEN 5000
3600 GOSUB 11000:REM COMBINE DATA
3620 GOSUB 8000:REM BEEF
4000 GOSUB 9600: REM CHECK COMBINED RESULTS
4100 INPUT"PRINT OUT THESE DATA (Y/N) "; DM$
412Ø FRINT
4140 IF DM$<>"Y" THEN 4200
4160 GOSUB 20000
4200 INPUT"IS TIME RESOLUTION OK(Y/N)";DM$
422Ø PRINT
4240 IF DM$<>"Y" THEN 5000
4400 INPUT WANT TO SAVE RESULTS ON DISK (Y/N) "; DM$
4420 FRINT
4440 IF DM$="N" THEN 5000
4460 GOSUB 9000: REM CHECK DISK
448Ø PRINT
4500 PRINT" SAMPLE LABEL : ";LB$
4520 PRINT" FLOW RATE : ";FR$
4540 PRINT" INI. VOLT : ";IV;" VOLT"
```

```
4560 PRINT" FINAL VOLT : ";FV;" VOLT"
4580 PRINT" PREVIOUS REMARK : ":RM$
4600 FRINT
4640 PRINT"* FOR AREA, FILE NAME STARTS WITH A- *"
4660 PRINT"* FOR SAMP, FILE NAME STARTS WITH S- *"
4700 PRINT
4720 PRINT"PREVIOUS DATA FILE ID: "; ID$
4800 INPUT"NEW DATA FILE ID"; ID$
4820 INPUT"REMARK (ND)";RM$
4840 INPUT ARE THEY OK (Y/N) "; DM$
4860 PRINT
4880 IF DM$<>"Y" THEN 4800
4900 PRINT" HANG ON, RECORDING DATA !"
4920 FRINT
4940 GOSUB 6000: REM RECORD DATA
4960 GOSUB 8000:REM BEEP
5000 INPUT"ANDTHER RUN (Y/N) "; DM$
5020 PRINT
5040 IF DM$="Y" THEN 5100
5060 PRINT"* YOU ARE SURE. RIGHT?":STOP
5080 GOTO 5000
5100 GOSUB 14000: REM SET UP EXPT. CONDITIONS
5120 GOSUB 13000: REM SET STANDBY POTENTIALS
5140 GOSUB 13500: REM APPLY STANDBY POTENTIALS
5200 PRINT"** MAKE SURE"
5220 PRINT" A SAMPLE IS READY TO INJECT"
524Ø PRINT"
              THEN PUSH 'G' TO GO ON !"
5260 FRINT
5280 GET A$: IF A$<>"G" THEN 5280
5300 INPUT"WANT TO CLEAN ELECTRODE (Y/N) "; DM$
5320 PRINT
5340 IF DM$="Y" THEN 5500
5400 PRINT"TIME OF DELAY";TD
5410 INPUT"IS IT OK(Y/N)"; DM$
5420 PRINT
5440 IF DM$="Y" THEN 5600
5460 INPUT"TIME OF DELAY(>=240 SEC)";TD
548Ø GOTO 54ØØ
5500 GOSUB 7500: REM SET UP PARAMETERS OF SQUARE WAVE
5520 GOSUB 8500: REM APPLY SQUARE WAVE
5600 GOSUB 8800: REM EXPT. POTENTIAL READY
5620 TI$="000000"
5640 PRINT"-- TIMING BEGIN ---
5660 PRINT
5680 GOSUB 8000: REM BEEP
5700 INPUT WANT TO CLEAR MATRIX (Y/N) "; DM$
5720 PRINT
5740 IF DM$<>"Y" THEN 5800
5760 GOSUB 10000: REM CLEAR MATRIX
5780 GOSUB 8000: REM BEEP
5800 GOTO 2000
5900 END
6000 REM SUBROUTINE TO SAVE DATA
6100 FRINT"--- IF DRIVE LIGHT BLINKS, STOP"
6120 FRINT" FROGRAM IMMEDIATELY. THEN TYPE"
614Ø PRINT"
              'CLOSE 2', FOLLOWED BY TYFING"
616Ø FRINT"
              'GOTO 4000' !!"
6180 PRINT
6200 REM SAVE DATA INTO A SEQ. FILE
6220 OPEN 2,8,2, ID$+DA$+", S,W"
6240 PRINT#2, DA$
6260 PRINT#2,LB$
6280 FRINT#2,FR$
6300 PRINT#2,UL
```

```
6320 PRINT#2.LL
634Ø PRINT#2,HZ
6360 FRINT#2,DU
6380 PRINT#2,TD
6400 PRINT#2, IV
6420 PRINT#2, FV
6440 PRINT#2,U1
6460 PRINT#2, TR
6480 PRINT#2,U2
6500 PRINT#2,TT
6520 PRINT#2,NC
6540 PRINT#2,RM$
6560 FOR S≈0 TO NC
6580 FOR M=0 TO 16
6600 PRINT#2,DD% (M,S)
6620 NEXT M
6640 NEXT S
666Ø CLOSE 2
6800 RETURN
7500 REM SET UP PARAMETERS OF SQUARE WAVE
7510 PRINT"SQUARE WAVE FARAMETERS FOR REACTIVATION ---
7520 PRINT" UPPER POTENTIAL LIMIT :";UL; "VOLT"
7530 PRINT" LOWER POTENTIAL LIMIT :";LL; "VOLT"
7540 PRINT" FREQUENCY : ";HZ; "HZ"
755Ø PRINT"
            DURATION :"; DU; "SEC"
756Ø PRINT
7570 PRINT"DELAY PRIOR TO COLLECT DATA :"; TD; "SEC"
758Ø PRINT
7570 INPUT"ARE ALL ABOVE OK (Y/N) "; DM$
7600 PRINT
7610 IF DM$="Y" THEN 7800
7700 PRINT"INPUT NEW SQUARE WAVE PARAMETERS ---"
7710 INPUT" UPPER LIMIT(+2.05 V)";UL
7720 INPUT" LOWER LIMIT(+0.95 V)";LL
7730 PRINT" FRED. IS CA. 5 HZ":HZ=5
7740 INPUT" DURATION (30 SEC) "; DU
775Ø PRINT
7760 INFUT"DELAY(>=240 SEC)"; TD
777Ø PRINT
7780 GOTO 7500
7800 REM CONVERT UL TO DIGITAL
7810 DM%=INT(UL+4095/6.012+2042)
7820 OH%=INT (DM%/256)
7830 OL%=INT (DM%-OH%*256)
7840 REM CONVERT LL TO DIGITAL
7850 DM%=INT(LL*4095/6.012+2042)
7860 RH%=INT(DM%/256)
7870 RL%=INT (DM%-RH%*256)
7900 RETURN
8000 REM GENERATE A BEEFER (REF. F.185)
8020 PRINT"
              PUSH 'S' TO STOP THE BEEP &"
              WATCH THE SCREEN FOR THE HINT !"
8040 PRINT"
8060 FRINT
8100 GET AS: IF AS="S" THEN 8400
8200 S=54272
8220 FOR L=S TO S+24: POKE L,0:NEXT
8240 POKE S+5,9:POKE S+6,0
8260 FOKE S+24,15
8280 HF=37:LF=162:DR=500
8300 POKE S+1,HF:POKE S,LF
8320 POKE S+4,33
8340 FOR T=1 TO DR:NEXT
8360 POKE 5+4,32:FOR T=1 TO 50:NEXT
838Ø GOTO 81ØØ
B400 RETURN
8500 REM SUBROUTINE FOR APPLYING SQUARE WAVE
```

```
8520 PRINT"-- NOW, REACTIVATE ELECTRODES &"
8540 PRINT" WAIT";DU; "SEC FOR ELECTRODE CLEAN !"
8560 FRINT
8600 REM FREQUENCY OF S.W. IS 5 HZ
8620 TI$="000000"
8640 REM APPLY UPPER LIMIT FIRST, THEN LOWER LIMIT
8660 POKE L1,0L%:POKE H1,0H%
8680 POKE L2, OL%: POKE H2, OH%
8700 FOR DM=1 TO 80: NEXT DM
8720 FOKE L1, RL%: FOKE H1, RH%
8740 POKE L2, RL%: POKE H2, RH%
8760 FOR DM=1 TO 75: NEXT DM
8780 IF TI/60<DU THEN 8660
879Ø RETURN
8800 REM APPLY EXPT. POTENTIAL TO BOTH DAC
8820 POKE L1, IL%: POKE H1, IH%
8840 POKE L2, FL%: POKE H2, FH%
8860 PRINT"-- EXPT. POTENTIALS JUST APPLIED TO"
8880 PRINT" BOTH DAC AND READY FOR ELECTRODES"
8900 PRINT
892Ø RETURN
*"
9020 PRINT"* MAKE SURE
7040 PRINT"* A RIGHT DISK IN DRIVE *"
9060 PRINT"* & SPACE ALSO ENOUGH *"
9100 PRINT
9120 INPUT WANT TO CHECK DISK (Y/N) "; DM$
914Ø PRINT
9160 IF DM$="Y" THEN STOP
9180 RETURN
9200 REM LET OVERSHOOT CURRENT SUBSIDE DOWN
9210 IF TI/60<=BT-40 THEN 9250
9220 PRINT"** DO NOT INJECT THE SAMPLE !"
9230 PRINT" TIMING IS NOT GOOD !":FRINT
924Ø GOTO 952Ø
9250 IF TI/60<=BT-40 THEN 9250
9260 GOSUB 8000:REM BEEP
9270 PRINT"-- 20 SEC LATER, MOVE"
9280 PRINT"
              FLOW OUTLET INTO THE WASTE JAR !"
9290 PRINT
9300 IF TI/60<=BT-20 THEN 9300
9320 PRINT"** PUT FLOW OUTLET INTO A WASTE JAR &"
             15 SEC LATER, QUICK INJECT SAMPLE !"
9340 FRINT"
9360 FRINT
9400 IF TI/60<=BT-5 THEN 9400
9420 PRINT"** QUICK INJECT SAMPLE !"
9440 PRINT" & NOW, TIME IS ";TI$
9460 PRINT
9500 IF TI/60<=BT THEN 9500
9520 RETURN
9600 REM CHECK FART OF RAW DATA BEFORE SAVING
9620 PRINT" -- ANY DATA SHOULD NOT = 0 OR 4095"
964Ø PRINT"
             Z MUST <=";N
9660 PRINT"
              IF Z<Ø THEN END OF CHECKING DATA"
              NEW STARTING SET #";Z
968Ø INPUT"
9700 PRINT
9720 IF Z<0 THEN 9900
9740 PRINT" M DD%Z DD%Z+1 DD%Z+2 DD%Z+3 DD%Z+4"
9760 FOR M=0 TO 16
9780 PRINT M; DD% (M,Z); DD% (M,Z+1); DD% (M,Z+2); DD% (M,Z+3); DD% (M,Z+4)
9800 NEXT M
9820 PRINT
9840 PRINT"STARTING SET #(Z) =";Z
986Ø FRINT
988Ø GOTO 962Ø
```

```
9900 RETURN
10000 REM CLEAR THE WHOLE MATRIX
10020 PRINT" HANG ON, EMPTY THE MATRIX !"
10040 FRINT
10060 FOR DM=0 TO NR+2
10080 FOR M=0 TO 16
10100 DD% (M, DM) =0
10120 NEXT M
10140 NEXT DM
10160 RETURN
10500 REM FIND OUT HOW MANY SETS OF RAW DATA COLLECTED
10520 NR=700
10540 IF DD% (0,NR) >0 THEN 10600
10560 NR=NR-1
10580 GOTO 10540
10600 FRINT"LAST DATA SET(NR) AT ";NR
10620 N=NR
10640 PRINT
10660 RETURN
11000 REM ROUTINE TO TREAT RAW DATA
11020 REM J REFRESENTS ORIGINAL ROW
11040 REM N REPRESENTS EACH ROW AFTER DATA COMBINATION
11060 REM DD%(0,N) REFRESENTS INTEGRATION TIME
11080 REM M REPRESENTS THE M ELECTRODE
11100 PRINT TIME OF RESOLUTION (TR) :"; TR; "SEC"
11120 INPUT"IS IT OK (Y/N) "; DM$
1113Ø FRINT
1114Ø IF DM$="Y" THEN 11200
11150 INPUT"NEW TR (<=1.00 SEC)";TR
1116Ø PRINT
1118Ø GOTO 111ØØ
11200 PRINT"WAIT (ABOUT 5 MIN.) TO COMBINE"
11220 PRINT" DATA ACCORDING TO TR !"
11240 PRINT
1126Ø PRINT"
             --- IF PROGRAM STOPPED BY ILLEGAL"
1127Ø PRINT"
                QUANTITY, TYPE 'GOTO 5000' !"
1128Ø PRINT
11300 D1=TR*(1.023E6):XX%=INT(D1/U2)
11310 IF DD%(0,1)<=DD%(0,2) THEN 12200
1132Ø DD%(Ø,1)=DD%(Ø,1)-DD%(Ø,2)
11330 FOR M=1 TO 16
11340 DD%(M,1)=DD%(M,2)-DD%(M,1)
11350 NEXT M
11360 IF DD%(0,2)>DD%(0,4) THEN 11390
11370 DD%(0,2)=DD%(0,2)+(32767-DD%(0,4))
1138Ø GOTO 114ØØ
11390 DD%(0,2)=DD%(0,2)-DD%(0,4)
11400 FOR M=1 TO 16
11410 DD%(M,2)=DD%(M,4)-DD%(M,3)
1142Ø NEXT M
1144Ø J=5
1146Ø N=3
11500 FOR M=0 TO 16
11520 DD%(M,N)=0
1154Ø NEXT M
11600 IF DD%(0,J+1)=0 AND DD%(0,J+3)=0 THEN 12300
11620 IF DD%(0,J-1)>DD%(0,J+1) THEN 11700
11640 DD%(0,N)=DD%(0,J-1)+(32767-DD%(0,J+1))+DD%(0,N)
1168Ø GOTO 118ØØ
11700 DD\%(0,N) = DD\%(0,J-1) - DD\%(0,J+1) + DD\%(0,N)
11800 FOR M=1 TO 16
11820 DD% (M,N) = DD% (M, J+1) - DD% (M, J) + DD% (M,N)
11840 IF DD%(M,N)>=30700 THEN 12600
1186Ø NEXT M
12000 IF DD%(0,N)>=XX% THEN 12100
12020 J=J+2
```

12040 GOTO 11600 12100 N=N+1 12120 J=J+2 1214Ø GOTO 115ØØ 12200 REM PRINT OUT MESSAGE TO IMPLY RAW DATA NOT RIGHT 12220 PRINT"** SOMETHING WRONG ABOUT RAW DATA" 1224Ø PRINT 12260 GOSUB 8000:REM BEEP 1228Ø GOTO 127ØØ 12300 REM CALCULATE THE ACTUAL TOTAL COVERED TIME 12320 NC=N-1:REM LAST SET DO NOT HAVE INT. DATA 1234Ø TT=Ø 12360 FOR Z=1 TO NC 1238Ø TT=TT+DD%(Ø,Z) 12400 NEXT Z 12420 TT=TT*U2/(1.023E6):REM IN UNIT OF SEC 12440 PRINT"** TOTAL COVERED TIME IS";TT;"SEC" 12460 PRINT" ";NC;"SETS OF DATA ARE OBTAINED !" 1248Ø PRINT 12500 N=NC 12520 GOTO 12700 12600 REM TIME OF RESOLUTION NOT RIGHT 12620 PRINT"** TR IS TOO LONG & HAS TO BE SHORTER !" 1264Ø PRINT 12700 RETURN 13000 REM SET STANDBY POTENTIAL 13100 PRINT"STANDBY VOLT ON BOTH DAC :";SV; "VOLT" 13120 INPUT"IS IT OK(Y/N)";DM\$ 1314Ø PRINT 13160 IF DM\$="Y" THEN 13300 13200 INPUT"STANDBY VOLT (+1.0) "; SV 1322Ø FRINT 1324Ø GOTO 131ØØ 13300 REM CONVERT ANALOG POTENTIALS TO DIGITAL 13320 DM%=INT(SV*4095/6.012+2042) 13340 SH%=INT(DM%/256) 13360 SL%=INT (DM%-SH%*256) 13400 RETURN 13500 REM APPLY STANDBY POTENTIAL 13600 FOKE L1,SL%:POKE H1,SH% 13620 POKE L2, SL%: POKE H2, SH% 13640 PRINT"-- STANDBY POTENTIAL JUST APPLIED" 13660 PRINT" TO BOTH DAC !!" 1368Ø FRINT 13800 RETURN 14000 REM SET EXPT. CONDITIONS 14100 PRINT"SAMPLE LABEL : ";LB\$ 1412Ø PRINT 14140 PRINT" !! CHECK PUMP SCALE !!" 14160 PRINT" FLOW RATE(FR\$) : ";FR\$ 1418Ø PRINT 14200 FRINT"INIT. VOLT ON DAC #1 :"; IV; "VOLT" 14220 PRINT"FIN. VOLT ON DAC #2 :"; FV; "VOLT" 1424Ø FRINT 14260 INFUT"ARE ALL ABOVE OK(Y/N)"; DM\$ 1428Ø PRINT 14300 IF DM\$="Y" THEN 14700 14500 INPUT"SAMPLE LABEL (CONC, SPECIES) ";LB\$ 1452Ø PRINT 14540 PRINT"* CHECK PUMP SCALE (60.0)" 14560 INFUT" FR\$ (ON SCALE 60.0)"; FR\$ 1458Ø PRINT 14600 INPUT VOLT ON DAC #1(+1.05/+1.95 V)"; IV 14620 INPUT VOLT ON DAC #2(+1.95/+1.95 V) "; FV 1464Ø FRINT 14660 GOTO 14100

```
14700 REM CONVERT ANALOG POTENTIALS TO DIGITAL
14720 DM%=INT(IV*4095/6.011+2037)
14740 IH%=INT (DM%/256)
14760 IL%=INT(DM%-IH%*256)
14800 DM%=INT (FV*4095/6.014+2047)
14820 FH%=INT(DM%/256)
14840 FL%=INT (DM%-FH%*256)
14900 RETURN
15000 REM SET TIME PARAMETERS FOR DATA COLLECTING
15010 PRINT"U1 CLOCK TIME = U1*(1/60) SEC"
15020 PRINT" U1 VALUE :";U1
15030 PRINT"U2 CLOCK PERIOD(CP) =";CP;"SEC"
15040 PRINT" LEAST U2 BYTE =";U2
15050 PRINT"SET-UP OF COVERED TIME =";CT;"SEC"
15060 PRINT" (GREATER THAN 60 SEC BE OK)"
15070 PRINT" COUNTER HI-BYTE(CH) =";CH
15080 PRINT" COUNTER LO-BYTE(CL) =";CL
15090 PRINT
15100 INPUT ARE ALL ABOVE OK (Y/N) "; DM$
1512Ø PRINT
15140 IF DM$="Y" THEN 15700
15200 PRINT"U1 CLOCK TIME = U1*(1/60) SEC"
15220 INPUT" U1 VALUE WANTED (<=60) "; U1
15240 PRINT
15300 PRINT"LEAST UZ BYTE DEPENDS ON U1 VALUE (64= 2 S)"
15320 INPUT" LEAST U2 BYTE(>=64)";U2
15360 CP=(127*256+255)*U2/(1.023*10^6)
15380 PRINT" CLOCK PERIOD(CP) =";CP;" SEC"
15400 PRINT
15500 PRINT"TOTAL COVERED TIME = (CH*256+CL)*CP SEC"
15520 INPUT" COUNTER HIBYTE (=0)"; CH
15540 INPUT" COUNTER LOBYTE (=30)"; CL
1558Ø CT=(CH*256+CL)*CP
15600 PRINT" SET-UP TOTAL COVERED TIME =";CT;"SEC"
15610 PRINT" (AROUND 60 SEC BE OK)"
15620 PRINT
1564Ø GOTO 15000
15700 POKE 49905,U1:REM $C2F1=U1BLD
15720 POKE 49906,U2:REM $C2F2=U2ALD
15740 POKE 49907, CL:REM $C2F3=CNTLO
15760 POKE 49908.CH: REM $C2F4=CNTHI
15800 RETURN
16660 GOSUB 8000:REM BEEF
20000 REM PRINT OUT ORIGINAL DATA
20100 DPEN4,4:CMD4
20200 PRINT" SAMPLE LABEL : ";LB$
20220 PRINT
20240 PRINT" FLOW RATE : ";FR$
20260 PRINT
20280 FRINT" POTENTIAL ON DAC#1 :"; IV; " VOLT"
20300 PRINT" POTENTIAL ON DAC#2 :";FV;" VOLT"
20320 FRINT
20340 FRINT"
               U1 VALUE :":U1
20360 PRINT" LEAST U2 BYTE =";U2
20380 PRINT" U2 CLOCK PERIOD(CP) =";CP;" SEC"
20400 PRINT" SET-UP TOTAL COVERED TIME =";CT; "SEC"
20420 PRINT
20440 PRINT"
                 AFTER ELECTRODES REACTIVATED,"
20450 PRINT"
                   PROGRAM STARTED TO COLLECT DATA"; TD; "SEC LATER !!"
20460 PRINT
20480 FRINT#4:CLOSE4
20500 PRINT"** Z MUST <=";N;" % = MULTIFLE OF 5"
20520 PRINT" IF Z<0 THEN END OF PRINTING "
20540 INPUT"
                  NEW STARTING SET #";Z
20560 PRINT
20580 IF Z<Ø THEN 21000
```

```
20600 OPEN4,4:CMD4
20700 PRINT
20720 PRINT
20730 PRINT
20730 PRINT
20740 PRINT" ** Z =";Z
20740 PRINT" M DD%Z DD%Z+1 DD%Z+2 DD%Z+3 DD%Z+4"
20760 FOR M=0 TO 16
20780 PRINT M;DD%(M,Z);DD%(M,Z+1);DD%(M,Z+2);DD%(M,Z+3);DD%(M,Z+4)
20820 PRINT M;DD%(M,Z);DD%(M,Z+1);DD%(M,Z+2);DD%(M,Z+3);DD%(M,Z+4)
20820 PRINT M
20820 PRINT
20840 PRINT#4:CLOSE4
20900 GOTO 20500
21000 RETURN
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READY.

APPENDIX D

SOFTWARE "DATACQ"

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The "DATACQ" subroutine, shown at the end of this section, is used to perform data acquisition in conjunction with the 12-bit data acquisition circuitry and is written in assembly language to increase speed of data acquisition and to reduce memory requirements of the software. The flow chart of this subroutine is shown in Figure 27.

The subroutine begins with masking all of the high level interrupts of the C-64, followed by setting up the PIA, Timer #1, Timer #2, and Cycle Counter (CC) of Timer #2. Both timers are stopped prior to their setup. The initial setup on Timer #2 is recorded and stored. Then, starting with the first electrode, the applied potential on each electrode is converted to a digital value while the switch of the current follower for that electrode is closed, followed by storing the digital value in the RAM of the C-The same processes are applied to the next electrode 64. and so on until all 16 electrodes are completed. Both timers are then started and the time on Timer #2 is immediately recorded and stored. Starting with the first electrode again, the switch is opened and the converted integrating current at the electrode is recorded and stored The same procedures continued until all 16 electrodes also. are processed.

The overflow bit of Timer #2 is then Checked. If the bit is set, the cycle counter is decreased by one. After that, the overflow bit of Timer #1 is checked. Otherwise,



Figure 27. Flow chart of software "DATACQ".

the overflow bit of Timer #1 is checked directly. If the overflow bit of Timer #1 is set, the time on Timer #2 at that moment is stored in the RAM of the C-64, followed by the clearing of that bit and the restarting of Timer #1. Next, beginning with the first electrode, the analog signal generated by the electrode is converted to a digital value and stored in the RAM of the C-64. The reset switch is closed to discharge the capacitor of the current follower of this electrode and after 30 microseconds the switch is opened again. The offset value at this integrator is obtained after opening the switch of the integrator 30 microseconds. The same procedures are applied to the next electrode until all 16 electrodes are serviced.

If the overflow bit of Timer #1 is not set, the digital values converted from the integrating current on each electrode are checked, one by one, according to their order. If any of digital values obtained on any electrode is over the preset threshold value, i.e., close to saturation, a flag is set. Then, the time at this moment on Timer #2 is stored in the RAM. Starting with the first electrode again, the present integrated current is converted to the digital value and stored in the RAM of the C-64, followed by the immediate discharge of the capacitor of the current follower of this electrode. Then, the switch is re-opened. The offset value of each integrator is obtained as previously described. The same procedures are applied to the next

electrode until all 16 electrodes are serviced. After that, the RAM of the C-64 is checked to determine whether enough memory is available. If not enough, the subroutine will stop collecting data and clear all high level interrupt mask. Otherwise, the cycle is repeated, as shown in Figure 27.

If neither the overflow bit of Timer #1 is set nor any of the digital value readouts is over the preset threshold value, the same procedures are repeated, i.e., checking whether the overflow bits of Timer #2 and #1 are set, converting the integrated analog signals to digital values, and checking whether any of the digital values was over the threshold value, etc., until the preset time period is over.

Different data acquisition rates may be accommodated by changing the setup of both timers via the "COLLECT.DATA" program. The fastest sampling rate for this data acquisition system in this subroutine is approximately 5.7 ms/datum on each electrode.

Obviously, a discontinuity of integrating analyte current is created while resetting the integrator. The reset process should be as quick as possible. However, a current surge, which decays with time, is also observed after the switch of the integrator re-opens and which almost disappears within 30 microsecond. Hence, there is a short holding time of 30 microseconds before collecting the offset value of the integrator.

LINE#	LOC	CODE	LINE
aaaa1	aaaa		ARC SURBOUTINE "DATACO"
66667	66666		FOR DISCEPTATION
00002	8888		•2/21/89
66664	aaaa		,2/21/0/
00005	aaaa		*=\$C000
00006	CØØØ		AD=\$DEØØ
00007	CØØØ		AC=\$DEØ1
ØØØØ8	CØØØ		BD=\$DEØ2
00007	CØØØ		BC=\$DEØ3
ØØØ1Ø	CØØØ		LATCH = $DE60$
00011	CØØØ		SET = \$DE8Ø
ØØØ12	CØØØ		SAVX = \$FD
00013	CØØØ		SAVY = \$FE
ØØØ14	CØØØ		FLAG = \$C2FØ
ØØØ15	CØØØ		U1BL0 = \$C2F1
ØØØ16	CØØØ		U2ALO = \$C2F2
00017	CØØØ		CNTLD = \$C2F3
ØØØ18	CØØØ		CNTHI = \$C2F4
ØØØ19	Cøøø		;
00020	CØØØ		SET-UP OF ADC FIA REGISTERS
ØØØ21	CØØØ		;FIRST, PUTS S&H IN HOLD STATE
ØØØ22	CØØØ		; (NOTE !! MUST BE IN SAMPLE STATE
00023	CØØØ		; BEFORE ENTRY)
00024	CØØØ		WILL LEAVE S&H IN SAMPLE STATE
00025	CØØØ	A7 00	ASETUP LDA #Ø
00026	CØØ2	8D Ø1 DE	STA AC
0002r	0005	8D 03 DE	STA BC
00028	C008	8D 00 DE	STA AD
00029	COOB	A7 FØ A7 FØ	LDA #\$FØ
00030	Cala	80 102 DE	
00031	C010	H7	EDA #POL
00032	COLZ	80 91 DE	STA DC
00033	CØ15	60 03 DE	
00035	C/219	0.9	
000000	CØ19		ONE ADD CONVERSION
00037	CØ19		LOW DATA IN REG-X & HI DATA IN REG-A
00038	CØ19	A9 34	CNVRT LDA #\$34
00039	CØ1B	BD Ø1 DE	STA AC
00040	CØ1E	8D Ø3 DE	STA BC
00041	CØ21	A9 3C	LDA #\$3C
00042	CØ23	8D Ø3 DE	STA BC
00043	CØ26	AØ Ø6	LDY #6
00044	CØ28	88	LPADC DEY
00045	CØ29	DØ FD	BNE LFADC
ØØØ46	CØ2B	AD Ø2 DE	LDA BD
ØØØ47	CØ2E	29 ØF	AND #\$ØF
ØØØ48	CØ3Ø	AE ØØ DE	LDX AD
ØØØ49	CØ33	AØ 3C	LDY #\$3C
90050	CØ35	8C Ø1 DE	STY AC
ØØØ51	CØ38	60	RTS
00052	CØ39		;

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DATACO.....PAGE ØØØ1

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MAIN ROUTINE.....PAGE 0002 LINE# LOC CODE LINE

00054	CØ39		;	
00055	CØ39		; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ;	
00056	CØ39		;# #	
ØØØ57	CØ39		;# MAIN DATA ACQ RTN #	
ØØØ58	CØ39		;# #	
ØØØ57	CØ39		; ********	
ଡ଼ଡ଼ଡ଼ୄଌଡ଼	CØ39		;	
ØØØ61	CØ39		ARYTAB = \$2F	
00062	CØ39		ARPTR = \$FB	
ØØØ63	CØ39	78	SEI	
ØØØ64	CØ3A	A5 2F	DATACO LDA ARYTAB	
00065	CØ3C	18	CLC	
ØØØ66	CØ3D	69 Ø9	ADC #9	
ØØØ67	CØ3F	85 FB	STA ARPTR	
00068	CØ41	AS 30	LDA ARYTAB+1	
ØØØ69	CØ43	69 ØØ	ADC #Ø	
ØØØ7Ø	CØ45	85 FC	STA ARPTR+1	
ØØØ71	CØ47	2Ø CC CØ	JSR SETU2	
ØØØ72	CØ4A	20 C0 C0	JSR SETU1; SET TIMER U	1
ØØØ73	CØ4D	20 E2 CØ	JSR STPCLK	
00074	CØSØ	20 B7 CØ	JSR MSKINT	
ØØØ75	CØ53	2Ø 46 C1	JSR R5000	
00076	CØ56	AD ØD DD	LDA \$DDØD	
ØØØ77	CØ59	A9 55	LDA #\$55	
ØØØ78	CØ5B	8D ØF DD	STA \$DDØF	
ØØØ79	CØSE	8D ØE DD	STA \$DDØE	
ØØØ8Ø	CØ61	20 F3 C0	JSR STRCLK	
ØØØ81	CØ64	2Ø 92 C1	JSR R6000	

00083	CØ67	A9 Ø2		CHKU2	LDA	#%ØØØØØØ1Ø
ØØØ84	CØ69	2C ØD	DD		BIT	\$DDØD
ØØØ85	CØ4C	FØ 18			BEQ	CHKU1
ØØØ86	CØ6E	38			SEC	
ØØØ87	CØ6F	AD F3	C2		LDA	CNTLO
ØØØ88	CØ72	E9 Ø1			SBC	#1
ØØØ87	CØ74	8D F3	C2		STA	CNTLO
ØØØ 7 Ø	CØ77	AD F4	C2		LDA	CNTHI
00071	CØ7A	E9 ØØ			SBC	#Ø
ØØØ92	CØ7C	8D F4	C2		STA	CNTHI
ØØ Ø 93	CØ7F	DØ Ø5			BNE	CHKU1
ØØØ94	CØ81	AD F3	C2		LDA	CNTLO
00075	CØ84	FØ 25			BEQ	CLRMSK
00096	CØ86	A9 Ø2		CHKU1	I DA	#%000000010
00077	CØ88	2C ØD	DC		BIT	SDCOD
00078	CØ8B	DØ ØF			BNE	SAVEST
00077	CØBD	A9 00				#Ø
00100	CØ8F	8D FØ	62		STA	FLAG
00101	CØ92	20 E3	C1		JSR	87000
00102	CØ95	AD FØ	C2		I DA	FLAG
00103	CØ98	C9 Ø1			CMP	#1
00104	CØ9A	DØ FA			BNE	CHKUU
00105	CØ9C	20 53	CØ	SAVEST	JSR	STRCLK
00106	CØ9F	20 36	C2	5	JSR	R9000
00107	CØA2	20 15	C1		JSR	MOVETE
00108	CØAS	A9 90			I DA	#\$90
00107	CØA7	C5 FC			CMP	ABPTR+1
00110	CØA9	10 BC			BPI	CHKU2
00111	CØAB	58		CL RMSK	CLT	0
00112	CØAC	A9 7E		02/11/0/	I DA	#\$7F
00113	CØAE	8D ØD	DC		STA	\$DCØD
00114	CØB1	A9 7F			I DA	#\$7F
00115	CØB3	BD ØD	ממ		STA	\$DDØD
00116	CØBA	60	~-		RTS	+0000
ØØ117	CØB7	A9 FF		MSKINT	LDA	#\$FF
00118	CØB9	8D ØD	DC		STA	\$DCØD
00119	CØBC	8D ØD	DD		STA	\$DDØD
00120	CØBE	60			RTS	+2202
00121	CØCØ	A7 ØØ		SETUI		#Ø
00122	CØC2	8D Ø7	DC		STA	\$DC07
00123	CØCS	AD F1	62		I DA	
00124	CØCB	8D Ø6	DC		STA	\$DCØ6
00125	CØCB	60	~ -		RTS	+2020
00126	CØCC	A9 7E		SETU2	IDA	#127
00127	CØCE	8D Ø7	ממ		STA	\$0007
00128	CØD1	A9 FF	~~		LDA	#255
00129	CØD3	8D Ø6	DD		STA	\$DDØ6
00130	CØD6	A9 ØØ			LDA	#Ø
00131	CØDB	8D Ø5	DD		STA	\$DDØ5
00132	CØDB	AD F2	C2		LDA	UZALO
00133	CØDE	8D Ø4	DD		STA	\$DDØ4
00134	CØE1	6Ø			RTS	
00135	CØE2	A7 54		STPCLK	LDA	#\$54
00136	CØE4	8D ØF	DC		STA	\$DCØF
ØØ137	CØE7	8D ØF	DD		STA	\$DDØF

,

MAIN ROUTINE.....PAGE ØØØ3 CODE

LINE

LINE# LOC

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MAIN ROUTINE.....PAGE ØØØ4

LINE#	LOC	CODE		LINE		
00138	CØEA	A9 54			LDA	#\$54
00139	CØEC	BD ØE	DC		STA	\$DCØE
00140	CØEE	BD ØE	DD		STA	\$DDØE
00141	CØF2	60			RTS	
00142	CØES	AD ØD	DC	STRCLK	LDA	\$DCØD
00143	CØEA	49 55		01110211	LDA	#\$55
00144	CØFR	BD ØF	DC		STA	\$DCØF
00145	COFR	SD ØF	DC		STA	\$DCØE
00144	CØFF	60	20		RTS	
66147	COFF	<u>0</u> 0 00		SAVINE	IDY	#0
00149	C101	91 FB		2	STA	(ARPTR) Y
00149	C103	80			TXA	
00150	C104	6.1			TNY	
00151	C105	91 FB			STA	(ARPTR) Y
00152	C107	AS FR			I DA	ARPTR
00153	C109	18				
00154	C100	69 02			ADC	#2
00155	CIØC	85 FB			STA	ARPTR
00154	CLOE	45 FC			I DA	ARPTR+1
00157	C110	69 00			ADC	#Ø
00158	C112	85 FC			STA	ARPTR+1
00159	C114	60			RTS	
00160	C115	AS FB		MOVPTR	LDA	ARPTR
00161	C117	18			CLC	
00162	C118	69 22			ADC	#34
00163	C11A	85 FB			STA	ARPTR
ØØ164	CIIC	A5 FC			LDA	ARPTR+1
ØØ165	C11E	67 ØØ			ADC	#Ø
ØØ166	C12Ø	85 FC			STA	ARPTR+1
ØØ167	C122	6Ø			RTS	
ØØ168	C123	A6 FD		RSTLCH	LDX	SAVX
ØØ169	C125	CA			DEX	
ØØ17Ø	C126	8E 6Ø	DE		STX	LATCH
ØØ171	C129	2Ø 3C	C1		JSR	DELAY
ØØ172	C12C	EB			INX	
ØØ173	C12D	8E 6Ø	DE		STX	LATCH
ØØ174	C13Ø	20 19	CØ		JSR	CNVRT
ØØ175	C133	AØ 22			LDY	#34
ØØ176	C135	91 FB			STA	(ARPTR),Y
ØØ177	C137	8A			TXA	
ØØ178	C138	C8			INY	
ØØ179	C139	91 FB			STA	(ARFTR),Y
ØØ18Ø	C13B	60			RTS	
ØØ181	C13C	EA		DELAY	NOP	
ØØ182	C13D	EA			NOP	
ØØ183	C13E	EA			NOP	
00184	C13F	EA			NUP	
00185	C140	EA			NOP	
00186	C141	EA			NOP	
00187	C142	EA			NUP	
00188	C143	EA			NOP	
00189	C144	EA			NOP	
60190	C145	619			RIS	

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R5000- GET REAL TIME.....PAGE 0005

LINE#	LOC	CODE	LINE	
ØØ192	C146	AD Ø7 D	D R5ØØØ	LDA \$DDØ7
aa104			a	JCD COUINE
66195		A9 90	6	
aa196	C151	80 02 0	F	STA BD
00197	C154	A7 ØØ		LDA #Ø
00198	C156	85 FD		STA SAVX
00199	C158	A7 ØØ		LDA #Ø
ØØ2ØØ	C15A	85 FE		STA SAVY
ØØ2Ø1	C15C	2Ø 7Ø C	1	JSR GETPOT
ØØ2Ø2	C15F	A7 9Ø		LDA #\$90
ØØ2Ø3	C161	8D Ø2 D	Ε	STA BD
ØØ2Ø4	C164	A7 ØØ		LDA #Ø
ØØ2Ø5	C166	85 FD		STA SAVX
ØØ2Ø6	C168	A7 1Ø		LDA #\$10
ØØ2Ø7	C16A	85 FE		STA SAVY
ØØ2Ø8	C16C	20 70 C	1	JSR GETPOT
ØØ2Ø9	C16F	60		RTS
ØØ21Ø	C17Ø	A6 FE	GETPOT	LDX SAVY
ØØ211	C172	8E 8Ø D	E	STX SET
ØØ212	C175	EB		INX
ØØ213	C176	E8		INX
ØØ214	C177	86 FE		STX SAVY
ØØ215	C179	A6 FD		LDX SAVX
ØØ216	C17B	8E 60 D	E	STX LATCH
ØØ217	C17E	20 3C C	1	JSR DELAY
ØØ218	C181	2Ø 19 C	Ø	JSR CNVRT
ØØ219	C184	20 FF C	Ø	JSR SAVINF
ØØ22Ø	C187	A6 FD		LDX SAVX
ØØ221	C187	E8		INX
ØØ222	C18A	E8		INX
ØØ223	C18B	86 FD		STX SAVX
ØØ224	C18D	EØ 1Ø		CPX #\$1Ø
ØØ225	C18F	DØ DF		BNE GETPOT
ØØ226	C191	6Ø		RTS

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R6000- OPEN SWITCH &.....PAGE 0006

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LINE#	LOC	CODE		LINE		
ØØ228	C192	AD Ø7	DD	R6ØØØ	LDA	\$DDØ7
ØØ229	C195	AE Ø6	DD		LDX	SDDØA
ØØ23Ø	C178	20 FF	сø		JSR	SAVINE
ØØ231	C17B	A7 8Ø			LDA	#\$80
ØØ232	C19D	8D Ø2	DE		STA	BD
ØØ233	CIAØ	A7 ØØ			LDA	#Ø
ØØ234	C1A2	85 FD			STA	SAVX
ØØ235	C1A4	A7 ØØ			LDA	#Ø
ØØ236	C1A6	85 FE			STA	SAVY
ØØ237	C1A8	2Ø BC	C1		JSR	GETOFF
ØØ238	CIAB	A7 90			LDA	#\$9Ø
ØØ239	C1AD	8D Ø2	DE		STA	BD
ØØ24Ø	C1BØ	A7 ØØ			LDA	#Ø
ØØ241	C1B2	85 FD			STA	SAVX
ØØ242	C1B4	A7 1Ø			LDA	#\$1Ø
ØØ243	C1B6	85 FE			STA	SAVY
ØØ244	C1B8	2Ø BC	C1		JSR.	GETOFF
ØØ245	C1BB	60			RTS	
ØØ246	CIBC	A6 FE		GETOFF	LDX	SAVY
ØØ247	C1BE	8E 8Ø	DE		STX	SET
ØØ248	C1C1	E8			INX	
ØØ249	C1C2	E8 .			ΙΝΧ	
ØØ25Ø	C1C3	86 FE			STX	SAVY
ØØ251	C1C5	A6 FD			LDX	SAVX
ØØ252	C1C7	8E 6Ø	DE		STX	LATCH
ØØ253	C1CA	2Ø 3C	C1		JSR	DELAY
ØØ254	CICD	E8			INX	
ØØ255	CICE	86 FD			STX	SAVX
ØØ256	CIDØ	8E 6Ø	DE		STX	LATCH
ØØ257	C1D3	20 19	CØ		JSR	CNVRT
ØØ258	C1D6	2Ø FF	CØ		JSR	SAVINF
ØØ259	C1D9	A6 FD			LDX	SAVX
00260	CIDB	E8			INX	
00261	CIDC	86 FD			STX	SAVX
00262	CIDE	EØ 1Ø			CPX	#\$1Ø
00263	C1EØ	DØ DA			BNE	GETOFF
ØØ264	C1E2	60			RTS	

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R7000- CHECK DATA IF.....PAGE 0007

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LINE#	LOC	CODE		LINE		
ØØ266	C1E3	A9 8Ø		R7ØØØ		#\$8Ø
ØØ267	CIES	8D Ø2	DE		STA	BD
00268	C1E8	A7 Ø1			LDA	#1
00269	C1EA	85 FD			STA	SAVX
ØØ27Ø	C1EC	A7 ØØ			LDA	#Ø
ØØ271	C1EE	85 FE			STA	SAVY
ØØ272	C1FØ	2Ø ØB	C2		JSR	CMPVAL
ØØ273	C1F3	AD FØ	C2		LDA	FLAG
ØØ274	C1F6	C9 Ø1			CMP	#1
ØØ275	C1F8	FØ 10			BEQ	RT7ØØØ
ØØ276	C1FA	A7 7Ø			LDA	#\$9Ø
ØØ277	C1FC	8D Ø2	DE		STA	BD
ØØ278	C1FF	A7 Ø1			LDA	#\$Ø1
ØØ279	C2Ø1	85 FD			STA	SAVX
ØØ28Ø	C2Ø3	A9 1Ø			LDA	#\$1Ø
ØØ281	C2Ø5	85 FE			STA	SAVY
ØØ282	C2Ø7	2Ø ØB	C2		JSR	CMPVAL
ØØ283	C2ØA	60		RT7ØØØ	RTS	
ØØ284	C2ØB	A6 FE		CMPVAL	LDX	SAVY
ØØ285	C2ØD	8E 8Ø	DE		STX	SET
ØØ286	C21Ø	E8			INX	
ØØ287	C211	E8 .			INX	
ØØ288	C212	86 FE			STX	SAVY
ØØ287	C214	A6 FD			LDX	SAVX
ØØ29Ø	C216	8E 6Ø	DE		STX	LATCH
ØØ291	C219	20 19	CØ		JSR	CNVRT
ØØ292	C21C	C9 ØF		CHKHI	CMP	#\$ØF
ØØ293	C21E	FØ 1Ø			BEQ	SETFLG
ØØ294	C22Ø	C9 ØØ		CHKLO	CMP	#\$ØØ
ØØ295	C222	FØ ØC			BEQ	SETFLG
ØØ296	C224	A6 FD			LDX	SAVX
ØØ297	C226	E8			INX	
ØØ298	C227	E8			INX	
ØØ299	C228	86 FD			STX	SAVX
ØØ3ØØ	C22A	EØ 11			CPX	#\$11
00301	C22C	DØ DD			BNE	CMPVAL
ØØ3Ø2	C22E	FØ Ø5			BEQ	RTCMPV
00303	C23Ø	A9 Ø1		SETFLG	LDA	#1
ØØ3Ø4	C232	8D FØ	C2		STA	FLAG
ØØ3Ø5	C235	60		RTCMPV	RTS	

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R9000- GET TIME & EN.....PAGE 0008

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LINE#	LOC	CODE		LINE		
ØØ3Ø7	C236	AD Ø7	DD	R9ØØØ	LDA	\$DDØ7
ØØ3Ø8	C239	AE Ø6	DD		LDX	\$DDØ6
ØØ3Ø7	C23C	2Ø FF	CØ		JSR	SAVINF ~
ØØ31Ø	C23F	A9 8Ø			LDA	#\$8Ø
ØØ311	C241	8D Ø2	DE		STA	BD
ØØ312	C244	A7 Ø1			LDA	#1
ØØ313	C246	85 FD			STA	SAVX
ØØ314	C248	A7 ØØ			LDA	#Ø
ØØ315	C24A	85 FE			STA	SAVY
ØØ316	C24C	2ø 6ø	C2		JSR	SAVVAL
ØØ317	C24F	A7 7Ø			LDA	#\$90
ØØ318	C251	8D Ø2	DE		STA	BD
ØØ319	C254	A7 Ø1			LDA	#\$Ø1
ØØ32Ø	C256	85 FD			STA	SAVX
ØØ321	C258	A7 1Ø			LDA	#\$1Ø
ØØ322	C25A	85 FE			STA	SAVY
ØØ323	C25C	2Ø 6Ø	C2		JSR	SAVVAL
ØØ324	C25F	ЬØ			RTS	
ØØ325	C26Ø	A6 FE		SAVVAL	LDX	SAVY
ØØ326	C262	8E 8Ø	DE		STX	SET
ØØ327	C265	E8			INX	
ØØ328	C266	EB			INX	
ØØ327	C267	86 FE			STX	SAVY
ØØ33Ø	C269	A6 FD			LDX	SAVX
ØØ331	C26B	8E 6Ø	DE		STX	LATCH
ØØ332	C26E	20 19	CØ		JSR	CNVRT
ØØ 33 3	C271	- AØ ØØ			LDY	#Ø
ØØ334	C273	71 FB			STA	(ARPTR) .Y
ØØ335	C275	8A			ТХА	
ØØ336	C276	C8			INY	
00337	C277	71 FB			STA	(ARPTR) Y
ØØ338	C279	20 23	C1		JSR	RSTLCH
00339	C27C	A5 FB			LDA	ARPTR
00340	C27E	18			CLC	
00341	C27F	69 Ø2			ADC	#2
00342	C281	85 FB			STA	ARPTR
00343	C283	A5 FC			I DA	ARPTR+1
ØØ344	C285	67 00			ADC	#Ø
00345	C287	85 FC			STA	APPTR+1
00346	C289	A6 FD			LDX	SAVX
00347	C288	EB			INX	
ØØ348	C28C	EB			INX	
00349	C28D	86 FD			STX	SAVX
00350	C28F	EØ 11			CPX	#\$11
ØØ351	C291	DØ CD			BNE	SAVVAL
ØØ352	C293	60			RTS	
00353	C294			- END		

ERRORS = ØØØØØ

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APPENDIX E

SOFTWARE "NORMALIZE

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The program "NORMALIZE", shown at the end of this section, is used to obtain the normalization factors of all CFAEs in the ME cell. The flow chart of this software is shown in Figure 28.

The program first transfers the coulometric data, which has been saved on the floppy disk for the normalization factors of CFAEs, back to the RAM of the C-64. Then the first set of digital values are converted into the potentials which represent the applied potentials on each CFAEs. The rest of data are converted into the amperometric data. After screening all amperometric data, a band of data at each CFAE obtained prior to the injection of the sample solution is chosen to represent the residual current generated by the supporting electrolyte solution. Also, a band of highest data is chosen to represent the current which is generated by the sample solution. The representative sample solution currents and the residual currents within the resolution time period at each CFAE are obtained through the average of these two band of data, respectively. The net current on each electrode, which is generated by the sample species only, is obtained by subtracting the residual current from the current generated by the sample solution. Then one averaged net current of a CFAE is chosen. All net current values are divided by this chosen value to get the normalization factors for all 16 CFAEs in the ME cell. Finally, these factors are saved onto



Figure 28. Flow chart of software "NORMALIZE".

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a disk for normalization of the data for pseudo-voltammetry.

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11 REM * 2/12/1989 20 REM * 'NORMALIZE' * 21 REM * 30 REM * USE CALC. CAPACITANCE TO 31 REM * NORM. ELECTRODE AREA BY PEAK AREA OR HEIGHT. 32 REM * 40 REM * THEN, NORM. AREA IS SAVED 41 REM * IN A FILE (N-A/H-*) 60 REM DAS= DATE OF RUNNING EXPERIMENT 62 REM LB#= SAMPLE LABEL (SPECIES & CONC.) 64 REM FR\$= FLOW RATE (PUMP SCALE) 66 REM IV = INITIAL VOLTAGE ON DAC 7548 #1 68 REM FV = FINAL VOLTAGE ON DAC 7548 #2 70 REM TR = TIME RESOLUTION (TR SEC) 71 REM U1 = U1BLO 72 REM U2 = U2ALO 73 REM TT = ACTUAL TOTAL COVERED TIME 74 REM NT = TOTAL DATA SET # SAVED FROM EXPT. 75 REM RM\$= EXPT. REMARK 76 REM NR = RESIDUAL CURRENT SET'S # 78 REM NS = SAMPLE CURRENT SET'S # 80 REM RS(M,N)=RESULTS FROM CALCULATION 82 REM RS(Ø,N)=INTEGRATION TIME 84 REM RS(M,Ø)=REAL APPLIED VOLTAGE ON EVERY ELECTRODE 86 REM RC (M) = AVE. RESIDUAL CURRENT 88 REM SC(M, SS)=PEAK AREA/HEIGHT FOR SAMP.CURRENT 90 REM TP (N) =TIME AT WHICH DATA COLLECTED 92 REM NA(M) =NORM.AREA OF ELECTRODE M 94 REM CA(M)=CAPACITANCE OF ELECTRODE M 100 DIM RS(16,200), RC(16), SC(16,6), NA(16), TP(200) 120 DIM MN(16),SD(16),RD(16),R(16),CA(16),SF(16),SL(16) 140 ID\$="'NONE'":REM DATA FILE ID 160 PRINT 200 PRINT"** MAKE SURE (1) LINE 10720 CORRECT," 220 PRINT" (2) RIGHT DISK IN DRIVE," 24Ø PRINT" & (3) BEEP IS HEARD !" 260 PRINT 280 GOSUB 14000:REM BEEP 300 SS=0:REM SET # FOR REPEATABILITY 400 SS=SS+1 500 INPUT"KNOW AREA DATA FILE ID(Y/N)";DM\$ 520 PRINT 540 IF DM\$<>"N" THEN 700 580 PRINT"* TYPE '@\$' & LOOK FILE *" 600 PRINT"* NAME STARTS WITH 'A-' *" 620 PRINT"* THEN TYPE 'CONT' *" 660 PRINT 68Ø STOP 720 PRINT"* FOR AREA DATA FILE ID, *" 740 PRINT"* IT STARTS WITH 'A-' *" 780 PRINT 800 PRINT"PREVIOUS AREA DATA FILE ID : "; ID\$ 820 INPUT"NEW AREA DATA FILE ID (A-+)"; ID\$

```
840 INPUT"IS FILE NAME OK(Y/N)":DM$
860 PRINT
880 IF DM$<>"Y" THEN 500
900 PRINT"** IF DRIVE LIGHT BLINKS,"
910 PRINT"
           STOP PROGRAM IMMEDIATELY. THEN"
            TYPE 'CLOSE 2', FOLLOWED BY TYPING"
'GOTO 500' !!"
920 PRINT"
930 PRINT"
94Ø PRINT
950 PRINT"
            WAIT (ABOUT 5 MINUTES)"
960 PRINT"
            TO LOAD & TREAT DATA !!
                                     "
980 PRINT
1000 REM GET DATA FROM A FILE TO NORMALIZE ELECTRODE AREA
1020 OPEN 2.8.2.ID$+".S.R"
1040 INPUT#2, DA$
1060 INPUT#2,LB$
1080 INPUT#2, FR$
1100 INPUT#2, UL: REM UPPER LIMIT OF S.W.
1120 INPUT#2,LL:REM LOWER LIMIT OF S.W.
1140 INPUT#2, HZ: REM FREQ. OF S.W.
1160 INPUT#2, DU: REM DUATION OF S.W.
1180 INPUT#2, TD: REM TIME OF DELAY
1200 INPUT#2, IV
1220 INPUT#2, FV
123Ø INPUT#2,U1
124Ø INPUT#2,TR
1260 INPUT#2,U2
128Ø INPUT#2,TT
1300 INPUT#2,NT
1310 INPUT#2,RM$
1320 IF IV=FV THEN 1500
134Ø CLOSE 2
1380 FRINT"* PROGRAM IS TO NORMALIZE AREA *"
                                            *"
1400 FRINT"* LOAD WRONG DATA FILE
1440 PRINT
146Ø STOP
1500 REM TREAT DATA BEFORE PUT INTO A MATRIX
1520 CY=1.023E6:REM CY=FREQUENCY OF TIMER
1540 GOSUB 13000: REM LOAD CAPACITANCE FOR EACH INTEGRATOR
1560 REM ORIGINAL U2 SET-UP FOR 1 CYCLE
1580 INPUT#2,TM%
1600 RS(0,0)=TM%*U2/CY
1620 REM CALC. REAL AFPLIED POTENTIAL
1640 FOR M=1 TO 16
1660 INFUT#2,TM%
1680 RS(M,0)=(TM%-2049)/204.5
1700 NEXT M
1720 TP (Ø)=Ø
1740 FOR N=1 TO NT
1760 INPUT#2,TM%
1780 RS(0,N)=TM%*U2/CY:REM INTEGRATION TIME
1800 TP(N)=RS(0,N)+TP(N-1):REM TIME SCALE
1820 FOR M=1 TO 16
1840 INPUT#2,TM%
1860 RS (M,N) =TM%/204.5*CA (M) /RS (0,N)
1880 NEXT M
1900 NEXT N
1920 CLOSE 2
1740 GOSUB 14000:REM BEEP
2000 GOSUB 9900:REM CHANGE REMARK
2020 REM SHOW ALL PARAMETERS ON SCREEN
2030 PRINT"** PUSH 'G' TO GO ON,"
2040 PRINT"
            THEN HOLD 'CTRL' KEY TO VIEW"
2050 PRINT
2060 GET A$: IF A$<>"G" THEN 2060
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2070 GOSUB 9700: REM SHOW SAMPLE ID 2080 GOSUB 9000: REM SHOW EXPT. CONDITIONS 2100 INPUT ARE ALL ABOVE CORRECT (Y/N) "; DM\$ 212Ø PRINT 2140 IF DM\$="Y" THEN 2200 2160 PRINT"** MAKE CORRECTION THEN TYPE 'CONT' !!":STOP 2180 PRINT: GOTO 2000: REM RECHECK INFORMATION 2200 REM CHECK APPLIED VOLTAGE ON ELECTRODE 222Ø PRINT 2240 PRINT"ELECTRODE", "VOLT.APPLIED" 2260 PRINT 2280 FOR M=1 TO 16 2300 PRINT " # ";M,RS(M,Ø) 232Ø NEXT M 234Ø PRINT 2400 RF=0:RL=0 2420 FOR M=1 TO 16 2440 SF(M)=0:SL(M)=0 246Ø NEXT M 2500 REM CHOOSE RANGES FOR RESI:CURRENT & SAMP.CURRENT SETS 2520 BN=2:ED=21:REM START FROM SET 3 TO SET 23 2600 PRINT"** ENTER ELECTRODE #'S TO CHECK !!" 2620 PRINT" (END OF CHECK IF 1ST INPUT <=0)" 2640 PRINT 266Ø INPUT" 1ST & 2ND ELECTRODE (<=16) "; X1, X2 ARE BOTH OK(Y/N)";DM\$ 2680 INPUT" 2700 PRINT 2720 IF DM\$<>"Y" THEN 2600 2740 IF X1<=0 THEN 3500 2800 PRINT"** PUSH 'G' TO GO ON. THEN" -2820 PRINT" DETERMINE THE WAY & FIND RANGES" 284Ø FRINT" TO NORMALIZE ELECTRODE SURFACE AREA" 286Ø PRINT 2900 GET A\$: IF A\$<>"G" THEN 2900 2920 GOSUB 8800:REM FIND RANGE 3000 PRINT" 1ST ELECTRODE CHECKED :";X1 3020 PRINT" 2ND ELECTRODE CHECKED :";X2 3040 PRINT" INITIAL SET TO CHECK :";BN 3060 PRINT" END SET TO CHECK : ";ED 3080 PRINT 3100 INPUT" IS CHECK RANGE OK(Y/N)"; DM\$ 312Ø PRINT 3140 IF DM\$="Y" THEN 2600 3200 PRINT"** INPUT INITIAL & END SET TO CHECK" 3210 PRINT" END SET MUST BE SAMLLER THAN":NT 3220 PRINT 324Ø INPUT" INITIAL SET & END SET "; BN, ED 326Ø INPUT" ARE BOTH OK(Y/N)";DM\$ 3280 PRINT 3300 IF DM\$<>"Y" THEN 3200 3400 IF ED<=NT THEN 2600 3420 PRINT" ** END SET TO CHECK IS OVER RANGE **" 3440 FRINT 3460 GOTO 3200 3500 INPUT INPUT RANGE FOR RESI.CURRENT"; RF, RL 3510 INPUT"ARE BOTH OK (Y/N) "; DM\$ 3520 PRINT 3540 IF DM\$<>"Y" THEN 2600 3550 GOSUB 12000:REM CHECK RANGES SETUP 3560 INPUT"ARE ALL OK(Y/N)";DM\$ 3570 PRINT 3580 IF DM\$<>"Y" THEN 2600 3600 INPUT"PRINT OUT DATA TO PLOT (Y/N) "; DM\$ 3620 PRINT 3640 IF DM\$="N" THEN 3800 3640 INPUT"HAVE PARAMETERS PRINTED DUT(Y/N)";DM\$

3680 PRINT 3700 IF DM\$="Y" THEN 3790 3720 OPEN4,4:CMD4 3740 GOSUB 9000 3750 PRINT CHR\$(12):REM TOP OF THE NEXT PAGE 376Ø PRINT#4 3780 CLOSE4 3790 GOSUB 16000 3800 INPUT"GOT ALL RANGES(Y/N)"; DM\$ 382Ø PRINT 384Ø IF DM\$="Y" THEN 4000 3860 INPUT"CHECK THE SAME SET AGAIN(Y/N)"; DM\$ 3880 PRINT 3900 IF DM\$<>"N" THEN 2600 3920 INPUT"LOAD ANOTHER DATA SET (Y/N) "; DM\$ 3940 PRINT 3960 IF DM\$="Y" THEN 500 398Ø STOP 4000 REM GET RESI.CURRENT & TOTL.SAMP.CURRENT 4010 PRINT"WHICH WAY TO NORM. ELECTRODE AREA ?" 4020 INPUT" (1) BY PEAK AREA; (2) BY PEAK HEIGHT"; OP 4030 INPUT" IS IT OK(Y/N)"; DM\$ 4040 PRINT 4050 IF DM\$<>"Y" THEN 4010 4060 IF OP=1 THEN GOSUB 15000 4070 IF OP=1 THEN 4100 4080 IF OF=2 THEN GOSUB 15500 4100 REM SHOW AVERAGED CURRENTS ON SCREEN 4140 PRINT"ELECTRODE RESI.CURRENT* NET SAMP.CURRENT" 416Ø PRINT 4180 FOR M=1 TO 16 4200 PRINT " # ";M;RC(M);SC(M,SS) 4220 NEXT M 424Ø FRINT 4300 INPUT"PRINT OUT RESULTS (Y/N) "; DM\$ 4320 PRINT 4340 IF DM\$="N" THEN 5000 4360 GOSUB 8400: REM SURE FRINTER ONLINE 4400 INPUT "HAVE PARAMETERS PRINTED OUT (Y/N) "; DM\$ 442Ø FRINT 4440 IF DM\$="Y" THEN 4600 4460 OPEN4,4:CMD4 4480 GOSUB 9000: REM PARAMETERS 4500 PRINT CHR\$(12):REM TOP OF THE NEXT PAGE 4520 PRINT#4 454Ø CLOSE4 4600 GOSUB 9900: REM CHANGE REMARK 4700 OPEN4,4:CMD4 4720 GOSUB 9700: REM SHOW SAMPLE ID & REMARK 4740 GOSUB 17000: REM RESULTS OF CURRENTS 4760 PRINT CHR\$(12):REM GET TO THE TOP OF THE NEXT PAGE 478Ø PRINT#4:CLOSE 4 4800 INPUT"IS FRINTOUT OK(Y/N)";DM\$ 4820 FRINT 4840 IF DM\$<>"Y" THEN 4300 5000 INPUT"ARE RESULTS OK(Y/N)";DM\$ 5020 PRINT 5040 IF DM\$="Y" THEN 5400 5100 INPUT"CHECK THE SAME SET AGAIN(Y/N)"; DM\$ 512Ø PRINT 5140 IF DM\$<>"N" THEN 3000 5200 INPUT"LOAD ANOTHER SET DATA (Y/N) "; DM\$ 5220 PRINT 5240 IF DM\$="Y" THEN 500 5260 PRINT WHAT DO YOU WANT ?":STOP 5280 PRINT

5300 GOTO 5000 5400 INPUT "RESULTS FOR STATISTICS (Y/N) "; DM\$ 5420 PRINT 5440 IF DM\$="Y" THEN 5500 546Ø GOTO 51ØØ 5500 PRINT"PRESENT DATA FILE ID : "; ID\$ 5520 PRINT"PRESENT SET # FOR REPEATABILITY : ";SS 554Ø PRINT 5600 INPUT"DONE REPEATABILITY CALC. (Y/N) "; DM\$ 562Ø PRINT 5640 IF DM\$="Y" THEN 5800 5660 INPUT"MORE? ARE YOU SURE (Y/N) "; DM\$ 5680 PRINT 5700 IF DM\$="Y" THEN 400 5720 STOP 574Ø GOTO 56ØØ 5800 INPUT"DONE? ARE YOU SURE (Y/N) "; DM\$ 582Ø PRINT 5840 IF DM\$="N" THEN 5720 5900 PRINT"-- APPLY STATISTICS TO THE RESULTS NOW!" 5920 PRINT 5940 I=SS:REM TOTAL SET FOR REPEATABILITY IS 'SS' 5760 GOSUB 10000 6000 REM CALCULATE NORM. AREA FROM A SINGLE SET OF DATA 6100 PRINT"** NORMALIZE ELECTRODE SURFACE AREA !" 6120 PRINT" #";ES; "ELECTRODE AS STANDARD" 614Ø INPUT" IS IT OK (Y/N) "; DM\$ 616Ø PRINT 6180 IF DM\$="Y" THEN 6300 6200 INPUT" NEW STANDARD ELECTRODE (2) ":ES 6220 PRINT 624Ø GOTO 61ØØ 6300 BA=MN(ES): REM TAKE ES ELECTRODE AS STANDARD 6320 FOR M=1 TO 16 634Ø NA (M) = MN (M) / BA 636Ø NEXT M 6380 GOSUB 7500; REM SHOW NORM, AREA 6400 INPUT"PRINT OUT THE NORM. AREA (Y/N) "; DM\$ 6410 PRINT 6420 IF DM\$="N" THEN 6500 6430 GOSUB 9900: REM CHANGE REMARK 6440 OFEN4,4:CMD4 6450 GOSUB 18000: REM SHOW SAMPLE, EXPT. DATE, & REMARK 6460 GOSUB 7500:GOSUB 7700 6480 FRINT CHR\$(12):REM TOP OF THE NEXT FAGE 6490 PRINT#4:CLOSE4 6500 INPUT"SAVE NORM. AREA ON DISK(Y/N)"; DM\$ 652Ø PRINT 6540 IF DM\$="N" THEN 7000 6560 PRINT"** MAKE SURE RIGHT DISK IN DRIVE !!" 6580 FRINT WANT TO CHECK DISK(Y/N) "; DM\$ 6600 INPUT" 662Ø PRINT 664Ø IF DM\$="N" THEN 6700 666Ø STOP 6680 GOTO 6500 6700 REM SAVE ELECTRODE SURFACE AREA TO A SEQ. FILE 6720 REM N-A-* (N-H-*) = NORM. ELECTRODE SURFACE AREA 6740 PRINT 6760 PRINT"LATEST DATA FILE ID : "; ID\$ 6800 INPUT "NORM. AREA ID (N-A/H-*) ";NI\$ 6820 INPUT"IS FILE NAME OK(Y/N) "; DM\$ 6840 PRINT 6860 IF DM\$="N" THEN 6800 6700 OPEN 2,8,2,NI\$+",S,W" 6720 FOR M=0 TO 16

6940 PRINT#2,NA(M) 696Ø NEXT M 6980 CLOSE2 7000 INPUT"CHECK SAME SET AGAIN(Y/N)"; DM\$ 7020 PRINT 7040 IF DM\$="Y" THEN 3000 7100 INPUT"ADD ONE MORE SET FOR STAT. (Y/N) "; DM\$ 712Ø FRINT 7140 IF DM\$="Y" THEN 400 7200 INPUT"REDO STATISTICS W/O LAST SET(Y/N)";DM\$ 7220 PRINT 7240 IF DM\$<>"Y" THEN 7300 7260 55=55-1 728Ø GOTO 56ØØ 7300 INPUT"START OVER AGAIN(Y/N)";DM\$ 7320 PRINT 7340 IF DM\$="Y" THEN 300 7360 FRINT WHAT DO YOU WANT ?":PRINT:STOP 738Ø GOTO 7ØØØ 7400 END 7500 REM SHOW APPLIED VOLTAGE & NORMALIZED AREA 7520 FRINT"ELECTRODE", "POTENTIAL NORM. AREA" 754Ø PRINT 7560 FOR M=1 TO 16 7580 FRINT" # ";M,RS(M,0);NA(M) 7600 NEXT M 762Ø FRINT 7640 PRINT 766Ø RETURN 7700 REM MESSAGE TO SHOW THE METHOD 7720 IF OP=1 THEN PRINT" ** RESULTS ARE BASED ON PEAK AREA !" 7740 IF OP=2 THEN PRINT" ** RESULTS ARE BASED ON PEAK HEIGHT !" 7760 PRINT 7780 PRINT 7800 RETURN 8000 REM CHECK EACH SET DATA FOR ALL ELECTRODE 8020 PRINT"** END OF CHECK IF NN < 0 !" 8040 INPUT" NEW DATA SET # TO CHECK(NN)"; NN 8080 FRINT 8100 IF NN<0 THEN 8360 8200 PRINT" M DATA (NN) DATA (NN+1)" 8220 FOR M=0 TO 16 8240 PRINT M;RS(M,NN);RS(M,NN+1) 8260 NEXT M 828Ø FRINT 8300 FRINT" DATA SET # : ";NN 8320 PRINT 834Ø GOTO 8Ø2Ø 8360 RETURN 8400 REM CHECK WHETHER PRINTER IS READY 8420 FRINT ** MAKE SURE FRINTER ONLINE 844Ø FRINT" THEN PUSH 'G' TO GO ON !" 8460 PRINT 8470 GET AS: IF AS<>"G" THEN 8470 848Ø RETURN 8500 REM SHOW A SINGLE SET DATA 8520 PRINT 8540 PRINT N; "TH SET DATA OF SAMPLE" 8540 PRINT" DATA COLLECTED AT";RS(0,N);"SEC" 8580 PRINT" INTEGRATION TIME IS";TP(N);"SEC" 8600 PRINT 8620 PRINT"ELECTRODE", "TOTL.SAMP.CURRENT" 8640 FRINT 8660 FOR M=1 TO 16 8680 PRINT" # ";M,RS(M,N) 8700 NEXT M

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872Ø PRINT 874Ø RETURN 8800 REM SUBROUTINE TO SEE ALL DATA ON TWO ELECTRODES 8820 FRINT"SET"; " ELECTRODE"; X1; " ELECTRODE"; X2 8840 FOR N=BN TO ED 8860 PRINT N:RS(X1,N);RS(X2,N) 888Ø NEXT N 8900 PRINT 8720 GOSUB 13500: REM RECORD RANGE 894Ø RETURN 9000 REM SHOW ALL PARAMETERS ON SCREEN 9020 PRINT"EXPERIMENTAL CONDITIONS ---9040 PRINT 9060 PRINT" FLOW RATE -- ";FR\$ 9080 PRINT 9100 PRINT" SQUARE WAVE FOR PRECONDITION ---9120 PRINT" POTENTIAL OF UPPER LIMIT :";UL; "VOLT" 9140 PRINT" POTENTIAL OF LOWER LIMIT :";LL; "VOLT" 9160 PRINT" FREQUENCY :";HZ; "HERTZ" 9180 PRINT" DURATION :";DU; "SEC" 9200 FRINT 9220 PRINT IV: "VOLTS OUTPUT FROM DAC7548 #1" 924Ø PRINT FV; "VOLTS OUTPUT FROM DAC7548 #2" 9260 FRINT 9280 PRINT" WAIT"; TD; "SEC TO SUBSIDE CHARGING CURRENT DOWN !" 9300 PRINT 9320 PRINT" U2ALO IS";U2 9340 PRINT" TOTAL COVERED TIME IS"; TT; "SEC" 9360 FRINT" U1 TIMER SETUP AS"; U1; "* (1/60) SEC" 9380 PRINT" TIME OF RESOLUTION IS ABOUT"; TR; "SEC" 9400 PRINT NT; "SETS OF DATA COLLECTED" 942Ø PRINT 7440 PRINT" REFERENCE ELECTRODES : ZINC IN 1 M SODIUM ACETATE SOLUTION." 9460 PRINT" WORKING ELECTRODES : CARBON FIBER ARRAY ELECTRODE" 9480 PRINT 9500 FRINT 9520 FRINT" ** REMARK : ";RM\$ 954Ø FRINT 9600 RETURN 9700 REM SHOW SAMPLE ID 9720 PRINT"SAMPLE -- ";LB\$ 9740 PRINT" AREA DATA FILE ID : "; ID\$ 9760 PRINT" DATE OF EXPERIMENT : ";DA\$ 9780 PRINT 9800 PRINT" ** REMARK : ";RM\$ 982Ø PRINT 984Ø PRINT 986Ø RETURN 9900 REM CHANGE REMARK 9910 PRINT"PREVIOUS REMARK : ";RM\$ 9920 INPUT"IS IT OK(Y/N)";DM\$ 993Ø PRINT 994Ø IF DM\$="Y" THEN 999Ø 9960 INPUT"NEW REMARK"; RM\$ 997Ø PRINT 998Ø GOTO 991Ø 999Ø RETURN 10000 REM STATISTICS FOR MEAN, STANDARD DEVIATION(SD), AND RELATIVE SD 10100 REM MEAN 10200 FOR M=1 TO 16 1Ø24Ø D1=Ø 10260 FOR X=1 TO I 10280 D1=SC(M,X)+D1 10300 NEXT X 10340 MN(M) = D1/I10500 REM SD & RELATIVE SD (RD)

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10620 D3=0
10640 FOR X=1 TO I
10660 D2=SC(M, X)-MN(M)
1Ø68Ø D3=D2*D2+D3
10700 NEXT X
10720 SD(M)=SQR(D3/(I-1)):REM SUBSTITUTE I-1 WITH I
10740 RD (M) = SD (M) /MN (M) *100
10760 NEXT M
10800 GOSUB 11500: REM SHOW RESULTS
11000 INPUT"PRINT OUT STATISTIC RESULTS (Y/N) ": DM$
11020 PRINT
11040 IF DM$="N" THEN 11300
11060 GOSUB 9900:REM CHANGE REMARK
11100 OPEN4,4:CMD4
11200 GOSUB 18000:REM SAMPLE, EXPT. DATE, & REMARK
11220 GOSUB 11500:REM SHOW RESULTS
11240 GOSUB 7700:REM SHOW METHOD
11260 PRINT CHR$(12)
11280 FRINT#4:CLOSE4
11300 RETURN
11500 REM SHOW STATISTIC RESULT
11600 PRINT"ELECTRODE"," CURRENTS FOR AREA NORMALIZATION"
11620 PRINT" "," MEAN STD DEV REL STD DEV
                                                    REL STD DEV(%)"
1164Ø PRINT
11700 FOR M=1 TO 16
11720 FRINT" #";M,MN(M);" ";SD(M);" ";RD(M)
1174Ø NEXT M
1176Ø PRINT
11780 PRINT
11800 FRINT" ** THE STATISTICS IS BASED ON"; I; "DATA SETS"
1184Ø FRINT
11860 RETURN
12000 REM SUBROUTINE SHOWN ALL RANGES
12020 PRINT"ELECTRODE", "RF RL SF SL"
12040 PRINT
12100 FOR M=1 TO 16
12120 PRINT" #";M,RF;RL;SF(M);SL(M)
1214Ø NEXT M
12160 FRINT
12180 PRINT"** WRITE DOWN WHICH ONES NOT RIGHT !"
12200 PRINT
12300 RETURN
13000 REM SUBROUTINE TO LOAD CAPACITANCE FOR EACH INTEGRATOR
13020 REM BASED ON THE RESULTS OBTAINED ON 10/31
13100 CA(1)=9.026E-10:CA(2)=8.618E-10
13120 CA(3)=8.941E-10:CA(4)=8.398E-10
13140 CA(5)=8.369E-10:CA(6)=8.985E-10
13160 CA(7)=8.932E-10:CA(8)=8.661E-10
13180 CA(9)=9.095E-10:CA(10)=8.982E-10
13200 CA(11)=8.954E-10:CA(12)=8.951E-10
13220 CA(13)=9.105E-10:CA(14)=8.888E-10
13240 CA(15)=8.190E-10:CA(16)=8.500E-10
13300 RETURN
13500 REM RECORD RANGE OF SAMPLE CURRENT MAXIMUN
13520 PRINT"** RANGES OF PEAK HEIGHT MAXIMUM"
13560 PRINT
13600 PRINT"
               FOR ELECTRODE #";X1
1364Ø INPUT"
               RANGE FOR SAMP. CURRENT"; SF(X1), SL(X1)
13660 PRINT
13800 FRINT"
               FOR ELECTRODE #":X2
1384Ø INPUT"
               RANGE FOR SAMP. CURRENT"; SF(X2), SL(X2)
13860 PRINT
1388Ø INPUT"
               ARE ALL OK (Y/N))"; DM$
13900 FRINT
13920 IF DM$<>"Y" THEN 13600
```

10600 D2=0

13940 RETURN 14000 REM GENERATE A BEEPER (REF. P.185) 14020 PRINT" PUSH 'S' TO STOP BEEPING &" 14Ø4Ø PRINT" WATCH SCREEN FOR HINT !" 14060 PRINT 14100 GET A\$: IF A\$="S" THEN 14400 14200 S=54272 14220 FOR L=S TO S+24:POKE L,0:NEXT 14240 POKE S+5,9:POKE S+6,0 14260 POKE S+24,15 14280 HF=37:LF=162:DR=500 14300 POKE S+1, HF: POKE S, LF 14320 POKE S+4,33 14340 FOR T=1 TO DR:NEXT 14360 POKE S+4,32:FOR T=1 TO 50:NEXT 1438Ø GOTO 141ØØ 14400 RETURN 15000 REM GET RESI.CURRENT & TOTL.SAMP.CURRENT BY PEAK AREA 15020 PRINT"-- HANG ON, CALC. PEAK AREA !" 15040 PRINT 15100 REM CALC. AVERAGE RESI.CURRENT 15110 FOR M=1 TO 16 1512Ø RC(M)=Ø 15130 FOR N=RF TO RL 1514Ø RC (M) =RC (M) +RS (M,N) 1515Ø NEXT N 1516Ø RC (M) =RC (M) / (RL-RF+1) 1517Ø NEXT M 15200 REM CALC. NET SAMPLE CURRENT PEAK AREA 15220 PRINT"THE FOLLOWING 'N' VALUE SHOULD" 15240 PRINT" NOT BE GREATER THAN";NT+1 1526Ø PRINT 15300 FOR M=1 TO 16 1531Ø SC(M,SS)=Ø 1532Ø N=RL+1 15330 SC(M,SS) = (RS(M,N) - RC(M)) *RS(Ø,N) +SC(M,SS) 1534Ø N=N+1 1535Ø IF N>NT THEN 154ØØ 15360 IF N<=20 THEN 15330 15370 IF RS(M,N) >=RC(M) THEN 15330 15380 FRINT" M =";M;"; N =";N 15400 NEXT M 1542Ø PRINT 15440 GOSUB 14000:REM BEEP 1546Ø RETURN 15500 REM GET RESI.CURRENT & TOTL.SAMP.CURRENT BY PEAK HEIGHT 15520 PRINT"-- HANG ON, CALC. PEAK HEIGHT !" 1554Ø PRINT 15600 REM CALC. AVERAGE RESI.CURRENT 15610 FOR M=1 TO 16 1562Ø RC(M)=Ø 1563Ø FOR N=RF TO RL 15640 RC(M) =RC(M) +RS(M,N) 1565Ø NEXT N 1566Ø RC(M)=RC(M)/(RL-RF+1) 1567Ø NEXT M 15700 REM CALC. NET SAMPLE CURRENT FEAK HEIGHT 15720 FOR M=1 TO 16 15740 SC(M,SS)=0 15760 FOR N=SF(M) TO SL(M) 15780 SC(M,SS)=SC(M,SS)+RS(M,N) 15800 NEXT N 1582Ø SC(M,SS)=SC(M,SS)/(SL(M)-SF(M)+1) 15840 SC(M,SS)=SC(M,SS)-RC(M) 1586Ø NEXT M 15700 GOSUB 14000:REM BEEF

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1592Ø RETURN
16000 REM PRINT OUT DATA FOR PLOTTING
16020 PRINT"** INPUT THE RANGE OF DATA TO PRINT"
16040 PRINT" END OF PRINT IF 1ST<0 !"
16060 PRINT
16100 INPUT"
                1ST SET, LAST SET"; BN,ED
1612Ø INPUT"
               ARE BOTH OK (Y/N) "; DM$
1614Ø PRINT
16160 IF DM$<>"Y" THEN 16100
1618Ø IF BN<Ø THEN 17000
16200 REM PRINT OUT THE TIME AT WHICH DATA WERE COLLECTED
16220 OPEN 4,4:CMD 4
16240 PRINT"AREA DATA FILE ID : ":ID$
16260 PRINT
1628Ø PRINT
16300 PRINT"SET", "TIME (DATA COLLECTED)"
16320 PRINT
16340 FOR N=BN TO ED
16360 PRINT N,RS(0,N)
163BØ NEXT N
16400 PRINT CHR$(12):REM GET TO THE TOP OF NEXT PAGE
16420 PRINT#4
1644Ø CLOSE4
16500 PRINT"** END OF PRINT IF FT<0 !!"
16540 PRINT" CHOOSE 4 ELECTRODES TO PRINT DATA !"
16560 INPUT" 1ST,2ND,3RD,4TH ELECTRODE";FT,SD,TD,FH
16620 INPUT ARE ALL ABOVE OK (Y/N) "; DM$
1664Ø FRINT
16660 IF DM$<>"Y" THEN 16500
16680 IF FT<0 THEN 17000
16700 OPEN 4,4:CMD 4
16720 PRINT"SET","
                                      ELECTRODE"
1674Ø FRINT" # ";"
                        #";FT," LL POT. TO DIGITAL
16620 DM%=INT (LL*4095/6.013+2042)
1664Ø RH%=INT (DM%/256)
1666Ø RL%=INT (DM%-RH%*256)
16700 RETURN
17000 REM PRINT OUT AVE. RESIDUAL & NET SAMPLE CURRENT
17100 PRINT"ELECTRODE","
                                                              SAMPLE RANGE"
                                       CURRENT*
1712Ø PRINT" ","
                               RESIDUE
                                              NET SAMPLE
                                                                    LAST"
                                                               1 ST
1714Ø PRINT
17160 FOR M=1 TO 16
17180 PRINT" # ";M," ";RC(M);SC(M,SS);" ";SF(M);"
                                                            ": SL (M)
17200 NEXT M
1722Ø PRINT
1724Ø PRINT
17260 PRINT" * PRESENT AS AVERAGED RESULTS !!"
1728Ø PRINT
17300 PRINT"
               FOR THE RANGE OF AVE. RESIDUAL CURRENTS"
17320 PRINT"
                1ST SET IS";RF;"& LAST SET IS";RL
174ØØ RETURN
18000 REM SHOW SAMPLE, EXPT. DATE, & REMARK
18100 PRINT"SAMPLE -- ";LB$
18120 PRINT
18140 PRINT" DATE OF EXPERIMENT : "; DA$
18160 PRINT
18180 PRINT" ** REMARK : ";RM$
18200 PRINT
1822Ø PRINT
18300 RETURN
20000 REM PRINT OUT ORIGINAL DATA
20100 OPEN4,4:CMD4
20200 PRINT" SAMPLE LABEL : ";LB$
20220 FRINT
20240 PRINT" FLOW RATE : ";FR$
20260 PRINT
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20280 FRINT" INI. VOLT :";VI;" VOLT"
20300 PRINT" FINAL VOLT :";VF;" VOLT"
20320 PRINT
20340 PRINT"
              U1 VALUE :";U1
20360 PRINT" LEAST U2 BYTE =";U2
20380 PRINT"
              U2 CLOCK PERIOD (CP) =";CP;" SEC"
20400 PRINT"
              SET-UP TOTAL COVERED TIME =";CT; "SEC"
20420 PRINT
20440 PRINT" AFTER RECONDITIONING ELECTRODES,"
               "; TD; "SEC LATER, STARTED TO COLLECT DATA!"
20450 PRINT"
20460 PRINT
20480 PRINT#4:CLOSE4
20500 FRINT"** Z MUST <=";N;" & = MULTIPLE OF 5"
20520 PRINT" IF Z<0 THEN END OF PRINTING "
                NEW STARTING SET #"; Z
20540 INPUT"
20560 PRINT
20580 IF Z<0 THEN 21000
20600 OPEN4,4:CMD4
20700 PRINT
20720 PRINT" ** Z =";Z
20730 PRINT
20740 PRINT" M DD%Z DD%Z+1 DD%Z+2 DD%Z+3 DD%Z+4"
20760 FOR M=0 TO 16
20780 PRINT M; DD% (M,Z); DD% (M,Z+1); DD% (M,Z+2); DD% (M,Z+3); DD% (M,Z+4)
20800 NEXT M
20820 PRINT
20840 PRINT#4:CLOSE4
20900 GOTO 20500
21000 RETURN
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APPENDIX F

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SOFTWARE "RESULT" The program "RESULT", shown at the end of this section, is used to get the normalized amperometric data for the construction of pseudo-voltammograms. The flow chart of this software is shown in Figure 29.

This program first loads the saved coulometric data for the construction of a pseudo-voltammogram from a disk to the RAM of the C-64 and the normalization factors of all 16 CFAEs. Then the applied potentials at each CFAE are obtained as described in the previous section. The coulometric data are normalized with normalization factors, then converted into the amperometric data by dividing by the integration time. The net currents at each CFAE, which are used to construct a pseudo-voltammogram, are obtained as described in the previous section too. Finally, all these normalized net currents for plotting a pseudo-voltammogram are printed out.





10 REM *********** 11 REM * 2/13/1989 20 REM * "RESULT" 21 REM * 30 REM * USE CALC. CAPACITANCE & ¥ 31 REM * NORM. ELECTRODE AREA TO 32 REM * CALC. PEAK AREA/HEIGHT OF NET SAMPLE CURRENT. 40 REM * 41 REM * STATISTICS IS INCLUDED. × 60 REM DAS= DATE OF RUNNING EXPERIMENT 62 REM LB\$= SAMPLE LABEL (SPECIES & CONC.) 64 REM FR\$= FLOW RATE (FUMP SCALE) 66 REM IV = INITIAL VOLTAGE ON DAC 7548 #1 68 REM FV = FINAL VOLTAGE ON DAC 7548 #2 70 REM TR = TIME RESOLUTION (TR SEC) 71 REM U1 = U1BLO 72 REM U2 = U2ALO 73 REM TT = ACTUAL TOTAL COVERED TIME 74 REM NT = TOTAL DATA SET # SAVED FROM EXPT. 75 REM RM\$= EXPT. REMARK 76 REM NR = RESIDUAL CURRENT SET'S # 78 REM NS = SAMPLE CURRENT SET'S # 80 REM RS (M,N) = RESULTS FROM CALCULATION 81 REM RS(Ø,N)=INTEGRATION TIME 82 REM RS (M, Ø) = REAL APPLIED VOLTAGE ON EVERY ELECTRODE 83 REM RB(M)=AVE. RESI.CURRENT AT BEGIN 84 REM RE(M)=AVE. RESI.CURRENT AT END 86 REM RC(M,SS)≈AVE. RESIDUAL CURRENT 88 REM SC (M, SS) = PEAK AREA FOR SAMP. CURRENT 89 REM SA (M, SS) = AVE. PEAK AREA OF SAMP. CURRENT 90 REM TP (N) =TIME AT WHICH DATA COLLECTED 92 REM NA(M,SS) =NORM.AREA OF ELECTRODE M FROM "SS" DATA SET 94 REM EA (M) = AVE. SURFACE AREA OF ELECTRODE M 96 REM CA(M) = CAPACITANCE OF ELECTRODE M 100 DIM RS(16,200), RC(16,6), SC(16,6), NA(16), TP(200) 110 DIM SD(16), RD(16), R(16), CA(16), SA(16) 120 DIM SF(16), SL(16) 140 ID\$="'NONE'":REM DATA FILE ID 160 PRINT 200 PRINT"** MAKE SURE (1)LINE 10720 CORRECT," 220 PRINT" (2) RIGHT DISK IN DRIVE, &" 24Ø PRINT" (3) BEEP IS HEARD !" 26Ø PRINT 280 GOSUB 14000:REM BEEP 300 SS=0:REM SET # FOR REPEATABILITY 320 GOSUB 12000:REM LOAD NORM.AREA 400 SS=SS+1 500 INPUT"KNOW SAMPLE DATA FILE ID(Y/N)";DM\$ 52Ø PRINT 540 IF DM\$="Y" THEN 700 56Ø PRINT"****************** *" 580 PRINT"* TYPE '@\$' & LOOK FILE *" 600 PRINT"* NAME STARTS WITH 'S-620 PRINT"* THEN TYPE 'CONT' *" 660 PRINT 680 STOP

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720 PRINT"* FOR SAMPLE DATA FILE ID, *"
740 PRINT"* IT STARTS WITH 'S-'
                                     J£ 11
78Ø PRINT
800 PRINT"PREVIOUS SAMPLE DATA FILE ID : "; ID$
820 INPUT"NEW SAMPLE DATA FILE ID(S-*)"; ID$
840 INPUT"IS FILE NAME OK(Y/N)"; DM$
86Ø PRINT
880 IF DM$<>"Y" THEN 500
900 PRINT"-- IF DRIVE LIGHT BLINKS,"
910 PRINT"
             STOP PROGRAM IMMEDIATELY. T
TYPE 'CLOSE 2', FOLLOWED BY"
                                        THEN"
920 PRINT"
930 PRINT"
             TYPING 'GOTO 500' !!"
940 PRINT
             WAIT (ABOUT 5 MINUTES)"
950 PRINT"
960 PRINT"
               TO LOAD & TREAT DATA !!"
780 PRINT
1000 REM GET SAMPLE DATA TO NORMALIZE
1020 OPEN 2,8,2,1D$+",S,R"
1040 INFUT#2,DA$
1060 INPUT#2.LB$
1080 INPUT#2,FR$
1100 INPUT#2, UL: REM UPPER LIMIT OF S.W.
1120 INPUT#2, LL: REM LOWER LIMIT OF S.W.
114Ø INPUT#2,HZ:REM FREQ. OF S.W.
1160 INPUT#2, DU: REM DUATION OF S.W.
1180 INPUT#2, TD: REM TIME OF DELAY
1200 INPUT#2.IV
122Ø INPUT#2,FV
1230 INPUT#2,U1
124Ø INPUT#2, TR
1260 INPUT#2,U2
128Ø INFUT#2,TT
1300 INPUT#2,NT
1310 INPUT#2,RM$
1320 IF IV<FV THEN 1500
1340 CLOSE 2
1400 PRINT"** FROGRAM TO NORMALIZE SAMPLE DATA"
1420 PRINT"
             LOAD WRONG DATA FILE !!"
144Ø PRINT
146Ø STOP
1500 REM TREAT DATA BEFORE PUT INTO A MATRIX
1520 CY=1.023E6:REM CY=FREQUENCY OF TIMER
1540 GOSUB 13000:REM LOAD CAPACITANCE FOR EACH INTEGRATOR
1560 REM ORIGINAL SET UP OF U2 TIMER
158Ø INPUT#2,TM%
1600 RS(0,0)=TM%*U2/CY
1620 REM CALC. REAL APPLIED POTENTIAL
1640 FOR M=1 TO 16
1660 INPUT#2,TM%
168Ø RS(M,Ø)=(TM%-2Ø43)/2Ø8
1700 NEXT M
1720 TP(0)=0
1740 FOR N=1 TO NT
1760 INPUT#2, TM%
1780 RS(0,N)=TM%+U2/CY:REM INTEGRATION TIME
1800 TP (N) =RS (0, N) +TP (N-1) : REM TIME SCALE
1820 FOR M=1 TO 16
184Ø INFUT#2,TM%
1860 RS(M,N)=TM%/208*CA(M)/RS(0,N)/NA(M)
1880 NEXT M
1900 NEXT N
1920 CLOSE 2
1940 GOSUB 14000: REM BEEF
2000 GOSUB 9900: REM CHANGE REMARK
```

2010 REM SHOW ALL PARAMETERS ON SCREEN

```
2030 PRINT"
              THEN HOLD 'CTRL' KEY TO VIEW"
2040 PRINT
2050 GET A$: IF A$<>"G" THEN 2050
2060 GOSUB 9700:REM SHOW SAMPLE ID
2070 GOSUB 9000: REM SHOW EXPT. CONDITIONS
2100 INPUT"ARE ALL ABOVE CORRECT (Y/N) "; DM$
212Ø PRINT
2140 IF DM$="Y" THEN 2200
2160 PRINT"** MAKE CORRECTION, THEN TYPE 'CONT'":STOP
2180 PRINT: 60T0 2000
2200 REM CHECK APPLIED VOLTAGE ON ELECTRODE
222Ø PRINT
2240 PRINT"ELECTRODE
                         APPLIED VOLTAGE"
2260 PRINT
2280 FOR M=1 TO 16
2300 PRINT "
              # ";M;"
                               ";R5(M,Ø)
2320 NEXT M
2340 PRINT
2400 RF=0:RL=0
2420 FOR M=1 TO 16
244Ø SF(M)=0:SL(M)=0
246Ø NEXT M
2500 REM CHOOSE RANGES FOR RESI:CURRENT & SAMP.CURRENT SETS
2520 BN=3:ED=23:REM START FROM SET 3 TO SET 23
2600 PRINT"** ENTER ELECTRODE #'S TO CHECK !!"
2620 PRINT" (END OF CHECK IF 1ST INPUT <=0)"
262Ø FRINT"
264Ø PRINT
266Ø INPUT"
              1ST & 2ND ELECTRODE (<=16) "; X1, X2
268Ø INPUT"
              ARE BOTH OK (Y/N) "; DM$
2700 FRINT
2720 IF DM$<>"Y" THEN 2600
2740 IF X1<=0 THEN 3500
2800 FRINT"** PUSH 'G' TO GO ON & THEN"
2820 PRINT" HOLD 'CTRL' KEY TO CHECK DATA "
2840 PRINT"
             & FIND RANGES FOR RESI.& SAMP.CURRENT"
286Ø FRINT
2900 GET A$: IF A$<>"G" THEN 2900.
2920 GOSUB 8800
3000 FRINT" 1ST ELECTRODE CHECKED :";X1
3020 PRINT" 2ND ELECTRODE CHECKED :";X2
3040 PRINT" INITIAL SET TO CHECK :"; BN
3060 PRINT" END SET TO CHECK :";ED
3080 PRINT
3100 INPUT" IS CHECK RANGE OK(Y/N)"; DM$
3120 PRINT
3140 IF DM$="Y" THEN 2600
3200 PRINT"** INPUT INITIAL & END SET TO CHECK"
3210 PRINT" END SET MUST BE SMALLER THAN"; NT
322Ø PRINT
324Ø INPUT"
               INITIAL SET & END SET ": BN, ED
3260 INPUT"
              ARE BOTH OK(Y/N) "; DM$
3280 PRINT
3300 IF DM$<>"Y" THEN 3200
3400 IF ED<=NT THEN 2600
3420 PRINT" ** END SET TO CHECK IS OVER RANGE **"
344Ø PRINT
3460 GOTD 3200
3500 INPUT INPUT RESIDUAL CURRENT RANGE"; RF, RL
3510 INPUT ARE BOTH OK (Y/N) "; DM$
3520 PRINT
3540 IF DM$<>"Y" THEN 2600
3550 GOSUB 13500: REM SHOW RANGE
3560 INPUT "ARE ALL OK (Y/N) "; DM$
3570 PRINT
3580 IF DM$<>"Y" THEN 2600
```

2020 PRINT"** PUSH 'G' TO GO ON,"

3600 INPUT"PRINT OUT DATA TO PLOT(Y/N)";DM\$ 3620 PRINT 3640 IF DM\$="N" THEN 3800 3660 INPUT "HAVE PARAMETERS PRINTED OUT (Y/N) ": DM\$ 3680 PRINT 3700 IF DM\$="Y" THEN 3790 3720 OPEN4,4:CMD4 374Ø GOSUB 9000 3760 PRINT#4 378Ø CLOSE4 3790 GOSUB 16000 3800 INPUT"GOT ALL RANGES (Y/N) "; DM\$ 3820 PRINT 3840 IF DM\$="Y" THEN 4000 3860 INPUT"CHECK THE SAME SET AGAIN(Y/N)"; DM\$ 388Ø PRINT 3900 IF DM\$<>"N" THEN 3000 3920 INPUT"LOAD ANOTHER DATA SET (Y/N) "; DM\$ 3940 PRINT 3960 IF DM\$="Y" THEN 500 3980 STOP 4000 REM GET RESI.CURRENT & NET SAMPLE CURRENT 4020 PRINT"WHICH WAY TO GET DATA FOR VOLTAMMOGRAM" 4040 INPUT" (1) BY PEAK AREA: (2) BY PEAK HEIGHT"; OP 4050 INPUT" IS IT OK (Y/N) "; DM\$ 4060 FRINT 4080 IF DM\$<>"Y" THEN 4020 4100 IF OP=1 THEN GOSUB 15000:REM BY AREA 4120 IF OP=1 THEN 4200 4140 IF OP=2 THEN GOSUB 15500:REM BY HEIGHT 4200 REM SHOW AVERAGED CURRENTS ON SCREEN 4220 PRINT"ELECTRODE RESI.CURRENT* NET SAMP.CURRENT" 4240 FRINT 4260 FOR M=1 TO 16 4280 PRINT " # ";M;RC(M,SS);SC(M,SS) 4300 NEXT M **4320 FRINT** 5000 INPUT"PRINT OUT RESULTS (Y/N) "; DM\$ 5020 PRINT 5040 IF DM\$="N" THEN 6000 5060 GOSUB 7700: REM SURE FRINTER ONLINE 5100 INPUT"HAVE PARAMETERS PRINTED OUT (Y/N) "; DM\$ 5120 PRINT 5140 IF DM\$="Y" THEN 5400 5200 OPEN4,4:CMD4 5240 GOSUB 9000 5260 PRINT CHR\$(12):REM GET TO THE TOP OF THE NEXT PAGE 528Ø PRINT#4 5300 CLOSE4 5400 GOSUB 9900: REM CHANGE REMARK 5420 OPEN4,4:CMD4 5440 GOSUB 9700: REM SAMPLE ID 5500 REM PRINT OUT AVE. RESIDUAL & TOTAL SAMPLE CURRENT 5520 PRINT"ELECTRODE", "VOLT. APPLY RESI.CURRENT* NET SAMP.CURRENT*" 554Ø FRINT 5560 FOR M=1 TO 16 5580 FRINT" # ";M,RS(M,0);" ";RC(M,SS);" ";SC(M,SS) 5600 NEXT M 5620 FRINT 564Ø PRINT 5660 PRINT" * ABOVE ARE AVERAGED RESULTS !!" 5670 GOSUB 7900:REM STATE METHOD 5480 PRINT 5690 FRINT 5700 PRINT"THE DATA RANGES ARE AS FOLLOWING :" 5710 FRINT

5720 PRINT"ELECTRODE"," RESIDUAL CURRENT NET SAMPLE CURRENT" 573Ø PRINT" # "," 1ST SET LAST SET 1ST SET LAST SET" 5740 FRINT 5760 FOR M=1 TO 16 5780 PRINT" ";M," ":RF:" "; RL, SF (M); " ": SL (M) " 5800 NEXT M 5820 PRINT 5840 PRINT CHR\$(12):REM GET TO THE TOP OF THE NEXT PAGE 5860 PRINT#4:CLOSE 4 5900 INPUT"IS PRINTOUT OK(Y/N)"; DM\$ 592Ø PRINT 5940 IF DM\$<>"Y" THEN 5000 6000 INPUT"ARE RESULTS OK (Y/N) "; DM\$ 6020 PRINT 6040 IF DM\$="Y" THEN 6500 6100 INPUT"CHECK THE SAME SET AGAIN (Y/N) "; DM\$ 612Ø PRINT 6140 IF DM\$<>"N" THEN 2600 6200 INPUT"LOAD NEW DATA SET(Y/N) "; DM\$ 6220 PRINT 6240 IF DM\$="Y" THEN 500 6260 GOTO 7000 6500 INPUT"RESULT FOR STATISTICS (Y/N) "; DM\$ 6520 PRINT 6540 IF DM\$="Y" THEN 6600 6560 GOTO 6100 6600 PRINT"PRESENT DATA FILE ID : "; ID\$ 6620 PRINT"PRESENT SET # FOR REPEATABILITY : ";SS 664Ø PRINT 6660 INPUT DONE REPEATABILITY CALC. (Y/N) "; DMS 6680 PRINT 6700 IF DM\$="Y" THEN 6800 6720 INPUT"ARE YOU SURE (Y/N) "; DM\$ 674Ø PRINT 6760 IF DM\$="Y" THEN 400 677Ø STOP 678Ø GOTO 666Ø 6800 INPUT"ARE YOU SURE (Y/N) "; DM\$ 6820 PRINT 684Ø IF DM\$="N" THEN 6770 6900 PRINT" ** NOW APPLY STATISTICS TO THE RESULTS" 6920 PRINT 6940 I=SS:REM TOTAL SET FOR REPEATABILITY IS 'SS' 6960 GOSUB 10000 7000 INPUT"CHECK SAME SET AGAIN(Y/N)"; DM\$ 7020 FRINT 7040 IF DM\$="Y" THEN 2600 7100 INPUT ADD ANOTHER DATA SET(Y/N) ; DM\$ 712Ø PRINT 7140 IF DM\$="Y" THEN 400 7200 INPUT"GET RID OFF PREVIOUS ONE (Y/N) "; DM\$ 722Ø PRINT 7240 IF DM\$="N" THEN 7300 726Ø SS=SS-1 7280 GOTO 6660 7300 PRINT"WANT TO START OVER/END! RIGHT?" 732Ø STOP 734Ø PRINT 7360 GOTO 7000 7500 END 7700 REM CHECK PRINTER ONLINE 7720 FRINT"** MAKE SURE FRINTER IS ONLINE." 7740 FRINT" THEN FUSH 'G' TO GO ON !" 774Ø PRINT" 776Ø FRINT 7780 GET A\$: IF A\$<>"G" THEN 7780 7800 RETURN

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7900 REM STATE METHOD
7920 IF DP=1 THEN FRINT" ** RESULTS ARE BASED ON PEAK AREA !"
7940 IF DP=2 THEN PRINT" ** RESULTS ARE BASED ON PEAK HEIGHT !"
7960 PRINT
798Ø RETURN
8000 REM SUBROUTINE TO CHECK WHOLE SET DATA
8020 PRINT"** END OF CHECK IF NN < 0 !"
8040 INPUT" NEW DATA SET # TO CHECK(NN)";NN
8080 PRINT
8100 IF NN<Ø THEN 8360
8200 PRINT" M
                   DATA (NN)
                                      DATA (NN+1) "
8220 FOR M=0 TO 16
8240 PRINT M; RS(M, NN); RS(M, NN+1)
8260 NEXT M
828Ø PRINT
8300 PRINT" DATA SET # : ";NN
8320 PRINT
834Ø GOTO 8Ø2Ø
836Ø RETURN
8500 REM SHOW A SINGLE SET DATA
8520 PRINT
8540 PRINT N; "TH SET DATA OF SAMPLE"
8560 PRINT" DATA COLLECTED AT";RS(0,N);"SEC"
8580 PRINT" INTEGRATION TIME IS"; TP (N); "SEC"
8600 FRINT
8620 PRINT"ELECTRODE", "TOTL.SAMP.CURRENT"
8640 PRINT
8660 FOR M=1 TO 16
8680 PRINT" # ";M,RS(M,N)
8700 NEXT M
8720 FRINT
874Ø RETURN
8800 REM SUBROUTINE TO SEE ALL DATA ON TWO ELECTRODES
8820 PRINT"SET"; " ELECTRODE"; X1; " ELECTRODE"; X2
8840 FOR N=BN TO ED
8860 PRINT N;RS(X1,N);RS(X2,N)
888Ø NEXT N
8900 PRINT
8920 GOSUB 14500: REM SET RANGE
894Ø RETURN
9000 REM SHOW ALL FARAMETERS ON SCREEN
9020 FRINT"EXPERIMENTAL CONDITIONS ---
9040 PRINT
9060 FRINT" FLOW RATE -- ";FR$
9080 FRINT
9100 PRINT" SQUARE WAVE FOR PRECONDITIONING ---
9120 PRINT" FOTENTIAL OF UPPER LIMIT :";UL; "VOLT"
914Ø PRINT"
             FOTENTIAL OF LOWER LIMIT :";LL; "VOLT"
9160 PRINT" FREQUENCY :";HZ; "HERTZ"
9180 PRINT"
             DURATION :";DU; "SEC"
9200 PRINT
9220 PRINT IV; "VOLTS OUTPUT FROM DAC7548 #1"
9240 PRINT FV; "VOLTS OUTPUT FROM DAC7548 #2"
926Ø FRINT
9280 PRINT" WAIT"; TD; "SEC TO SUBSIDE CHARGING CURRENT DOWN"
9300 PRINT
9320 PRINT" UZALO IS";U2
9340 FRINT" TOTAL COVERED TIME IS"; TT; "SEC"
9360 PRINT" U1 TIMER SETUP AS"; U1; "* (1/60) SEC"
9380 PRINT" TIME OF RESOLUTION IS ABOUT"; TR; "SEC"
9400 PRINT NT; "SETS OF DATA COLLECTED"
9420 PRINT
9440 PRINT" REFERENCE ELECTRODES : ZINC IN 1 M SODIUM ACETATE SOLUTION."
9460 PRINT" WORKING ELECTRODES : CARBON FIBER ARRAY ELECTRODE"
948Ø PRINT
9500 PRINT
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9520 PRINT" ** REMARK : ";RM$
954Ø PRINT
956Ø PRINT
958Ø RETURN
9600 REM SHOW SAMPLE, EXPT. DATE, & REMARK
9610 PRINT"SAMPLE -- ";LB$
9620 PRINT" NORM. AREA FILE ID : ";NF$
9630 PRINT" DATE OF EXPERIMENT : ";DA$
964Ø PRINT
9650 PRINT" ** REMARK : ";RM$
966Ø FRINT
967Ø FRINT
968Ø RETURN
9700 REM SHOW SAMPLE ID
9720 PRINT"SAMPLE --- ";LB$
9740 PRINT" SAMPLE DATA FILE ID : "; ID$
9760 FRINT" NORM. AREA FILE ID : ";NF$
9780 PRINT" DATE OF EXPERIMENT : ";DA$
9800 PRINT
9820 PRINT" ** REMARK : ";RM$
984Ø FRINT
986Ø PRINT
988Ø RETURN
9900 REM CHANGE REMARK
9910 FRINT"PREVIOUS REMARK : ";RM$
9920 INPUT"IS IT OK(Y/N)";DM$
9930 PRINT
9940 IF DM$="Y" THEN 9990
9950 INPUT"NEW REMARK"; RM$
9960 INPUT"IS IT OK(Y/N)";DM$
997Ø PRINT
9980 IF DM$<>"Y" THEN 9950
999Ø RETURN
10000 REM STATISTICS FOR MEAN, STANDARD DEVIATION (SD), AND RELATIVE SD
10100 REM MEAN
10200 FOR M=1 TO 16
10240 D1=0
10260 FOR X=1 TO I
10280 D1=SC(M,X)+D1
10300 NEXT X
10340 SA(M)=D1/I
10500 REM SD & RELATIVE SD (RD)
10600 D2=0
10620 D3=0
10640 FOR X=1 TO I
10660 D2=SC(M,X)-SA(M)
10680 D3=D2*D2+D3
10700 NEXT X
10720 SD(M)=SQR(D3/(I-1)):REM MAY SUBSTITUTE I-1 WITH I
10740 RD(M)=SD(M)/SA(M)*100
10760 NEXT M
10800 GOSUB 11500: REM SHOW RESULTS
11000 INPUT"PRINT OUT STATISTIC RESULTS (Y/N) "; DM$
11020 PRINT
11040 IF DM$="N" THEN 11460
11100 GOSUB 9900
11200 OFEN4,4:CMD4
11300 GOSUB 9600: REM SAMPLE, EXPT. DATE, & REMARK
11320 GOSUB 11500:REM SHOW RESULTS
11340 PRINT CHR$(12)
1136Ø FRINT#4
1138Ø CLOSE4
11400 INFUT"IS PRINTOUT OK(Y/N)";DM$
11420 PRINT
11440 IF DM$<>"Y" THEN 11000
1146Ø RETURN
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```
11500 REM SHOW STATISTIC RESULT.
11600 PRINT"ELECTRODE","
11620 PRINT" "," MEAN
                                     NET SAMPLE CURRENT"
                                             STD DEV
                                                          REL STD DEV(%)"
1164Ø PRINT
11700 FOR M=1 TO 16
11720 PRINT" #";M,SA(M);" ";SD(M);" ";RD(M)
11740 NEXT M
1176Ø FRINT
1178Ø PRINT
11800 PRINT" ** THE STATISTICS IS BASED ON"; I; "DATA SETS"
11820 IF OP=1 THEN PRINT" & ALSO BASED ON PEAK AREA !"
11840 IF OP=2 THEN PRINT" & ALSO BASED ON PEAK HEIGHT
                                 & ALSO BASED ON PEAK HEIGHT !"
11860 FRINT
1188Ø FRINT
11900 RETURN
12000 REM SUBROUTINE TO LOAD NORMALIZED AREA
12100 INPUT"KNOW NORM.AREA FILE ID (Y/N) "; DM$
1212Ø FRINT
12140 IF DM$="Y" THEN 12300
12200 PRINT"-- NORM.AREA FILE ID STARTS WITH 'N-*'"
12240 PRINT" TYPE '3$', THEN TYPE 'CONT'"
12260 PRINT: STOP
12300 PRINT
12320 PRINT"** FILE ID MUST START WITH 'N-A/H-*'"
1234Ø FRINT
12400 INPUT" NORM.AREA FILE ID";NF$
12420 INPUT" IS FILE NAME OK(Y/N)";DM$
12440 FRINT
12460 IF DM$<>"Y" THEN 12100
12500 PRINT"-- HANG ON, LOADING NORM.AREA !!"
12520 PRINT
12600 REM LOAD NORM. AREA
12700 OPEN 2,8,2,NF$+",S,R"
12720 FOR M=0 TO 16
12740 INPUT#2, NA (M)
12760 NEXT M
1278Ø CLOSE 2
12800 RETURN
13000 REM SUBROUTINE TO LOAD CAPACITANCE FOR EACH INTEGRATOR
13020 REM BASED ON THE RESULTS OBTAINED ON 10/31
13100 CA(1)=9.026E-10:CA(2)=8.618E-10
13120 CA(3)=8.941E-10:CA(4)=8.398E-10
13140 CA(5)=8.369E-10:CA(6)=8.985E-10
1316Ø CA(7)=8.932E-10:CA(8)=8.661E-10
13180 CA(9)=9.095E-10:CA(10)=8.982E-10
13200 CA(11)=8.954E-10:CA(12)=8.951E-10
13220 CA(13)=9.105E-10:CA(14)=8.888E-10
13240 CA(15)=8.190E-10:CA(16)=8.500E-10
13300 RETURN
13500 REM SUBROUTINE SHOWN RANGES
13520 PRINT"ELECTRODE", "RF RL SF SL"
1354Ø PRINT
13600 FOR M=1 TO 16
13620 PRINT" #";M;RF;RL;SF(M);SL(M)
13649 NEXT M
1366Ø PRINT
1368Ø RETURN
14000 REM GENERATE A BEEPER(REF. P.185)
14020 PRINT" PUSH 'S' TO STOP BEEPING &"
14040 PRINT"
                WATCH SCREEN FOR HINT !"
14060 PRINT
14100 GET A$: IF A$="S" THEN 14400
14200 5=54272
14220 FOR L=S TO S+24:POKE L,0:NEXT
14240 POKE S+5,9:POKE S+6,0
14260 POKE S+24,15
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1428Ø HF=37:LF=162:DR=5ØØ 14300 POKE S+1, HF: POKE S,LF 14320 POKE S+4,33 1434Ø FOR T=1 TO DR:NEXT 14360 POKE S+4,32:FOR T=1 TO 50:NEXT 1438Ø GOTO 141ØØ 14400 RETURN 14500 REM SET RANGE FOR SAMP.CURRENT PEAK HEIGHT MAXIMUM FOR ELECTRODE #";X1 14600 PRINT" 1464Ø INPUT" RANGE OF SAMPLE CURRENT"; SF(X1), SL(X1) 1466Ø PRINT 14700 PRINT" FOR ELECTRODE #";X2 1474Ø INPUT" RANGE OF SAMPLE CURRENT"; SF(X2), SL(X2) 14760 PRINT 14800 INPUT" ARE BOTH OK(Y/N)";DM\$ 14820 PRINT 1484Ø IF DM\$<>"Y" THEN 14600 14860 RETURN 15000 REM GET RESI.CURRENT & TOTL.SAMP.CURRENT BY PEAK AREA 15020 PRINT"-- HANG ON, CALC. PEAK AREA !" 15Ø4Ø PRINT 15100 REM CALC. AVERAGE RESI.CURRENT 1511Ø FOR M=1 TO 16 15120 RC(M,SS)=0 15130 FOR N=RF TO RL 15140 RC(M,SS)=RC(M,SS)+RS(M,N) 1515Ø NEXT N 15160 RC(M,SS)=RC(M,SS)/(RL-RF+1) 1517Ø NEXT M 15200 REM CALC. PEAK AREA OF NET SAMPLE CURRENT 15220 PRINT"-- THE FOLLOWING 'N' VALUE SHOULD BE" 1524Ø PRINT" NOT GREATER THAN"; NT+1 1526Ø PRINT 15300 FOR M=1 TO 16 1531Ø SC(M,SS)=Ø 1532Ø N=RL+1 15330 SC(M,SS)=(RS(M,N)-RC(M,SS))*RS(0,N)+SC(M,SS) 1534Ø N=N+1 15350 IF N>NT THEN 15410 1534Ø IF N<=2Ø THEN 1533Ø 15370 IF RS(M,20)<RC(M,SS) THEN 15400 15380 IF RS(M,N) >RC(M,SS) THEN 15330 1539Ø GOTO 1541Ø 15400 IF RS(M,N)<RC(M,SS) THEN 15330 1541Ø PRINT" M =";M;"; N =";N 15420 NEXT M 1544Ø PRINT 15460 GOSUB 14000:REM BEEP 1548Ø RETURN 15500 REM GET RESI.CURRENT & SAMP.CURRENT BY PEAK HEIGHT 15520 PRINT"-- HANG ON, CALC. PEAK HEIGHT !' 15540 PRINT 15600 REM CALC. AVERAGE RESI.CURRENT 15610 FOR M=1 TO 16 15620 RC(M,SS)=0 15630 FOR N=RF TO RL 15640 RC (M, SS) =RC (M, SS) +RS (M, N) 15650 NEXT N 15660 RC(M,SS)=RC(M,SS)/(RL-RF+1) 15670 NEXT M 15700 REM CALC. PEAK HEIGHT OF NET SAMPLE CURRENT 15720 FOR M=1 TO 16 15740 SC(M,SS)=0 15760 FOR N=SF(M) TO SL(M) 15780 SC (M,SS) = SC (M,SS) + RS (M,N) 15800 NEXT N

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15820 SC(M,SS)=SC(M,SS)/(SL(M)-SF(M)+1)
15840 SC(M,SS)=SC(M,SS)-RC(M,SS)
1586Ø NEXT M
15900 GOSUB 14000:REM BEEP
1592Ø RETURN
16000 REM PRINT OUT DATA FOR PLOTTING
16020 PRINT"** INPUT THE RANGE OF DATA TO PRINT"
16040 PRINT" END OF PRINT IF 1ST<0 !"
16060 PRINT
16100 INPUT"
               1ST SET, LAST SET"; BN, ED
1612Ø INPUT"
               ARE BOTH OK (Y/N) "; DM$
1614Ø PRINT
16160 IF DM$<>"Y" THEN 16100
1618Ø IF BN<Ø THEN 17000
16200 REM PRINT OUT THE TIME AT WHICH DATA WERE COLLECTED
16220 OPEN 4,4:CMD 4
16240 PRINT"AREA DATA FILE ID : "; ID$
1626Ø PRINT
1628Ø PRINT
16300 PRINT"SET", "TIME (DATA COLLECTED)"
1632Ø PRINT
16340 FOR N=BN TO ED
16360 PRINT N,RS(0,N)
1638Ø NEXT N
16400 PRINT CHR$(12):REM GET TO THE TOP OF NEXT PAGE
1642Ø PRINT#4
1644Ø CLOSE4
16500 PRINT"** END OF PRINT IF FT<0 !!"
16540 PRINT" CHOOSE 4 ELECTRODES TO PRINT DATA !"
1656Ø INPUT"
               1ST, 2ND, 3RD, 4TH ELECTRODE"; FT, SD, TD, FH
16620 INPUT ARE ALL ABOVE OK (Y/N) "; DM$
1664Ø PRINT
16660 IF DM$<>"Y" THEN 16500
1668Ø IF FT<Ø THEN 17000
16700 OPEN 4,4:CMD 4
16720 PRINT"SET","
                                    ELECTRODE"
1674Ø PRINT" # ";"
                        #";FT," #";SD," #";TD," #";FH
1675Ø PRINT
16760 FOR N=BN TO ED
16780 PRINT N;RS(FT,N);RS(SD,N);RS(TD,N);RS(FH,N)
16800 NEXT N
1682Ø PRINT CHR$(12)
1684Ø PRINT#4
1686Ø CLOSE4
16900 GOTO 16500
17000 RETURN
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READY.